



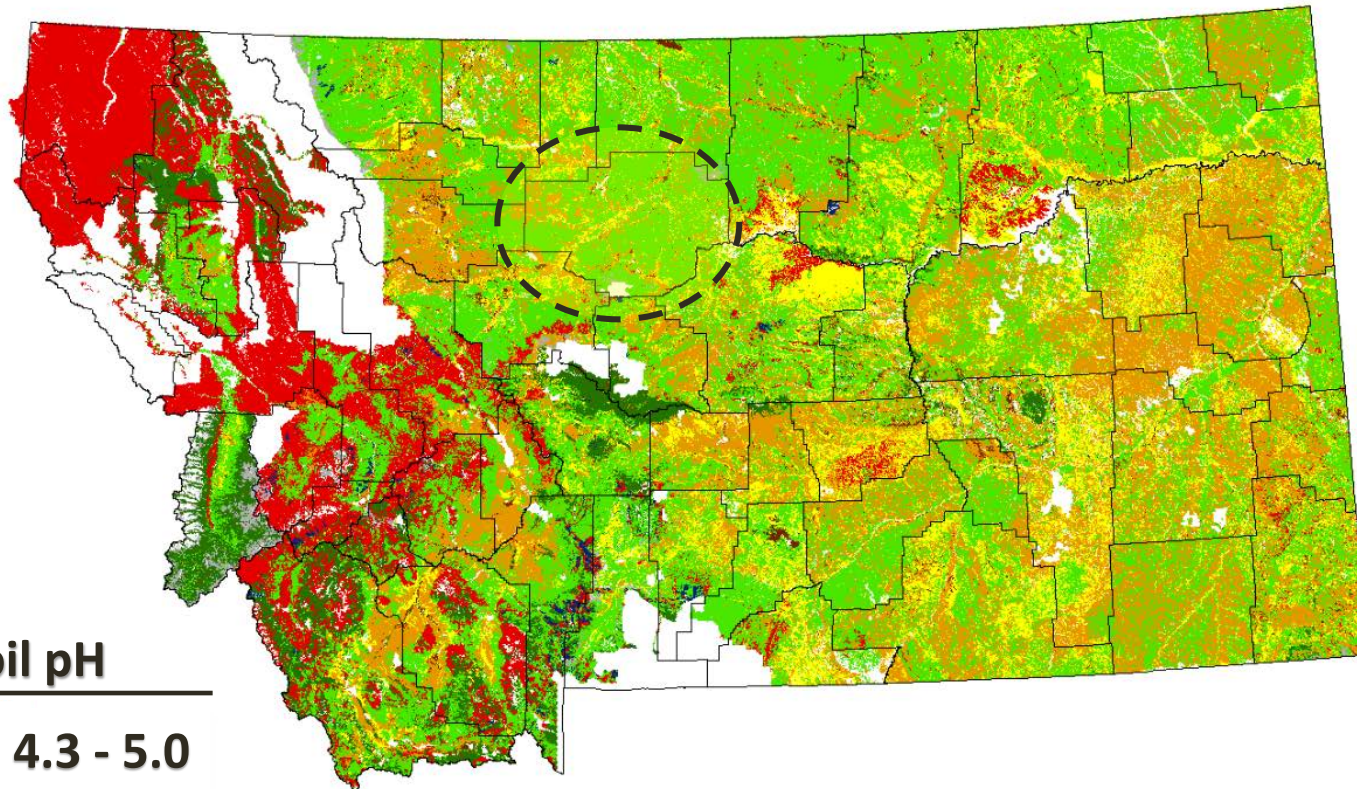
Understanding acidification and management of Montana soils

**Clain Jones and Richard Engel
Dept of Land Resources and Environmental Sciences**

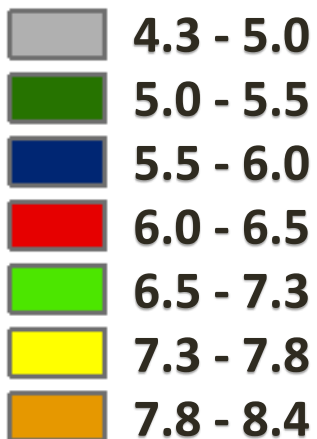
Questions for you

- How many of you have observed a downward trend in soil pH tests?
- How many think you have seen yield losses from acidic soils?
- How many of you have soil pH levels below 5.5?

Soil pH – NRCS published info



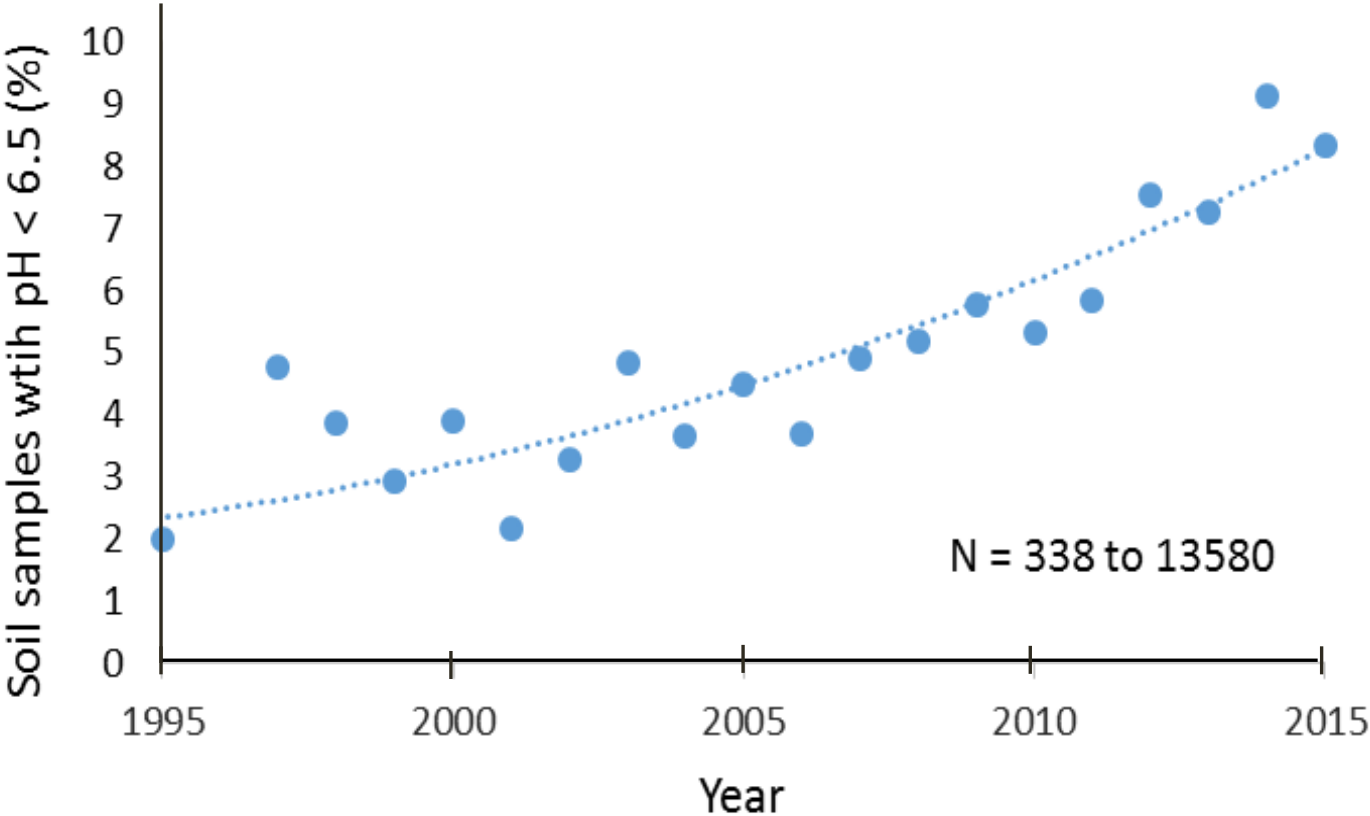
Soil pH



UNITES STATES DEPARTMENT OF AGRICULTURE
NATURAL RESOURCE CONSERVATION SERVICE

Trends in Montana to lower pH

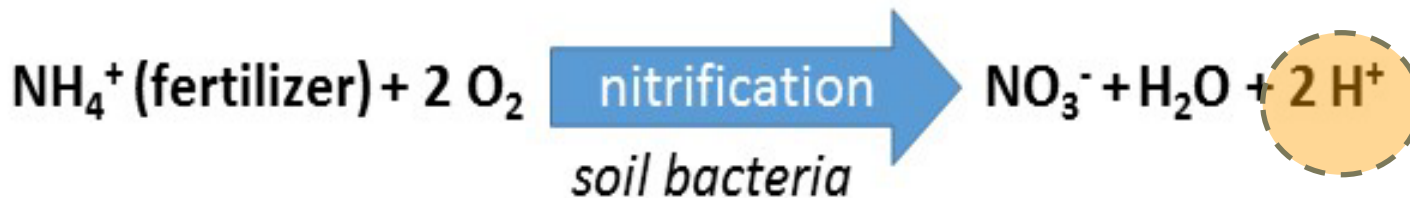
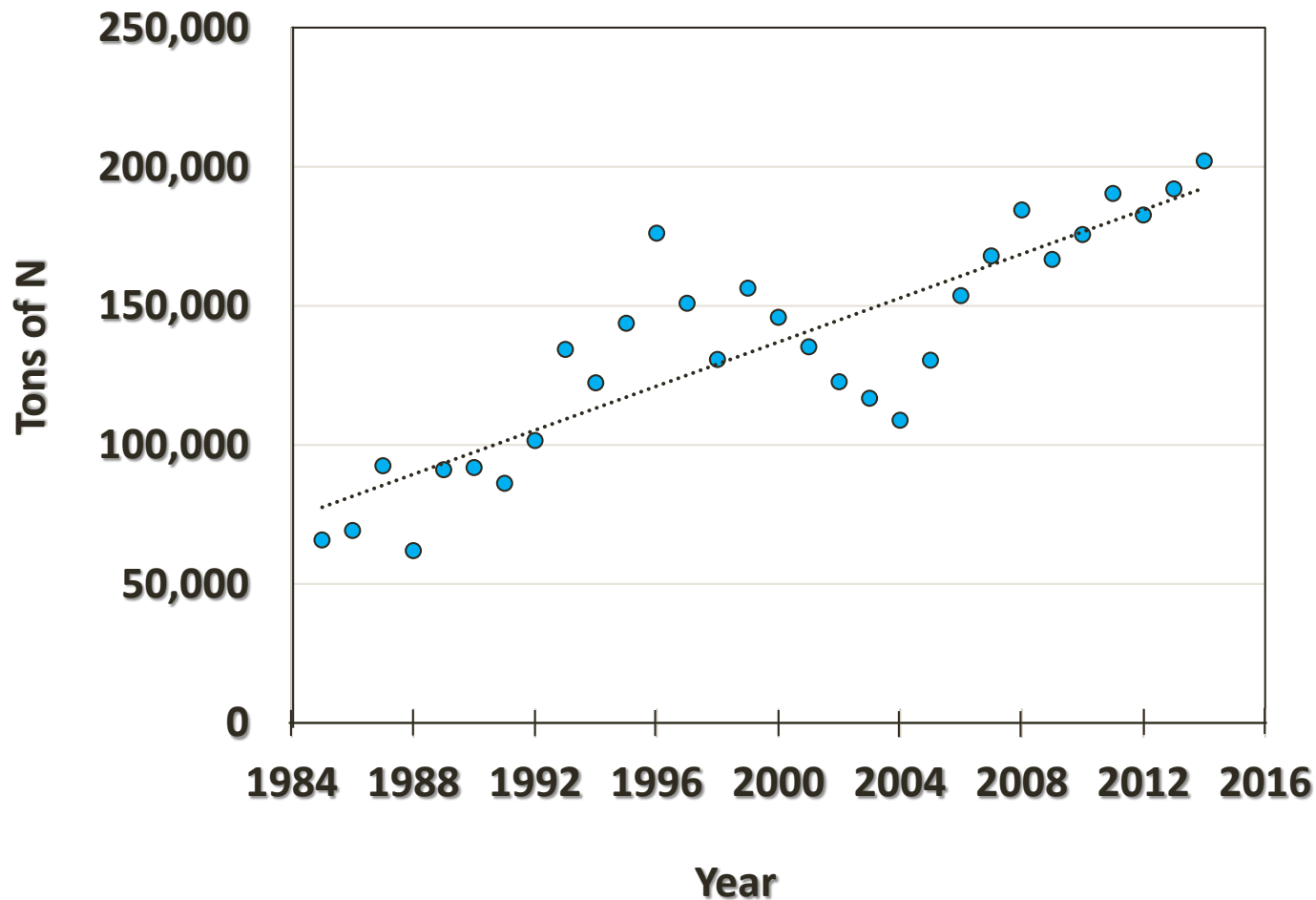
Data from Agvise (Northwood, ND)



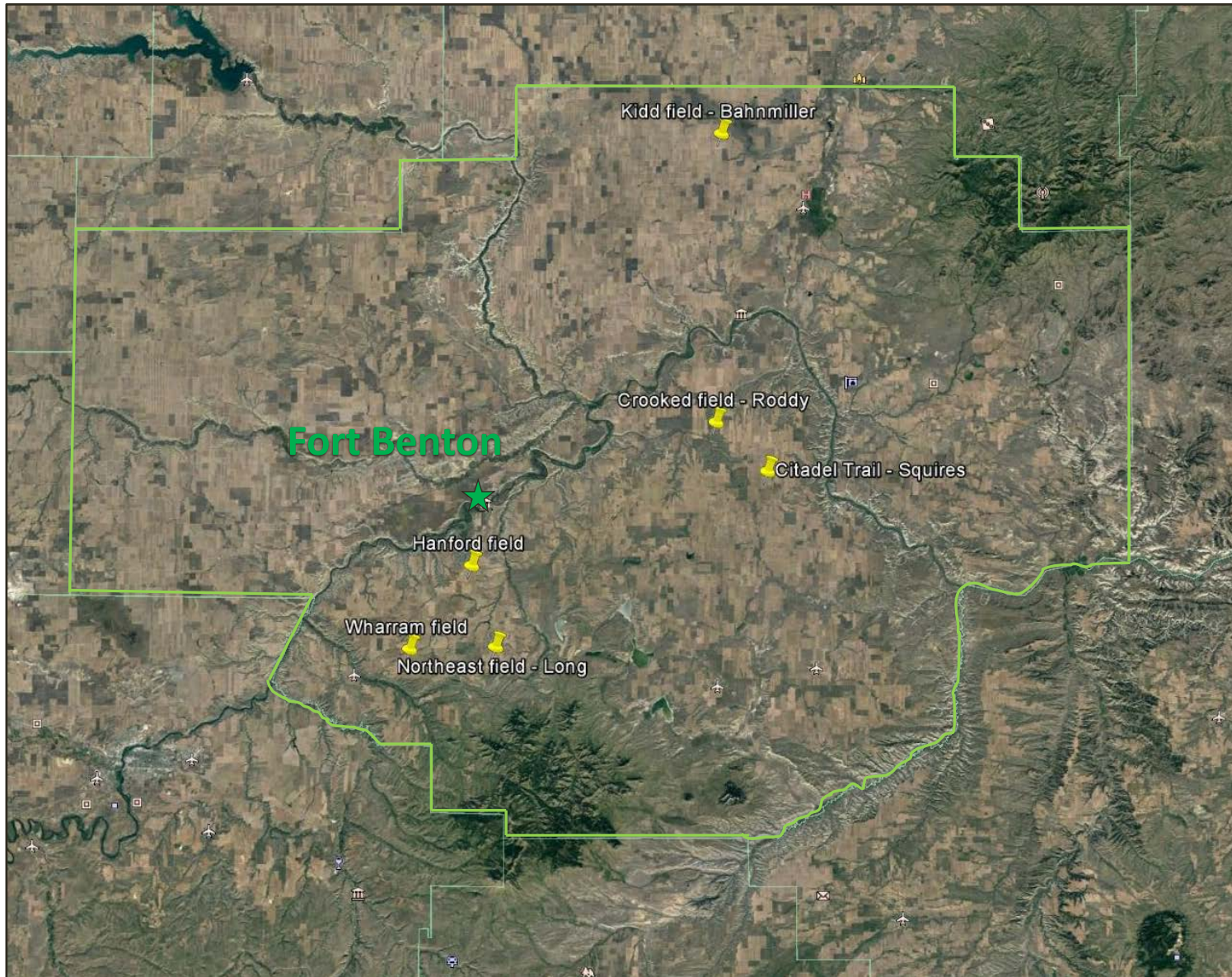
Soil acidification

- Natural processes – leaching of basic cations/salts, bicarbonates occurs over time which causes soils to acidify
- Farming – has accelerated
 - no-till – absence of mixing can create pH stratification
 - summer-fallow can result in more leaching of basic cations/salts
 - residue removal - basic cations in the crop residue are removed from the soils
 - fertilizer N – ammonium forms ***

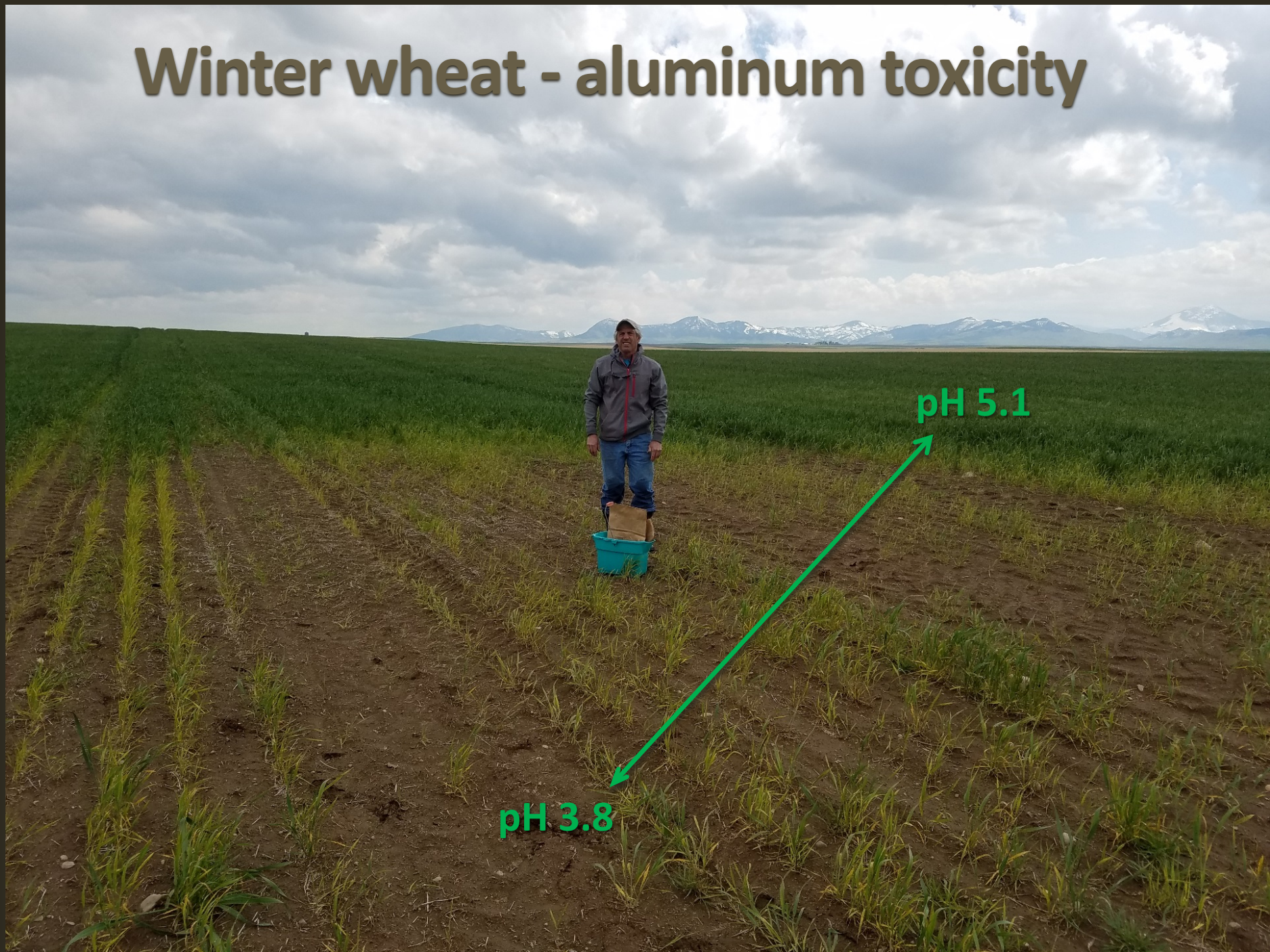
Montana N fertilizer consumption



Chouteau County field locations



Winter wheat - aluminum toxicity



pH 5.1

pH 3.8

Durum wheat – aluminum toxicity

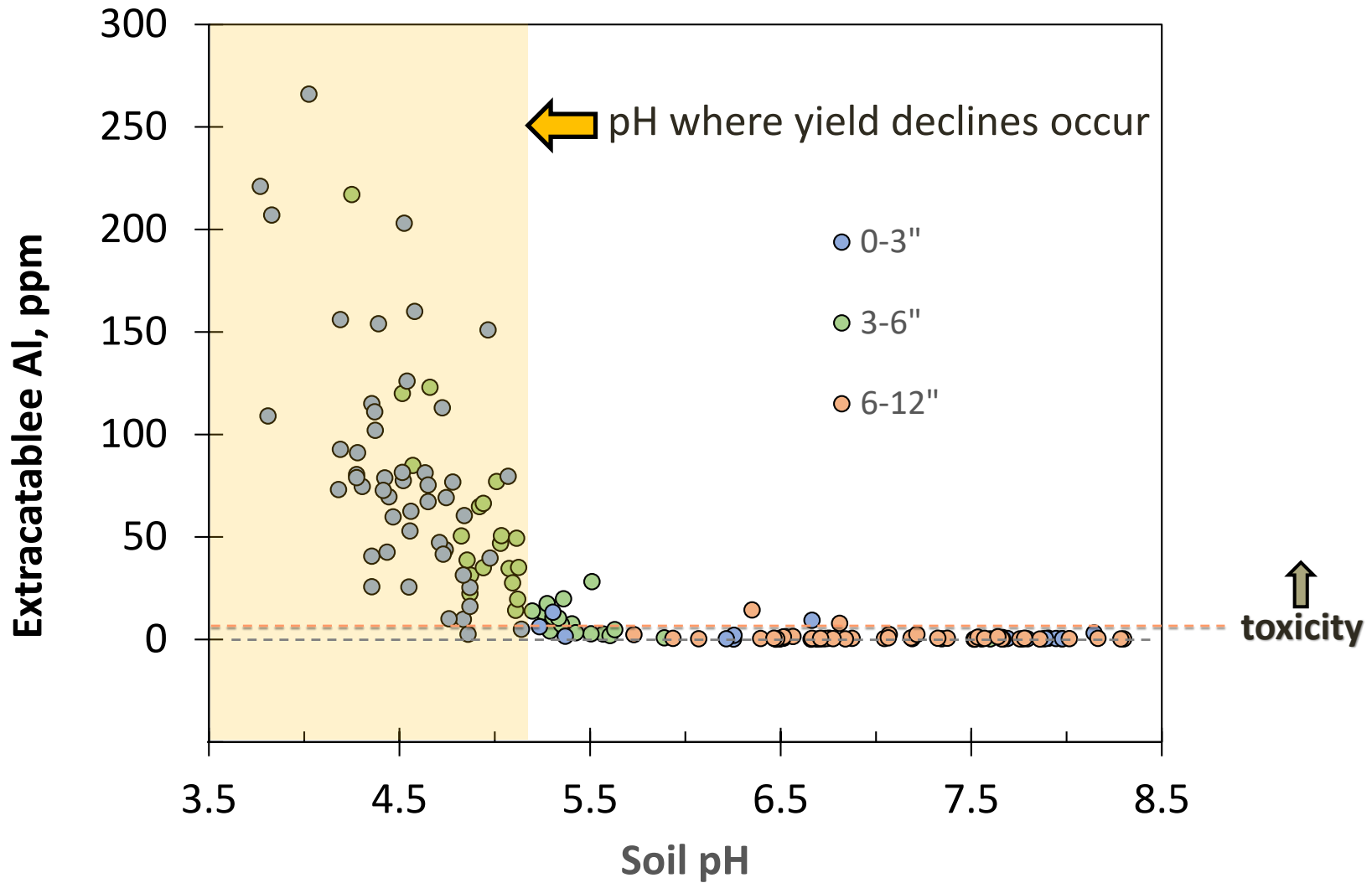


Plant Symptoms of Al toxicity

- Roots: witch's broom roots, thickened, twisted, club ends, stubby, no fine branching



Low pH increases Al to toxic levels

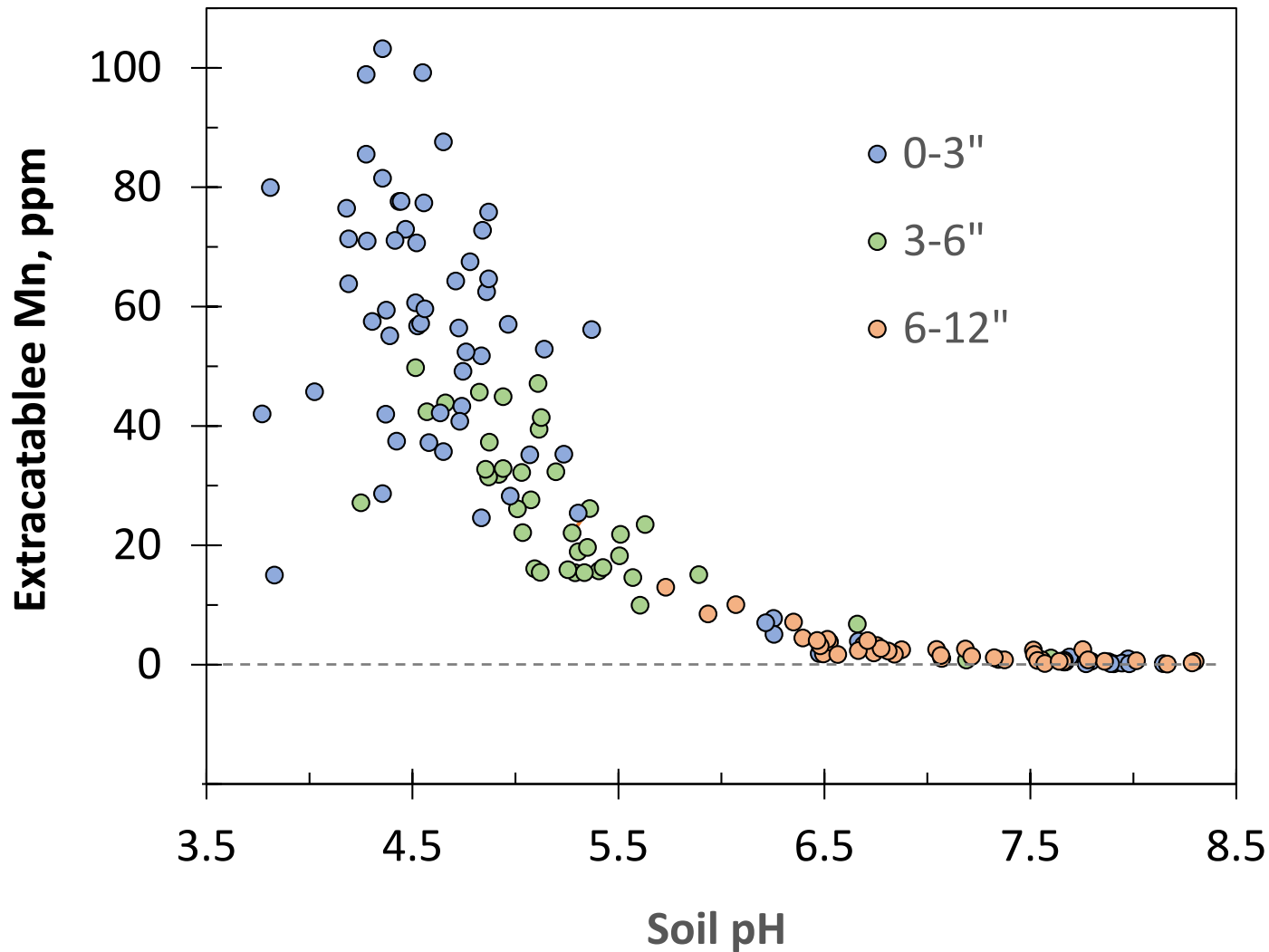


Engel unpub. data, 2016, 5 farms near Highwood, MT

Why does soil pH affect aluminum?

- aluminum toxicity is most important growth-limiting factor in acidic soils
- all Montana soils contain considerable aluminum; very abundant element in the soil; 6-7% of total soil dry mass
- solubility of aluminum containing minerals increases as pH becomes more acidic ***

Soil pH also affects available Mn



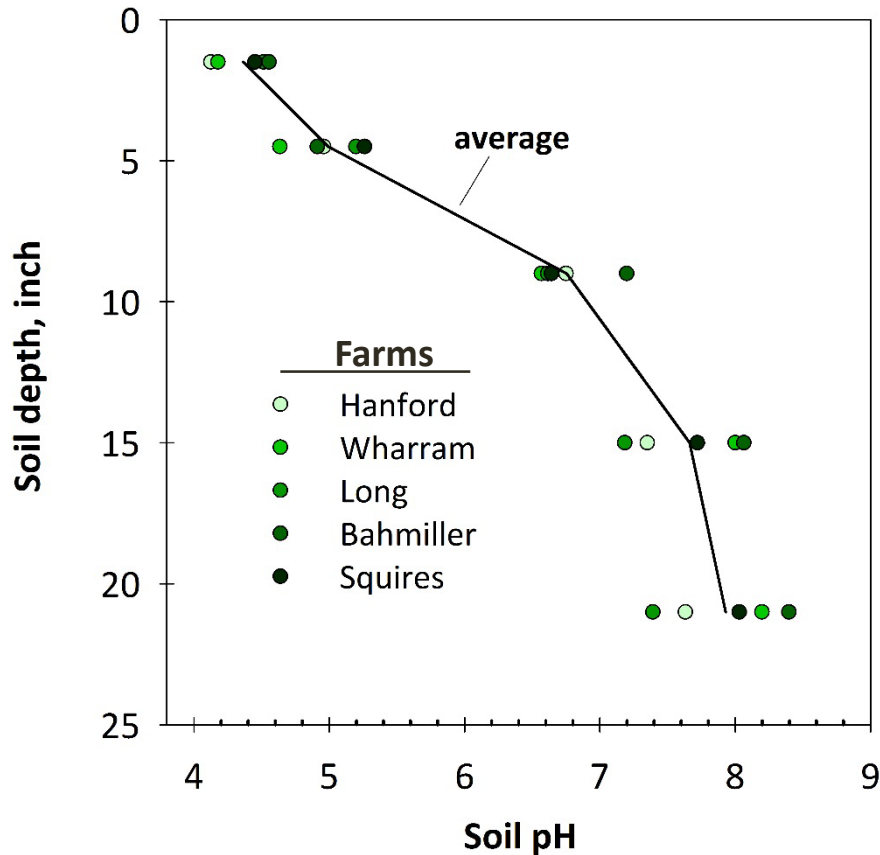
- ✓ Manganese toxicity has been associated with acidic soils - no evidence this has been a problem so far based on plant analysis

Manganese tissue conc (normal vs. poor growth areas)

| Cooperator | Tissue | Normal | Poor |
|-------------|--------|-----------------|------|
| | | ----- ppm ----- | |
| Bahnmilller | shoot | 135 | 257 |
| | | | |
| Squires | shoot | 47 | 182 |
| | | | |
| Wharram | shoot | 57 | 207 |
| | | | |
| Hanford | shoot | 57 | 127 |

Conclusion: normal vs. poor growth areas differed in Mn but below expected toxicity range >400-500 ppm

Soil pH stratification in Montana



Summary

- lowest pH likely to be found in shallowest depth because...
 1. N applied near surface
 2. subsoils have a lot of natural lime (Ca, Mg, Na carbonates).



Managing low pH

Remediation

- amendments
- tillage
- deep-rooted or perennial crops

Adaptation

- seed Al-tolerant varieties or crops

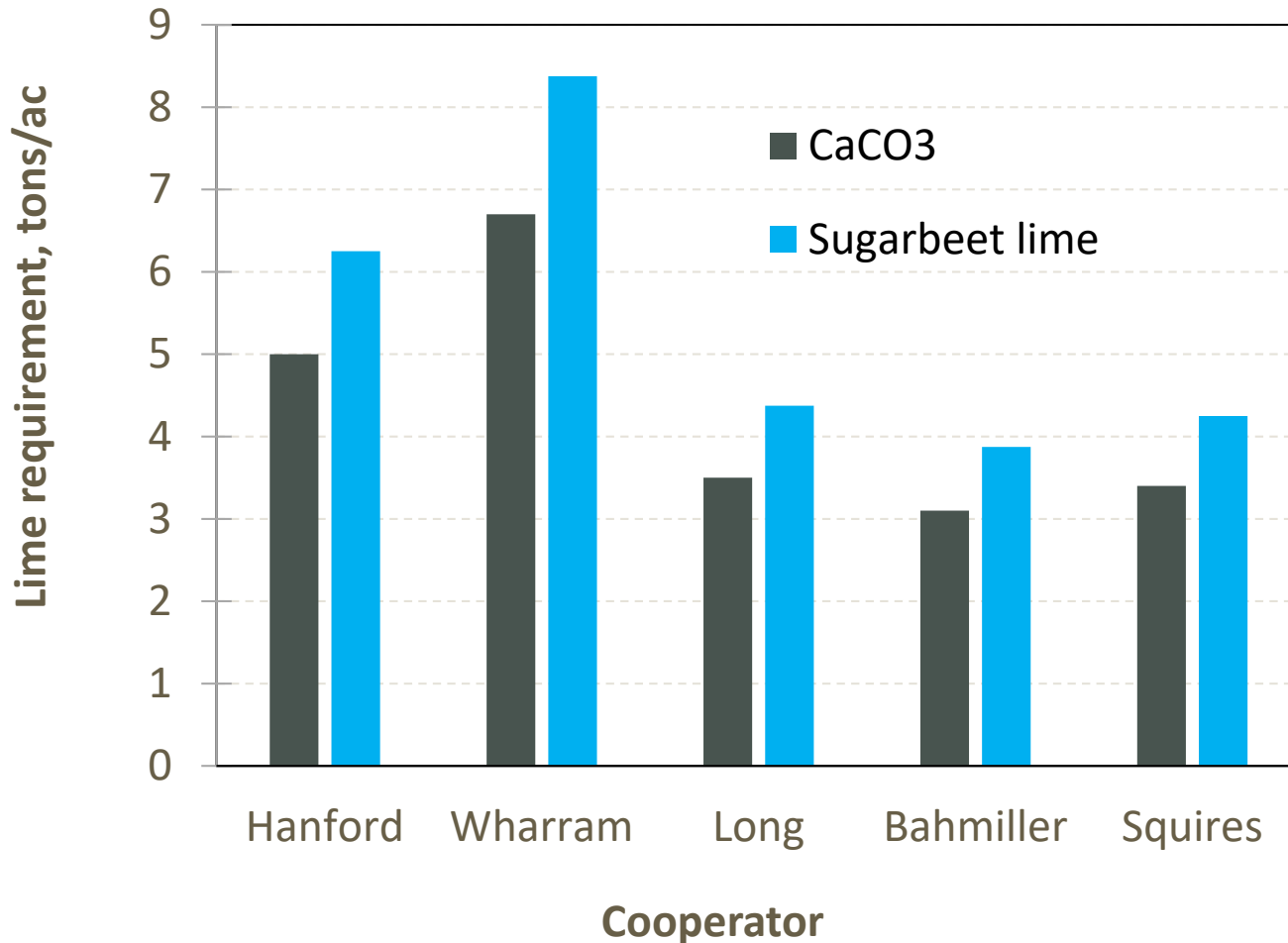
Prevention

- optimize N use efficiency - no left-over N
- N sources
- crop residue

Remediation – sugar beet lime

- good – it doesn't cost anything
- bad – shipping costs (35 \$ per ton); challenging material to work with (moisture and clumping) and incorporation is needed for best results
- rates of 3-6 tons/acre may be necessary to bring pH to acceptable level (pH 6); presently we do not have a lime recommendation test for Montana nor do we have experimental trials to evaluate its efficacy

Lime requirement (Sikora method) field sites with soil pH < 5



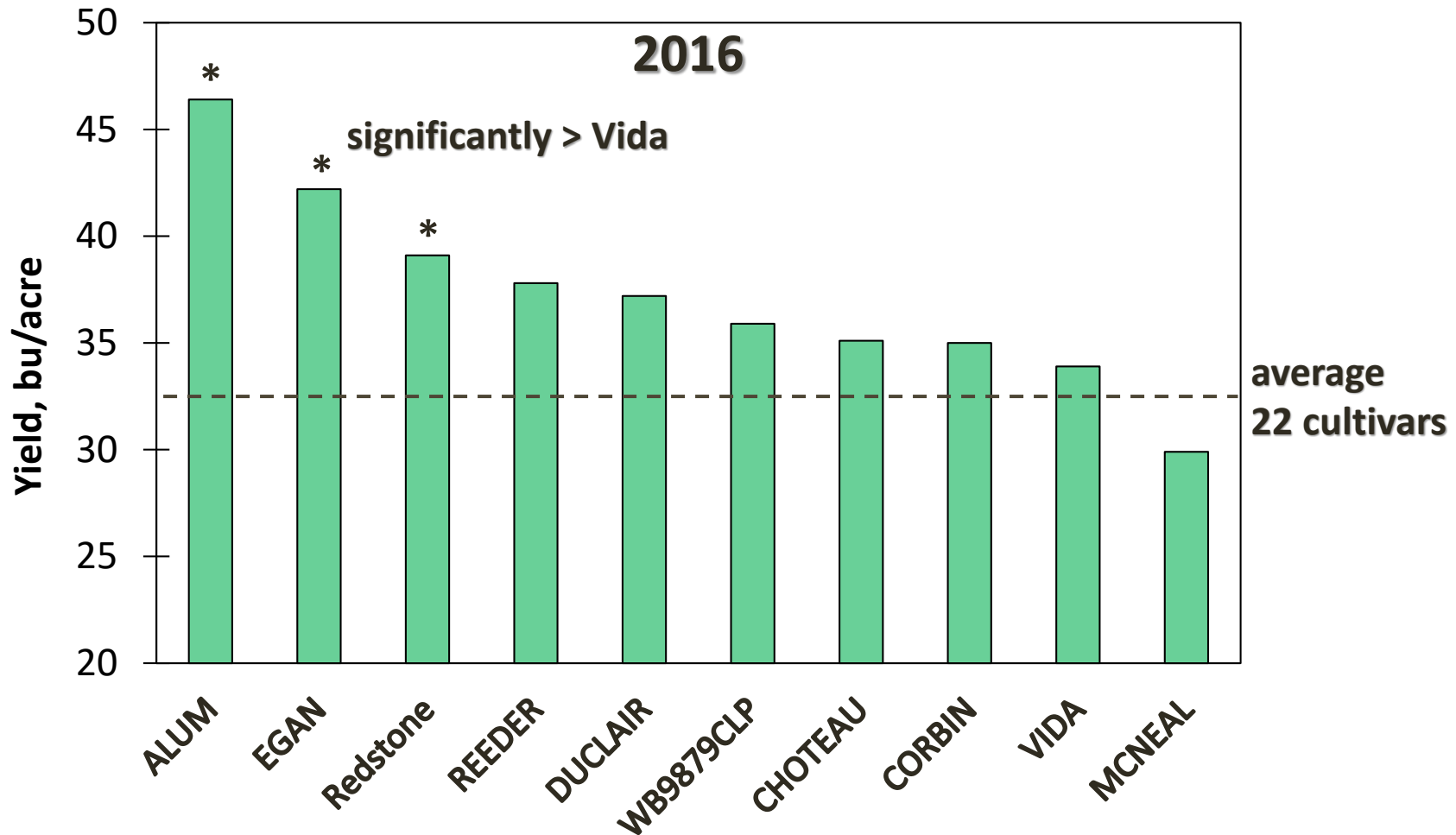
- ✓ lime requirements based on target pH 6.0;
- ✓ pH 5.5 target would require ↓ lime

Remediation – deep ripping



not sure this will work ?

Adaptation -spring wheat cultivars – Highwood Bench (soil pH 4.7)





ALUM

HARD RED SPRING WHEAT

Alum, a newly released hard red spring wheat from Washington State University, is intended to provide growers with low soil pH and aluminum toxicity a very solid and broadly adapted hard red spring wheat variety.

Alum has very good adult plant stripe rust resistance, Hessian fly resistance, above average test weight, very good aluminum tolerance, medium plant height with good straw strength, and very good-to-excellent yield potential across the PNW. Alum should be of particular interest to growers in Spokane, eastern Whitman, Columbia, and Walla Walla counties in Washington, and in northern Idaho.

AGRONOMICS

Yield Potential..... Very Good–Excellent
 Test Weight..... Very Good
 Protein..... Average
 Maturity..... Medium-Later
 Height..... Medium
 Quality..... Desirable
 Straw Strength..... Very Good

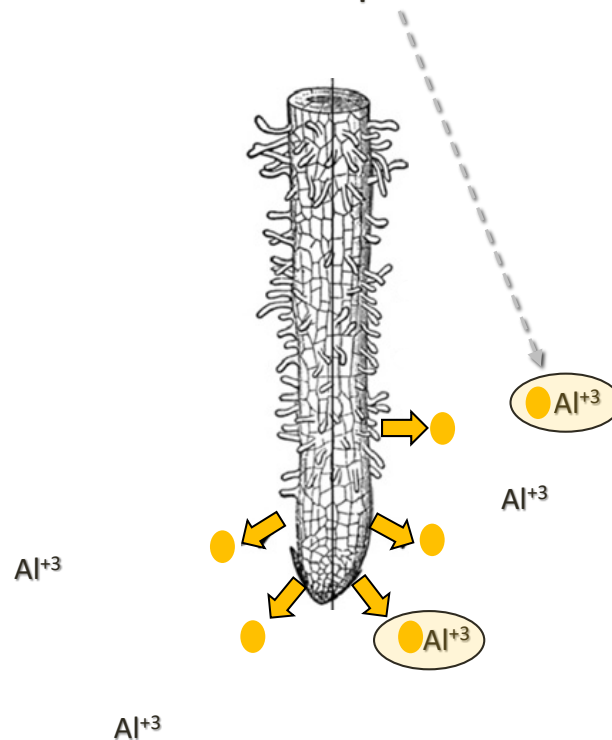
DISEASE RESISTANCE

Stripe Rust..... Very Good Adult Resistance¹
 Hessian Fly..... Resistant
 Aluminum Tolerance..... Excellent

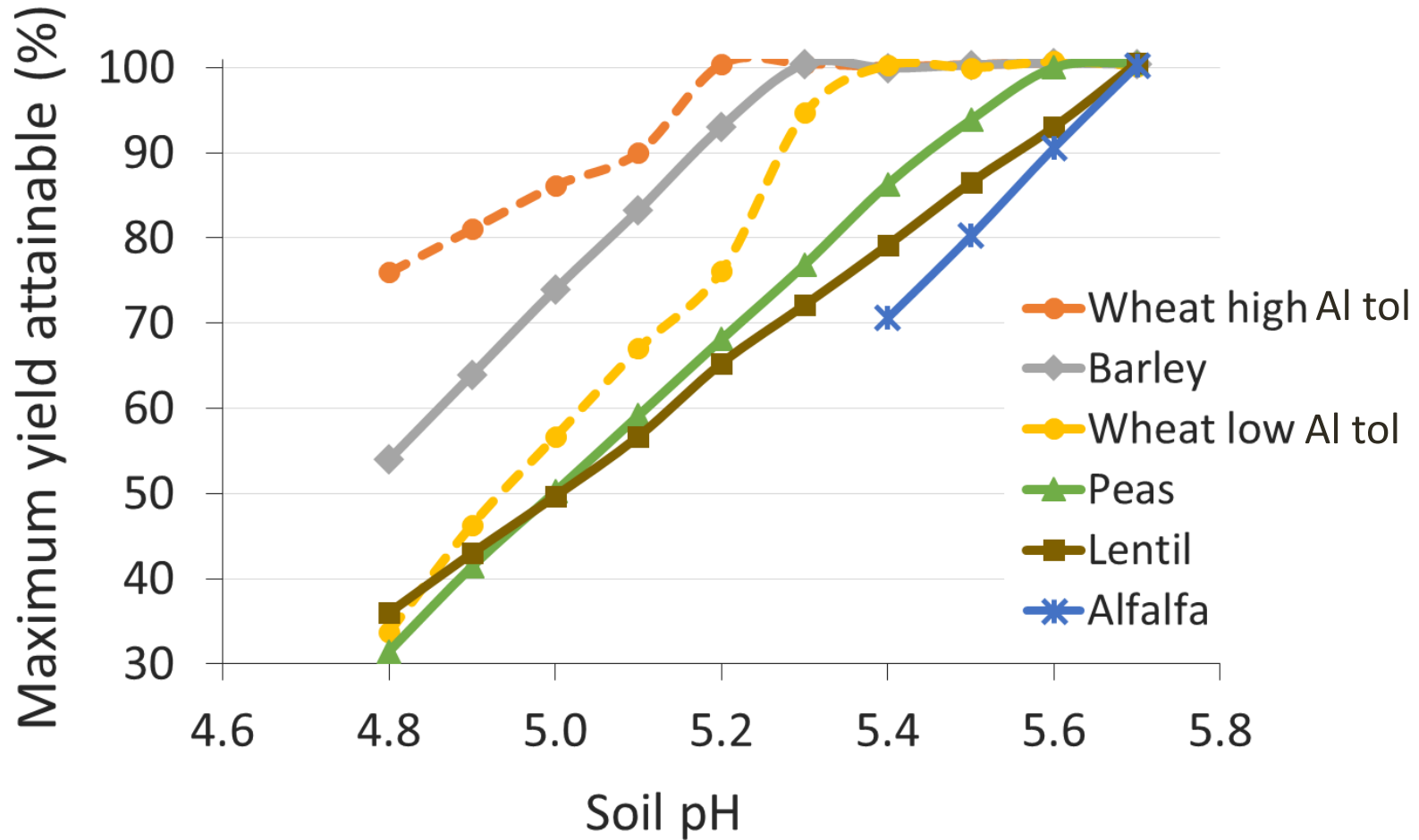
¹ Early season application of fungicides should be considered to limit seedling infection.

Aluminum tolerance – single locus (Alt1)

wheat cultivars with Alt1 release malate (organic acid) from root tips in response to high conc of solution Al^{+3} ; malate in turn chelates with Al^{+3} in the soil to form a non-toxic complex



Adaptation - crop species & tolerance to low soil pH



✓ peas and lentils are more susceptible than cereal grains

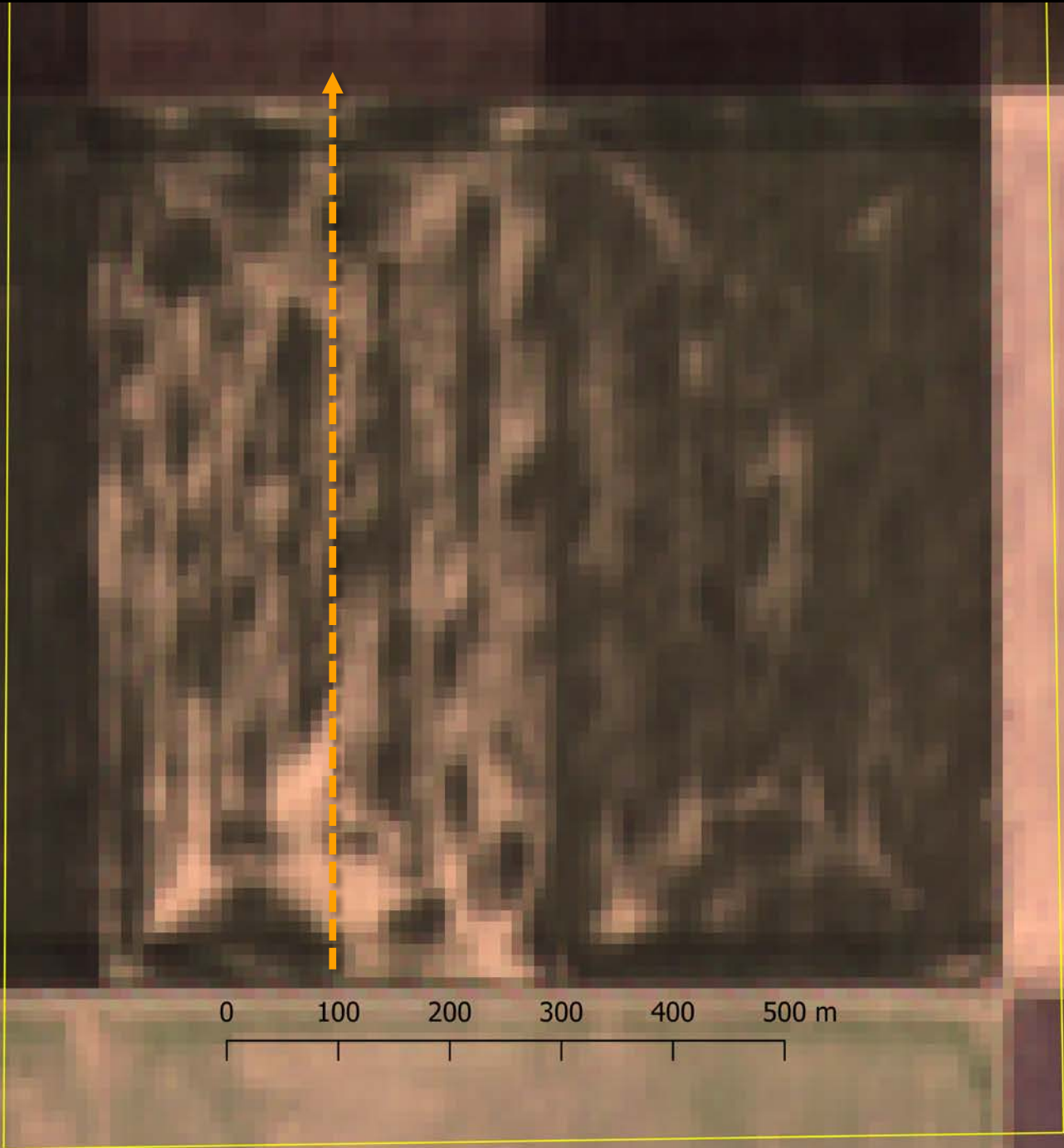
My Research - activities for 2017

- Sugar beet lime applications with tillage
- Lime requirement soil tests
- Legacy effects
- Soil pH mapping - gradients in the field are very large, acidity problem is not ubiquitous across the field landscape. Initial surveys suggest problem maybe confined to lower slope positions (relevant to \$ of liming)

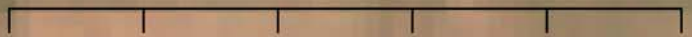


Legacy effects – an example from field





0 100 200 300 400 500 m



Legacy effects – durum & w. wheat field



| | |
|------------------|------|
| pH_{91} | 7.77 |
| pH_{90} | 8.15 |
| pH_{89} | 7.89 |
| pH_{88} | 7.98 |
| pH_{87} | 7.67 |
| pH_{86} | 6.52 |
| pH_{85} | 7.95 |
| pH_{84} | 7.80 |
| pH_{83} | 6.22 |
| pH_{82} | 4.36 |
| pH_{81} | 4.42 |
| pH_{80} | 4.28 |
| pH_{79} | 4.19 |
| pH_{78} | 4.76 |
| pH_{77} | 4.28 |
| pH_{76} | 4.45 |
| pH_{75} | 4.18 |
| pH_{74} | 4.87 |

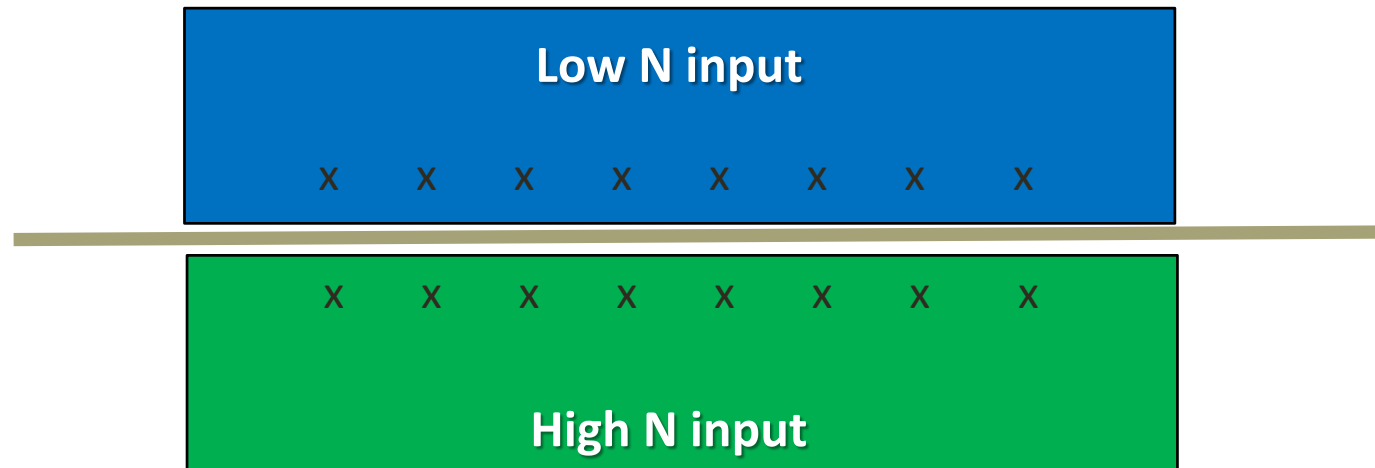
Cross-the-road (Big Sandy)

organic

conventional



Paired comparisons – high and low N input farms (legacy effects of N fertilizer)



Protocol

- ✓ soil cores (0-3, 3-6, and 6-12")
- ✓ compare pH
- ✓ extractable aluminum, exchangeable bases

Final comments

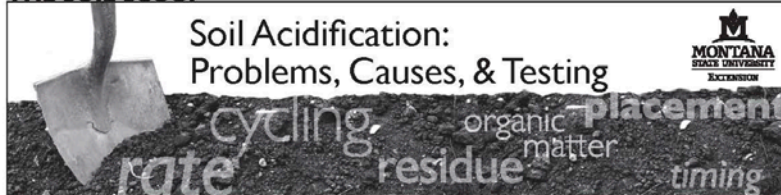
- Soil acidity is likely to increase in the coming years and decades in Montana as similar phenomena has occurred in other semiarid regions – Palouse, Great Plains
 - How many acres will be affected and how soon ??
 - Chouteau County – growers have observed problems, Highwood – higher precipitation area
- Likely will be more interest in aluminum tolerant cultivars of cereal grains
- Crop intensification & diversification - pulse acreage/production is rising; peas and lentils are very susceptible to low pH so may affect viability of some cropping systems

Soil Scoops

THE SOIL SCOOP

December 2016

Soil Acidification: Problems, Causes, & Testing



by Clain Jones, Montana State University Extension Soil Fertility Specialist, and Kathrin Olson-Rutz, Research Associate

Most agricultural soils in Montana have near-neutral to basic conditions with surface soil pH 6.5 to 8. However, some areas of Montana have acidic soils, especially in the seeding zone. The decrease of soil pH in the seeding zone can be relatively rapid. Idaho agricultural soils went from neutral pH to pH less than 6 in about 25 years. The Soil Scoop *Soil Acidification: Management* presents management options. More details on soil pH and references are available in *Soil pH and Soil Organic Matter*.

SOIL pH

Soil pH is a measure of the soil solution's acidity on a scale from 0 to 14. By definition, $\text{pH} = -\log[\text{H}^+]$, where $[\text{H}^+]$ is the hydrogen ion concentration. The negative sign means as pH goes up, acidity (and H^+ concentration) goes down. Acidic soils have low pH values (<7), basic soils have high pH levels (>7), and pH 7 is neutral. Each pH unit change represents a 10-fold change in acidity. For example, a soil with pH = 6 is 10 times more acidic than a soil with pH = 7.

AGRONOMIC CONCERNS OF LOW SOIL pH

- Aluminum (Al), H^+ , and manganese (Mn) toxicity lead to yellow foliage and poor growth. Aluminum toxicity creates plants with club or broom shaped roots (Figure 1).



FIGURE 1. Club roots of wheat from Al toxicity.

For more soil fertility information and resources see <http://landresources.montana.edu/soilfertility/>

- Plants are hungry due to low soil nutrient levels (Figure 2). Phosphorus (P) is tightly bound to clay or iron particles, while nitrogen (N), potassium (K), sulfur (S), calcium (Ca), and magnesium (Mg) are easily leached and lost from the soil.

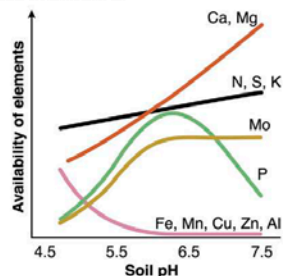


FIGURE 2. Nutrient availability as soil pH changes. Source: Government of Western Australia, Department of Agriculture and Food

- Reduced rhizobial growth; legume (e.g., pea, alfalfa) nodulation/N fixation is severely limited at pH <6
- Increased fungal diseases (e.g., *Rhizoctonia*)
- Changes in herbicide/pesticide effectiveness and residual (e.g., Group 2, B herbicides)

CONDITIONS CONTRIBUTING TO LOW SOIL pH

- Ammonium-based N fertilizer above plant needs (Figure 3)
- No-till (concentrates acidity in 3-5" rooting zone), although even with 9" moldboard plowing N fertilizer above plant needs caused soil acidification in the 0 to 9" soil layer (Rasmussen et al., 1989).
- Soils with high sand content and/or low levels of soil organic matter (SOM)

THE SOIL SCOOP

December 2016

Soil Acidification: Management



by Clain Jones, Montana State University Extension Soil Fertility Specialist, and Kathrin Olson-Rutz, Research Associate

Most agricultural soils in Montana have near-neutral to basic conditions with surface soil pH 6.5 to 8. However, some areas of Montana have acidic soils, especially in the seeding zone. The Soil Scoop *Soil Acidification: Problems, Causes & Soil Testing* introduces soil pH, why it is a concern, agronomic practices that contribute to soil acidification, and soil testing for pH.

MANAGEMENT: PREVENTION & MITIGATION

- Use practices and rates to optimize nitrogen (N) use efficiency – no left-over N. Base N rate on spring soil tests and realistic yield potential, split N applications (and don't apply 2nd time in dry springs), and place N in the root zone.
- Reduce nitrate (NO_3^-) loss. Use slow-release N sources or N sources with nitrification inhibitors, and plant deep rooted crops to 'catch' deep nitrate. Deep rooted crops can also pull base-forming cations (Ca^{2+} , K, Mg^{2+}) from the subsurface to the surface.
- Use calcium ammonium nitrate (27-0-0) which has less acidifying potential than urea (46-0-0).
- Use pulse crops in rotation – they don't need N fertilizer
- Plant aluminum (Al)- or low-pH tolerant crops (Figure 1) or varieties such as Judee, Wesley, Overland, or AAC Gateway (from 1 Oklahoma study)

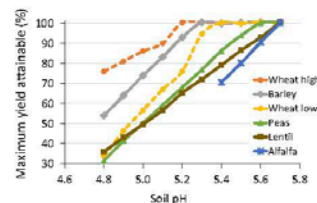


FIGURE 1. Crop species vary in tolerance to low pH, or high aluminum levels (McFarland et al., 2015).

For more soil fertility information and resources see <http://landresources.montana.edu/soilfertility/>

- Inversion till to mix acid zone throughout plow layer and bring up calcium carbonate (CaCO_3) from deeper layers. A soil with 5% CaCO_3 , typical in Montana, contains 100 tons of CaCO_3 in the top foot. One-time summer tillage doesn't negate long term benefits of no-till (Norton et al., 2014).

- Increase soil organic matter (SOM) to buffer pH changes and reduce Al, manganese (Mn) and H^+ toxicity. Leave crop residue in field to retain base cations or apply manure. Replace fallow with crops or cover crops.

- Seed place lime to compensate for annual N application or amend whole field with several years' of potential lime need. Broadcast conventional ag-lime needs to be tilled to affect the seeding zone. Other options are presented below.
- Band P with seed (binds some Al).

LIMING

Liming material reacts with carbon dioxide and water in the soil to yield bicarbonate (HCO_3^-), which takes H^+ and Al^{3+} (acid-forming cations) out of solution, raising soil pH. The benefits are varied and depend on the soil pH level reached (Table 1).

Source: Different materials have different 'potency' to raise soil pH. Calcium carbonate equivalent (CCE) compares a liming material to pure CaCO_3 . Lime Score (LS; Table 2), also called effective neutralizing value (ENV), combines CCE with factors for moisture and fineness to calculate

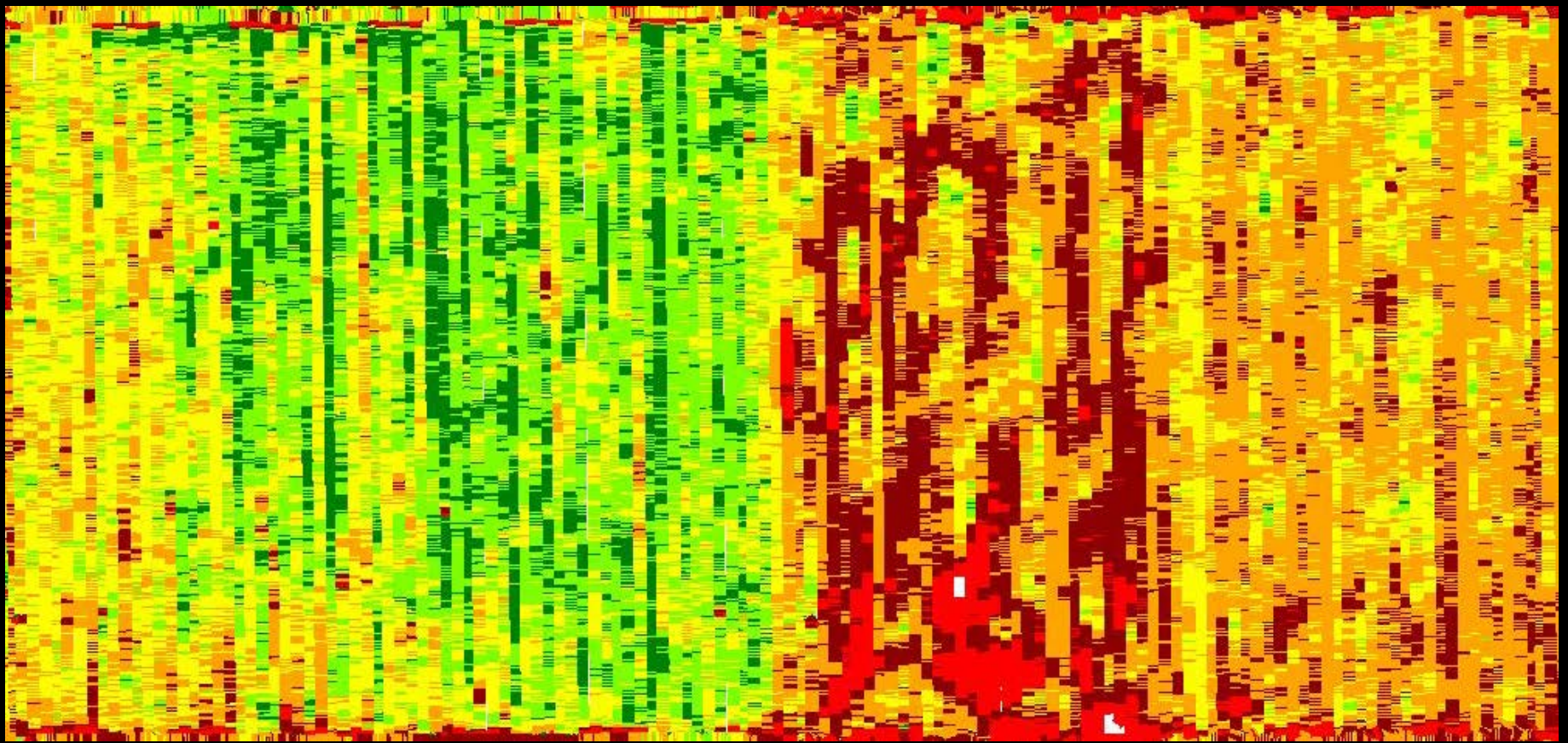
Table 1. The benefits of increasing pH of acidic soils*.

| Soil pH | Effect |
|-----------|--|
| 6.1 - 6.5 | Improve soil structure, reduce crusting, reduce power need for tillage |
| 5.6 - 6.0 | Increase soil microbial activity, increase rhizobia health for nitrogen-fixation and other mycorrhizal assisted crops (legumes and barley); increase plant nutrient availability |
| 5.1 - 5.5 | Reduce Al, H^+ , and Mn toxicity, increase availability of P and other nutrients |
| <5.1 | Few crops can produce if not limed |

* Alberta Agriculture & Forestry

A photograph of a herd of elk in a field during sunset. The elk are in the foreground, some facing left and some right. The background is a hazy, golden landscape with trees and bushes. The text "Thank you" is overlaid in the upper right corner in a bold, black, sans-serif font.

Thank you



Legend - Map data contains points with no GPS information.

| Color | Minimum | Maximum |
|-------|------------------|---------|
| ■ | 46.70 or Greater | |
| ■ | 37.69 | 46.70 |
| ■ | 30.50 | 37.69 |
| ■ | 23.57 | 30.50 |
| ■ | 12.76 | 23.57 |
| ■ | Less than | 12.76 |

Aluminum toxicity and P



double seeded durum and P