

The Last Best Place: Problems and Solutions at the Wildland Urban Farmland Interface

Foreword

By Beowulf Boswell and Patrick Lingle:

The Greater Yellowstone Ecosystem (GYE) contains over 22 million acres of land, providing crucial habitat for elk herds, grizzly bears, wolves, cutthroat trout, and numerous other species. For more than 11,000 years, humans have inhabited the GYE alongside these wildlife species (NPS 2025b). From early native tribes to modern day cities such as Bozeman, the GYE has seen a large increase in urbanization. In recent years, Bozeman's population has increased from 37,000 in 2010 to nearly 60,000 in 2025 (World Population Review 2025). With this rapid growth, issues such as habitat loss and habitat fragmentation have become a threat to wildlife across the GYE. Rapid urbanization leads to decreased biodiversity and habitat degradation (Chen et al. 2025). Grassland birds are an excellent example of the impacts of urbanization.

Habitat fragmentation in the GYE is not limited to the construction of apartment buildings or single-family homes, as changes in land use and cover also impact habitat. As cities expand, the number of structures vulnerable to fire also increases. As a consequence of this increase, more effort and more chemicals are being used to suppress wildfires in the wildland urban interface (WUI). In addition to the physical destruction of buildings, excess nutrients and harmful PFAS chemicals are introduced to the soil. These additions can contaminate water supplies and alter plant growth.

Lastly, transportation corridors and human barriers can also impact habitat, migration routes, and mortality rates (Malcom 2018). Major highways like Interstate 90 and US Highway 191 also act as a physical barrier for migration. In addition to highways, other physical barriers such as fences can make accessibility more difficult for migratory wildlife. These barriers can block off crucial habitat or routes to habitat, especially for elk populations within the GYE. Physical barriers are not limited to road barriers, river structures like dams and reservoirs can also impact salmon and trout migration routes and habitat. Understanding how to mitigate these new issues is critical in preserving local wildlife populations and migration routes. If these issues are not addressed, habitat areas will continue to decrease.

Agricultural systems have a large impact on habitat. The physical alteration of the landscape as well as possible water contamination from fertilizers and pesticides impact both fish and wildlife habitats (Skoog et al. 2024). Cutthroat trout are one of many species impacted by the increased sediment load from tilling.

Agricultural lands are a recent part of the equation for managing the wildland-urban interface. As water availability becomes less predictable, producers across the West face growing pressure to adapt to a shifting climate. Many dryland grain farmers are shifting from traditional fallow rotations to continuous or cover cropping systems that increase soil organic

matter, enhance water infiltration, and stabilize yields under dry conditions. However, rising temperatures and increasingly erratic precipitation patterns have added an additional layer of uncertainty, forcing producers to adjust planting schedules and adopt water-efficient management strategies to maintain soil health and yield stability. These same climate pressures make building soil resilience more important than ever.

Compost use has become one of the most effective ways to strengthen that resilience, restoring soil structure, boosting water-holding capacity, and supporting the microbial activity that sustains fertility over time. Composting is consistent with regenerative agricultural practices that prioritize soil health, animals, and people for treating the farm as an interconnected ecosystem and reducing the environmental harms of industrial agriculture. Together, these strategies illustrate how agricultural innovation can buffer farms against drought and degradation while reducing dependence on synthetic inputs.

Organic and regenerative agriculture emphasize the importance of minimizing the use of chemicals. Heavy use of glyphosate, the world's most common herbicide, leaves residues that harm crops and native plants while diminishing soil microbial diversity. The same is true for overused insecticides, which can wash into streams. A more sustainable approach relies on parasitoid wasps and other biological controls that limit pests without damaging ecosystems.

In addition to agricultural chemicals, water in the state can be contaminated from natural sources. In Montana, groundwater can harbor arsenic, uranium, and bacteria including *Mycobacteria* and *Legionella*, creating significant health risks for private well users. While treatment systems are available, they are expensive and challenging to maintain. Meanwhile, PFAS, so-called "forever chemicals" originating from firefighting foam and industrial runoff, represent a new class of persistent pollutants. These compounds accumulate in sediment and food webs, persisting for decades. Recent studies indicate that biochar filtration may provide an affordable and sustainable method for eliminating such contaminants. Despite these efforts, engineered fixes have consequences: dams trap sediment, slow water flow, and create low-oxygen zones that turn chemicals into their most toxic forms. This illustrates that even carefully designed solutions can introduce new challenges, highlighting the need for cautious and informed approaches.

The Gallatin Valley, part of the Greater Yellowstone Ecosystem, shows how development, agriculture, and climate pressures converge around water. Once dominated by farmland and open range, it has become one of Montana's fastest-growing regions, with Bozeman's population nearly doubling in just over a decade. That population growth, layered over a long history of cropland and hayfield irrigation, has pushed both surface and groundwater systems to their limits. The same rivers that shaped the valley's economy now reflect the cumulative effects of diversion, groundwater pumping, and warming temperatures.

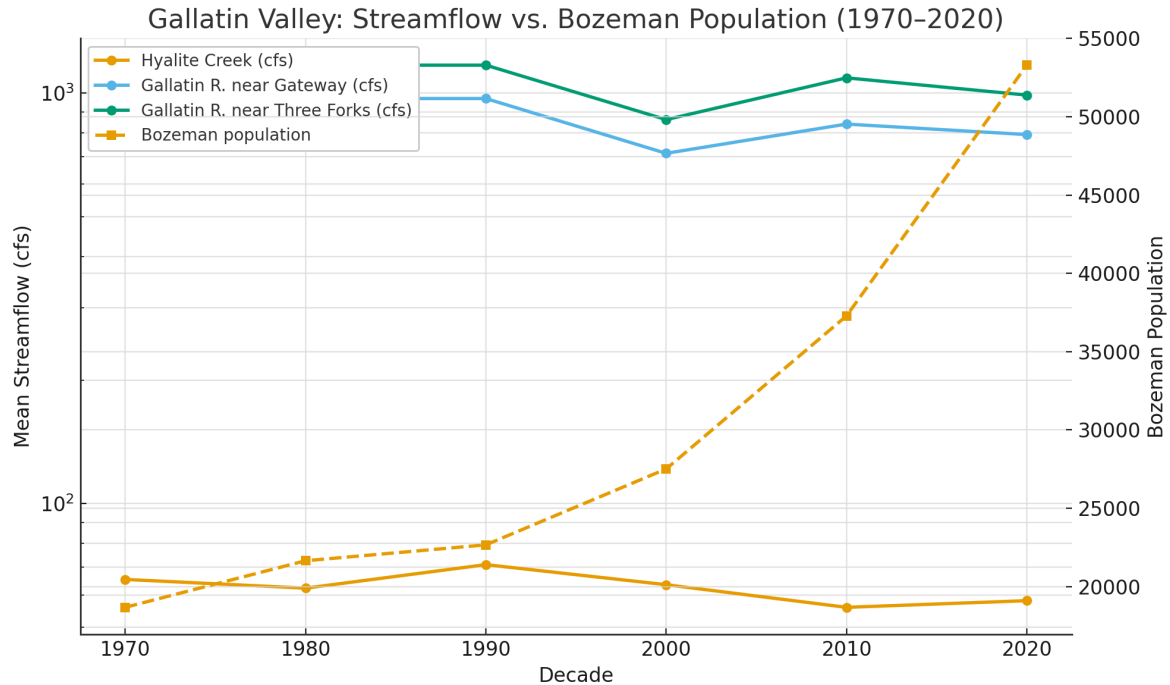


Figure 1. Decadal mean discharge (ft^3/s) for Hyalite Creek, Gallatin River near Three Forks, and Gallatin River near Gateway (log scale). Compared to Bozeman population (linear scale)

Table 1. Linear and log-scale slope results summarizing discharge decline rates

Site	Log Slope	Slope %	Slope CFS
Hyalite Creek	-0.003	-3.005	-0.187
Gallatin River near Three Forks	-0.003	-2.655	-3.144
Gallatin River near Gateway	-0.004	-4.323	-4.05

Long-term U.S. Geological Survey records from Hyalite Creek and the Gallatin River confirm a clear, basin-wide decline in streamflow since the 1970s (Table 1). However, the magnitude and drivers of this decline vary across the valley. In the upper basin, Hyalite Creek shows a relatively modest reduction in flow that aligns with climatic trends—earlier snowmelt, reduced snowpack, and shifting runoff timing. Because Hyalite receives minimal direct withdrawals and has limited anthropogenic alteration, its decline reflects the broader climatic pressures acting on headwater systems across the northern Rockies. However, these small reductions matter, as they shorten the late-summer baseflow period.

In contrast, mid- and lower-basin sites show steeper declines linked to the valley's growing water use. The Gallatin River near Gateway experiences the sharpest drop in discharge because it is at the center of agricultural diversions, groundwater pumping, and delayed irrigation return flows. Unlike headwater streams, this location receives almost no additional inflow from

side tributaries, meaning every gallon removed for irrigation in the Gallatin Canyon directly reduces discharge. Downstream near Three Forks, the cumulative effect becomes clear: even with some return flow re-entering the system, the total amount of water leaving the valley is decreasing, indicating that withdrawals, and shifting hydrologic timing now exceed natural replenishment. The rest of the valley is declining faster than Hyalite, with human use, not climate alone, driving the majority of the observed reductions.

Regional research across the Interior West shows that declining streamflows are occurring throughout snowmelt-fed river systems. More than 200 monitored watersheds across the Colorado, Columbia, and Missouri River basins exhibit 10–20% reductions in late-season discharge as irrigation intensifies and meltwater arrives earlier in the year (Ketchum, 2023). Statewide analyses show similar patterns, with long-term declines in peak flows across Montana and northern Wyoming accompanied by shifts in runoff timing driven by warming temperatures and reduced snowpack (Sando, 2025). These regional trends mirror the underlying processes shaping conditions in the Gallatin Valley, where altered flow timing and reduced discharge interact with expanding surface-water pathways to drive warmer, slower late-summer streams.

Yet declining discharge is only part of the hydrologic shift occurring in the Gallatin Valley. As more water is routed through an expanding network of irrigation ditches, subdivisions, and channels, the total surface area of flowing water across the landscape has increased dramatically. This redistribution exposes water to more solar radiation, warmer air temperatures, and longer residence times, allowing it to heat up before re-entering natural streams. This warmer return flow combines with reduced main-stem discharge to create shallower, slower, and hotter late-summer conditions. This transition directly sets the stage for ecological stress downstream.

These numbers point to more than just hydrological change; they reveal a system reaching its physical and legal limits. The valley's groundwater aquifer is over-allocated, and each new domestic well further reduces stream baseflow. Under Montana's "use it or lose it" water rights framework, senior right holders must continue diverting their full allotments each season or risk forfeiting them, leaving little incentive to conserve. This combination of growth, policy, and natural decline creates a feedback loop where even modest reductions in snowpack or runoff cascade into larger ecological and economic impacts. Warmer streams hold less oxygen, forcing species like westslope cutthroat trout into shrinking cold-water refuges, while lower flows reduce dilution of nutrients and contaminants, worsening water quality.

Despite these challenges, local adaptation is underway. Some new developments and agricultural operations are experimenting with greywater reuse, low-energy filtration, and nanomaterial treatment systems to offset summer demand. Riparian restoration projects are also reconnecting floodplains and stabilizing banks, while compact urban planning seeks to reduce the hydrological footprint of growth. Yet, evidence indicates that surface-level interventions alone are insufficient: lasting sustainability depends on the integration of groundwater monitoring, return-flow accounting, and conservation incentives that reconcile human and ecological water demands.

Bozeman and the Gallatin Valley offer a clear picture of the modern West: a landscape where population growth, water scarcity, and innovation all intersect. The trends in streamflow

are not just indicators of loss, but warnings of imbalance. Opportunities to rethink how water is shared, measured, and valued in a rapidly changing environment.

Natural cycles in the West are now operating within a new set of human and climatic constraints. Snowmelt, fire, and flood events still happen, but their timing and strength are increasingly shaped by warming temperatures, development, and the infrastructure built across these landscapes. Dams, diversions, roads, and other human altered land cover shift how water, nutrients, and wildlife move through an ecosystem. As these pressures build, the effects become clear in wildlife, whose daily movements and seasonal patterns reveal how much these ecosystems are being reshaped.

Wildlife

Introduction

The GYE and surrounding Montana landscapes face mounting pressures from human development that threaten native wildlife across multiple ecosystems. As Montana's population has grown and urbanization has intensified, native species are forced to adapt to dramatically altered environments or face extirpation (National Park Service, 2024). Habitat destruction and fragmentation caused by highways, fencing, and urban expansion have disrupted essential ecological processes that wildlife depend upon for survival. For centuries, Rocky Mountain elk have traversed the wilderness expanse of the GYE, embarking on long-distance migrations from mountainous summer ranges to lower elevation winter ranges each year. However, human-dictated landscape fragmentation has reduced the connectivity of these migration corridors, forcing elk to navigate an increasingly obstacle-laden environment. Similarly, native birds face parallel challenges from climate change, competition with non-native species, and especially habitat destruction. Beyond terrestrial ecosystems, Montana's aquatic environments face distinct but equally pressing threats stemming from the diversion of substantial volumes from the Gallatin Valley's rivers and streams during critical growing months, proving particularly harmful to species like the westslope cutthroat trout, which relies on cold, navigable waters with ample dissolved oxygen levels throughout its life cycle (USGS, 2025). Together, these interconnected pressures from habitat destruction, landscape fragmentation, and water management practices illustrate the broad-scale and multifaceted impacts of human development on Montana's diverse ecosystems.

This section focuses on the effects of human development on three different animals in Montana's ecosystems. The species of interest are elk, native birds, and westslope cutthroat trout.

Habitat Fragmentation Effects on Elk

By Maxwell Opton:

Every year, elk herds embark on long-distance migrations from mountainous summer ranges to lower elevation winter ranges. These migrations are essential for accessing seasonal forage and reproductive success within the herd. On a larger scale, elk transport nutrients across elevation gradients, help to shape plant communities, and influence predator dynamics (Sawyer et al., 2013). However, human development has been expanding and intensifying in this region,

disrupting these migrations. Within the GYE, housing density has tripled in the last 50 years (National Park Service, 2024). Human-dictated habitat fragmentation caused by highways, fencing, and overall urban expansion has reduced the connectivity of this landscape.

Yellowstone elk historically spent their winters in the foothills and plains before climbing into the mountains for summer (Skinner, 1925). Elk are grazing animals that eat primarily grasses, which dictates the land they occupy and migrate to. Elk initiate movement to lower-elevation winter ranges as soon as snowfall begins to accumulate, and forage accessibility becomes limited (Smith et al., 2019). The problem is that the lower elevations of winter forage lands are now occupied by ranches and crisscrossing fences. Estimates suggest that there are over 150,000 miles of fencing within the winter range habitat for elk in the GYE (Kudelska, 2025). Elk can get tangled in fencing or injured trying to jump over them, which increases mortality. These fences are a direct cause of habitat fragmentation, representing a barrier that elk have to migrate around to reach their ranges. Habitat fragmentation occurs when a once-continuous habitat is broken into smaller, disconnected patches that are separated by areas of varying habitat type (Wilcove et al. 1986). Anthropogenically, this can be caused by development in forms of roads, trails, fences, and infrastructure. We're changing the habitat in a way that suits us, which can then have more widespread effects. The larger impacts from this include reduced species diversity, increased forms of pollution, as well as the spread of invasives (Millhouser, 2019).

Roads and highways are not only crossing hazards for elk, but they're large-scale agents of fragmentation that reshape the ecology and management of migratory herds. Displacement effects of highways can extend up to 1.1 miles from the roadway, effectively creating a wide corridor of reduced habitat utility (Ruediger et al., 2005). When highways like 14 or 212 slice through migration routes, the loss of usable habitat includes broad swaths of adjacent winter and migration habitat. This displacement both fragments of seasonal ranges and compresses elk into fewer areas of refuge. Elk may also expend more energy to avoid roads, reducing foraging efficiency, and reproductive success.

The primary way of solving this is to create large, well-designed corridors through overpasses or tunnels accompanied by strategic fencing. These crossings need to be large, sited correctly, and paired with fencing that funnels the animals to the structure (Ruediger et al., 2005). While the crossings don't restore the adjacent habitat lost from highways, they allow migratory species to navigate human-modified terrain.

Habitat fragmentation in the GYE has disrupted the natural migrations of elk by breaking up continuous landscapes with roads, fencing, and development. These barriers force herds to alter routes, avoid high-traffic areas, and rely more on private lands, leading to increased mortality risks, disease transmission, and reduced access to forage. These changes extend across the Western U.S., reflecting a broader regional pattern of habitat connectivity loss and diminished ecosystem resilience. To protect migratory elk and the ecological functions they sustain, management must focus on maintaining and restoring landscape connectivity. Conservation strategies such as wildlife crossings, zoning regulations, and partnerships with private landowners are essential for ensuring elk can continue to move freely between seasonal ranges. Protecting these migrations will not only sustain elk populations but also help preserve the ecological balance of the GYE.

Native Birds Impacts

By Brooks Taylor:

With so many people moving to Montana, native birds face obstacles including climate change, competition with non-native species, and especially habitat destruction. As Montana has become more urbanized, birds need to adapt to urban environments or face extirpation. Habitat destruction has severely affected native birds not only in Montana, but also throughout the entire US. Loss of habitat puts birds in situations where they are forced to deal with anthropogenic stressors. As a result, nearly 1 billion birds die each year from humans (Loss, 2015). Some of the birds that have been directly threatened by habitat destruction in the GYE include the western meadowlark and loggerhead shrike. This is especially true for grassland birds due to reliance on grasslands for their shelter and foraging.

Grassland's conversion to urban areas remains one of the largest threats to biodiversity in North America, and this directly affects grassland birds. A major threat is a lack of grassland protection. Only 4% of the grasslands in the world are protected (Peterman, 2021). Grasslands are important due to the various ecosystem services and functions they provide. Functions and services of grasslands provide food through hunting, nourishment for the soul from being in nature, and important habitat for animals such as birds, squirrels and coyotes. As grasslands are turned into urban areas, the prairie ecosystem and its benefits are destroyed. Habitat destruction has negatively impacted grassland birds because of native grassland conversion to farmlands, severely affecting grassland bird species.

A species especially affected by loss of grasslands is the western meadowlark. Western meadowlarks rely on grasslands for nesting and cover from predators (Giovanni, 2015). They primarily prefer native grasslands and are less abundant in the grasslands of introduced species. The western meadowlark, a yellow, white, and brown passerine bird, lives in native, tall agricultural grasslands, roadsides, pastures, and orchards (MTNHP, 2025). Though their numbers are stable in Montana, they are declining at a rate of about 1% per year and have lost 47% of their population since 1970 (Mackin, 2024).

Unlike the western meadowlark whose population is stable, the Sprague's pipit is under considerably more threat by habitat destruction. This small grassland bird occupies the grasslands of southern Canada and the northern great plains. Their range includes the states and provinces of Alberta, southwestern Manitoba, Saskatchewan, central and eastern Montana, central North Dakota, and northwestern South Dakota (Staufer, 2025). This white and dark brown bird, like western meadowlark, lives primarily on the ground and feeds insects. Currently, they're in dire straits, losing roughly 4% of their habitat per year for a loss of 50% of their habitat since 1975 (Cornell, 2025). As a habitat specialist that only lives in native grasslands, the Sprague's pipit is unable to live in conventional farm systems.

Though more abundant than Sprague's pipit, loggerhead shrikes are also in rapid decline. Loggerheads are generalists who can live in a variety of habitats such as prairies, agricultural fields, low trees, golf courses, and pastures with fences (Cornell, 2025). In Montana, they are found throughout the entire state but are listed as a species of concern (Cornell, 2025). They possess many characteristics of raptors such as a sharpened beak and talons, but they are much smaller than most other birds of prey. The use of thorns and barbed wire to impale their prey. Despite their adaptability to different environments, they require open spaces to capture prey

which can include rabbits and lizards. As of now, they've lost 75% of their population since 1966 and lose about 2.5% of their habitat per year.

Grasslands and the birds living in them are in peril, but what is being done to protect these areas? One effort currently is the North American Grasslands Conservation Act. This act gives 60 million dollars in grants to support conservation efforts in the US and in Montana (United States Congress, 2024). This bill is funding restoration to grasslands falling under the criteria that are threatened by crop conversion and woody encroachment (United States Congress, 2024). Additionally, the Perry Sodbuster Act of 1985 is meant to prevent farmers from cultivating land that has high erosion potential (Rollins, 2020). If land has one-third of its acreage affected by eroded soil, then the entirety of the land can't be cultivated. This means land that isn't prone to erosion and full of native vegetation can be converted to cropland. Grassland that is grazed to the bottom few inches of the soil destroy bird habitat and gives weeds a prime opportunity to invade the soil. There is no definitive evidence showing these bird species decline solely because of habitat fragmentation as very few studies have been conducted. The only way to study whether habitat fragmentation contributes to decline is doing a controlled experiment where a group of grassland birds live in a prairie, the prairie is converted into an urbanized area or farmland, and the response of the birds to the fragmentation is recorded.

Habitat destruction is the biggest driver in biodiversity loss, and it's apparent that many of the birds we know and love in Montana are at risk of losing all their habitat, and this is especially true of grassland birds. All three of these birds; the loggerhead shrike, western meadowlark, and Sprague's pipit are rapidly declining as their habitat is developed and farmed. Grasslands are declining at rapid rates as the need for urban development is valued more than sustainability of our ecosystems. Western meadowlarks have lost nearly 47% of their habitat since 1950 and are losing 1% of their habitat per year. Loggerhead shrikes and Sprague's pipits are also losing their habitat at 1-5% per year as they slowly start to head towards possible extirpation. Habitat destruction is one of many obstacles grassland birds face that puts them in jeopardy of losing their effect on ecosystems such as vectors of seed dispersal and prey for other animals.

Westslope Cutthroat Trout Ecosystem Impacts

By Samuel Gabrielson:

A significant element of Montana water law is the application of the "use it or lose it" principle. Since most water sources are over-appropriated in Montana, those who hold water rights must put the water to beneficial use to be protected from challenges of abandonment. Abandonment of a water right includes losing the right to the next most senior right holder (DNRC, n.d.). This structure creates an incentive for rights holders to use their full allocation every year, even during typically wetter periods when conservation of the water could be beneficial, for fear of abandonment of their right. This incentive is harmful to certain ecological communities, such as that of the westslope cutthroat trout. The westslope cutthroat trout relies on cold, navigable waters with ample dissolved oxygen levels for their life cycle, which will be addressed in this study (USGS, 2025).

This framework of water allocation is visible throughout the Gallatin Valley in the form of diversion structures, ditch systems, and holding ponds. These structures represent centuries of investment in water capture and movement and can range in size from small headgates on creeks to elaborate systems on larger rivers, such as dams and levees. This infrastructure, built over the last couple hundred years, has created a highly engineered and altered landscape in which the vast majority of surface water during irrigation season is legally claimed (MSU Extension, 2025).

Irrigation intensification, land-use change, and groundwater development have collectively altered hydrological conditions across the northern Rocky Mountains. Negative summer flow responses have been documented across more than 200 basins in the Colorado River, Columbia River, and Missouri River systems, with return flows lagging by several months and proving insufficient to offset summer water losses (Ketchum, et al., 2023). A typical natural flow regime in Montana features high spring runoff driven by snowmelt, declining flows through summer as the snowpack depletes, and stable baseflows during fall and winter maintained by groundwater contributions (USGS, 2025). The decline observed at the Gallatin River near Three Forks reflects this pattern, suggesting that irrigation withdrawals and delayed return flows reduce available discharge in streams during the critical July–September low-flow period. These regional findings align with patterns observed in the Gallatin Valley and indicate that declining streamflows are driven by both climatic variability and intensified land and water use (Sando, et al., 2025).

The Gallatin Valley's aquifer system is now considered over-allocated. Thousands of domestic and subdivision wells draw from shallow groundwater sources that are hydraulically connected to surface water. Due to the Gallatin Valley's unique geology, the aquifer is segmented, which prevents continuous flow throughout (Rose and Waren, 2022). This reduced linkage means that each new well can incrementally reduce stream baseflow, complicating the enforcement of senior surface-water rights and creating a legal quagmire within Montana's prior-appropriation framework. Continued subdivision development, conversion of rangeland to residential properties, and fragmentation of riparian corridors have likely reduced aquifer recharge and further exacerbated summer low-flow conditions throughout the basin.

The Gallatin Valley data are consistent with regional studies of hydrological change. The observed trends support a narrative of converging climatic and anthropogenic pressures driving reductions in late-summer streamflow. In the upper valley, Hyalite Creek exemplifies the sensitivity of headwater systems to minor changes in baseflow. Mid-basin reaches, such as the Gallatin River near Gateway, highlight the cumulative effects of irrigation and municipal growth in areas of concentrated demand. The Gallatin River near Three Forks provides an integrated downstream perspective, where cumulative withdrawals and climatic changes manifest as reduced discharge at the basin outlet. The coherence of these findings with broader regional analyses suggests that hydrological change in the Gallatin Valley is both locally significant and reflective of larger-scale processes across the northern Rocky Mountains (Bell, et al., 2021).

Although this water is allocated by right, the persistence of agriculture and urban expansion in the Gallatin Valley has created unique pressure on these surface water systems. This convergence of municipal and agricultural demands has made water management in the Gallatin Valley a complex act of balancing established water rights, supporting economic growth, and protecting the hydrological integrity of natural systems which is increasingly unable to stand

against the rising competition. Removal of water for irrigation or municipal use, while natural flows are receding, only exacerbates the stress on aquatic ecosystems (USGS, 2025).

Headwater sites, though characterized by smaller absolute declines, are not exempt from this ecological risk. Even modest flow reductions can produce disproportionate habitat losses in high-gradient, coarse-bed channels typical of mountain headwaters; Hyalite Creek falls within this category (Ma, et al., 2023). Although, as reported, the recorded decline is only about 2 cfs per decade, such reductions during warm, low-flow periods can decrease velocity and depth in critical habitats, constraining ecological integrity and inducing ecological stress (Johnson, et al., 2024).

Disruption of these components in natural ecosystems due to surface water diversion causes a cascading effect through the food web. The Gallatin Valley suffers from a disrupted natural flow regime, which during the summer months (low-flow periods) causes warmer temperatures to persist in the river. Higher temperatures cause westslope cutthroat trout populations to decline, as they rely on cool waters for the increased dissolved oxygen it holds. This is not the only issue westslope cutthroat trout face due to low flow; loss of critical habitat, reduced spawning and rearing areas, and increased vulnerability to predation and disease are all increasingly common challenges (Earthzine, 2016). This limits fish from accessing their traditional spawning tributaries or eliminating tributaries all together. These issues also increase the risk of invasive species. Native fish, like the westslope cutthroat trout, are currently losing their habitat to the invasive rainbow trout. The mountain streams that westslope cutthroat trout prefer were historically too cold for rainbow trout but rising water temperatures have allowed rainbow trout to expand into cutthroat territory. This has led to westslope cutthroat and rainbow trout mating, reducing the number of genetically pure westslope cutthroats (USGS, 2025).

To mitigate the effects of surface water usage on stream ecosystems, the conservation programs for the Gallatin River have focused on a wide range of initiatives. Current and ongoing projects include improving riparian integrity, floodplain connectivity, and improving spawning habitat (*Gallatin River Drainage Physical Description*, 2020). Many of these projects are symptom focused, however, instead of mitigation focused. For example, the inclusion of groundwater analysis would deepen the understanding of declining surface flows. Irrigation and residential development increasingly draw from groundwater sources, so aquifer depletion may be masking or accelerating streamflow reductions. Systematic monitoring of well levels would help clarify this dynamic and provide a more complete understanding of the valley's water budget. Future management will require coordinated strategies that address both surface and groundwater. Conservation incentives, instream flow protections, and adaptive allocation policies remain central, but these should be paired with monitoring programs that link aquifer conditions to streamflow trends. Protecting headwater baseflows, reducing peak-season withdrawals in mid-basin reaches, and safeguarding cumulative flows at the basin outlet are all essential for maintaining hydrological stability. While being beneficial for restoring habitat and helping facilitate healthy waterways, current initiatives do not address the fundamental drivers of stream dewatering. Without addressing the usage patterns, the current long-term trends suggest that this ecosystem will continue to degrade (*Gallatin River Drainage Physical Description*, 2020).

Streamflows in the Gallatin Valley have declined significantly over the past 30–50 years, with reductions evident at Hyalite Creek, Gallatin River near Gateway, and Gallatin River near

Three Forks. These declines reflect the cumulative effects of Montana's water rights framework, irrigation intensification, rapid urban growth, and regional climatic shifts. These trends demonstrate the challenge of balancing agricultural production, municipal expansion, and ecological health in a constrained hydrological system.

In conclusion, the effects of surface water reductions have been indicated by reduced flow in each of the three sampled sites. From this, ecosystem effects of surface water reduction on the prized westslope cutthroat trout species have been analyzed. Although current conservation efforts and restoration projects have implemented a goal of improving habitat for local species, efforts will be needed to analyze surface water usage with an emphasis on reducing ecological stress. An important next step is to integrate groundwater into the analysis. Well depths and aquifer monitoring could provide a clearer picture of whether declining surface flows are being compounded by reductions in groundwater contributions to baseflow. A dual focus on surface and subsurface systems will be critical to sustaining the Gallatin Valley's ecological and economic resources under continued development and climate change. In a continually developing region of Montana, management actions will need to prepare for increased stress to surface water use.

Conclusion

The mounting pressure of human development across Montana's Greater Yellowstone Ecosystem, including the Gallatin Valley reveals a complex system of interconnected ecological challenges. From Rocky Mountain elk navigating fragmented migration corridors to native bird species losing critical nesting habitat and westslope cutthroat trout struggling in continuously warming, depleted streams; it is clear there is a human-driven trend reshaping Montana's ecosystems.

Each of the three focal species examined in this paper provide a distinct dimension of habitat disruption. Elk face physical barriers such as highways, fences, and urban expansion. Native birds receive pressure from the conversion of their habitat to agricultural and urban development. Westslope cutthroat trout endure the compounded effects of water diversion, reduced streamflow, and higher stream temperatures. Together, these three examples demonstrate the destruction and encroachment of three very different habitat compositions.

While current mitigation and restoration efforts are valuable, they are largely symptom focused. Wildlife crossings address the immediate mortality risks for elk, but ignore adjacent habitat lost due to highway driven displacement. Grassland conservation programs protect limited acreage, but cannot reverse decades of land conversion, or reduce urban sprawl. Stream restoration efforts improve local habitat conditions for trout, but do not address water right allocation framework driving the reduction of streamflow. Conservation in Montana as a whole will require a shift to integrated, large scale strategies that address the root causes of habitat degradation. This may include redesigning infrastructure to facilitate migration patterns, nesting areas, and ecosystem health.

The fate of not only elk, native birds, or westslope cutthroat trout, but all species in Montana is not predetermined, but requires action. With coordinated approaches that prioritize habitat connectivity, sustainable land use, and adaptive management, it is possible to maintain the ecological resources of the GYE and the economy of Montana's growing communities.

Achieving this balance demands the immediate recognition that continued development without ecological considerations will lead to irreversible losses in ecosystem diversity and function.

Anthropogenic and Natural Impacts on Soil and Water Quality in Intermountain West Ecosystems

Introduction

The Intermountain West is a vast region - ranging from the Sierra Nevadas in the West to the Western reaches of the Great Plains in the East. This region is diverse, containing ecosystems from alpine forests to salt deserts, and everything in between. This region is facing mounting pressures from both anthropogenic activities, and natural processes as climate change intensifies. Climate change has intensified drought conditions and wildfire frequency, and as human activity continues to expand into previously wild landscapes, these ecosystems are facing unprecedented stress. Understanding these impacts is critical for the ecosystems and the communities that depend on them.

Multiple pathways introduce and perpetuate contaminants in the Intermountain West ecosystems. Impacts at the focal point are long-term fire retardants, agricultural herbicides, waterway alterations via damming, and geogenic sources. While these sources all greatly differ in their origins, they all share a common thread: cascading effects through interconnected soil and water systems. The long-term effects of these landscape impacts are amplified by the Intermountain West's uniquely water limited conditions and biogeochemical characteristics.

The consequences of these extend beyond chemical presence. Soil acidification, microbial disruption, heavy metal accumulation, and altered nutrient cycling greatly shift ecosystem structure and function. Long-term fire retardants introduce large-scale nutrient pulses with cascading effects, herbicides threaten soil biodiversity essential for productivity, dams allow for buildups of contaminants, and naturally occurring uranium and arsenic pose health threats to communities that depend on contained waterways.

Climate change intensifies the pressures of these stressors through increased annual temperatures, reduced snowpack, and prolonged drought. The following review synthesizes current research, identifies necessary further research, and proposes management strategies that contextualize the long-term ecological costs of these short term interventions.

Fire Suppression and Soil Chemistry: Long-Term Fire Retardants in the Intermountain West

By Brodi Maidesil & Noah Heck:

Background and Context

Wildfire has been, and always will be, an impactful part of ecosystem processes. Fire has historically served as a cleansing agent. Removing overgrowth from the forest floor, cleaning litter on the prairies of the Great Plains, and serving as a benchmark for ecosystem succession.

Fire is a critical component in nutrient cycling throughout ecosystems, and in historically adapted fire ecosystems, many plants have adapted to the impacts of wildfire.

With westward settlement, the practice of prescribed burning saw a decline and was replaced by a policy of fire exclusion and extreme suppression, which has disrupted natural fire regimes and interrupted natural ecosystem processes (Fig. 2). A large factor in the disruption of these fire regimes was the “10 am” rule, instituted by the U.S. Forest Service in 1935 (Forest History Society, n.d.). This rule was that all fire starts, natural or human-caused, were to be extinguished by 10am the following day. In recent decades, management strategies have shifted back towards prescription burns as an effective management tool, but due to decades of rampant suppression and fuel accumulations, these natural ecosystems are unable to return to their historic state without large scale fire events. These disruptions to natural fire cycles provide context to understanding how modern suppression methods, particularly chemical suppressants, now influence dynamics at the soil level. This reliance on intensive intervention has created the need for widespread use of chemical fire retardants. Long-term fire-retardant applications present a critical trade-off: while they provide short term and effective suppression that protect human infrastructure, they also pose long term risks—including soil acidification, heavy metal accumulations, and microbiome destabilization—that is likely to reduce ecosystem resilience to future disturbance. With that, it begs the question of “How do long-term fire retardants impact soil dynamics and biogeochemical processes in the Intermountain West?”

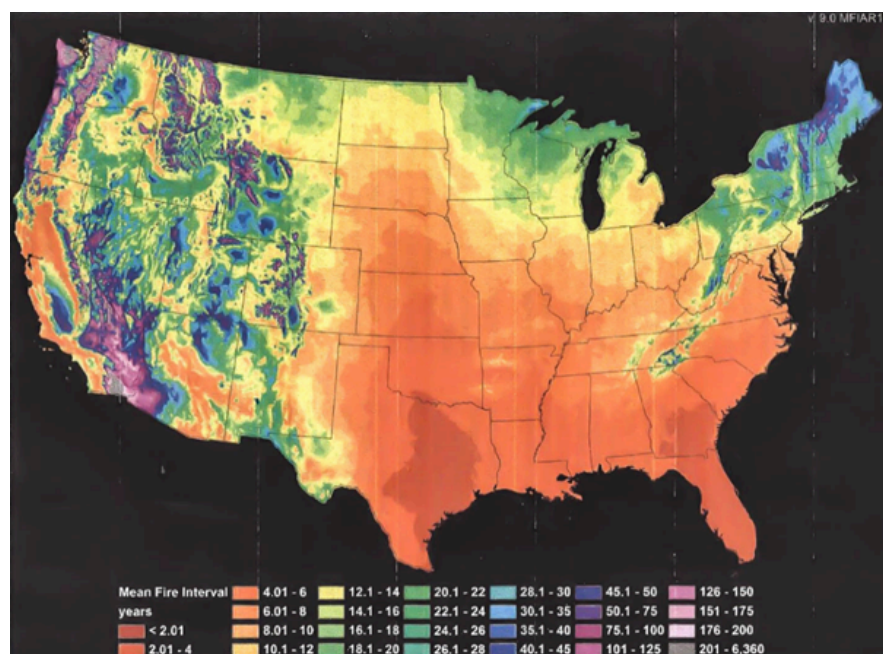


Figure 2: Adapted from *Driftless Prairie*, 2024. A map of the continuous U.S. showing fire return intervals. Note the frequency of FRI's typically seen in the Intermountain West.

Wildfire Suppressants

Answering this question requires examination of these retardants and their role as a wildfire management tool. To understand the tradeoffs associated with wildfire suppressants, it is necessary to examine and understand their context in modern wildfire management. In wildland firefighting, there are 3 classes of suppressants designated by the U.S. Forest Service (USFS). The first being long-term retardants (LTFR), commonly referred to by their brand name “Phos-Chek.” These are the primary suppressants used, as they are formulated to alter the way fire burns by forming a combustion barrier with cellulose tissue when those fuels are heated (Adams & Simmons, 1999). They are comprised of a proprietary blend of fertilizer salts (ammonium polyphosphates), gum thickeners, iron (for coloration), and water. These retardants are typically concentrated and diluted with water to ensure uniform dispersal. Due to this, they do not rely on their water content for effectiveness. They are typically delivered to a site aerially, and their effectiveness is determined by the volume of retardant per unit of surface area. In a study on toxicity of long-term retardants and their toxicity to Fathead Minnows (*Pimephales promelas*), Phos-Chek D75-R’s toxicity and persistence depended on the soil substrate it was applied to. It was found that D75-R remained at a toxic level of 15%-100% after 45 days of weathering, dependent on soil substrate (Little & Calfee, 2005). It should be noted that toxicity may have changed, as Phos-Chek D75-R is an earlier formulation used. While formulation has been slightly altered, current formulations like LC95A and LC95W still retain the same composition of ammonium polyphosphates.

The next class is foam fire suppressants. These are also known as “short term retardants”, as they rely on their water content to be effective. These primarily contain foaming and wetting agents. The foaming agents impact aerial dispersal, how fast water is lost from the foam, and how effectively it can cling to fuels. The wetting agents determine the suppressant’s ability to penetrate fuels.

The third suppressant type is water enhancers. Enhancers alter the physical characteristics of water. They change the effectiveness of aerial drops and their adhesion to fuels. They allow water to cling to non-horizontal and smooth surfaces. Both short term suppressants and water enhancers are typically delivered manually by ground crews, with applications on a much smaller scale compared to aerially dropped retardants (U.S. Forest Service, 2019). Given this increasing dependence on retardant applications, it is essential to thoroughly understand how large-scale applications interact with soil chemistry and nutrient cycling. These suppressants have become effective and necessary strategies in managing wildfires to minimize impacts on human infrastructure and the environment. However, the growing intensity and frequency of wildfire, in conjunction with human expansion into wildland areas, has amplified the scale and frequency at which suppressants are used – particularly in the wildland urban interface, where risks to human health and infrastructure are highest.

Wildland Urban Interface

This increased reliance on suppressants is directly tied to changes in human interactions and activities in fire-prone landscapes. Anthropogenic expansion into land that was once considered wildland has introduced additional conflicts within this new WUI. The WUI is commonly characterized as the space where urban development interfaces with both private and

public areas defined as wildlands (Davis, 1990). Human development into these wildland settings has altered natural fire return intervals (FRI), changing the intervals in which some ecosystems burn, drastically increasing the risk of a high intensity fire event (Fig 3). Many ecosystems across the West that are accustomed to short FRI's, have suddenly seen these intervals become longer, leading to excess fuel loading and delay of nutrient cycling as nutrients are locked in above ground detritus (dead and decaying biomass). Literature surrounding detritus pools and their relation to natural fire is scarce, with virtually no literature regarding the Intermountain West. Limited research has shown that in an Australian eucalypt forest, experimental plots that saw a “no burn” treatment exhibited statistically significant higher Nitrogen and Phosphorus concentrations in litter compared to plots that saw biennial (2y) and quadrennial (4y) burning. This indicates that nutrients were immobilized in detritus and microbial biomass rather than cycling back into the soil (Butler et al. 2020). This pattern of nutrient immobilization in detritus under fire-exclusion strategies creates a baseline of nutrient limitations, making subsequent mass nutrient inputs from retardants more ecologically disruptive. Increased fuel abundance due to changes in FRI's have compounded with anthropogenic induced climate change as the West has seen increased average aridity and decreased average precipitation. Those factors directly contribute to the ease at which fires can start, and the flammability of fuels in the path of the fire (Fig. 3).

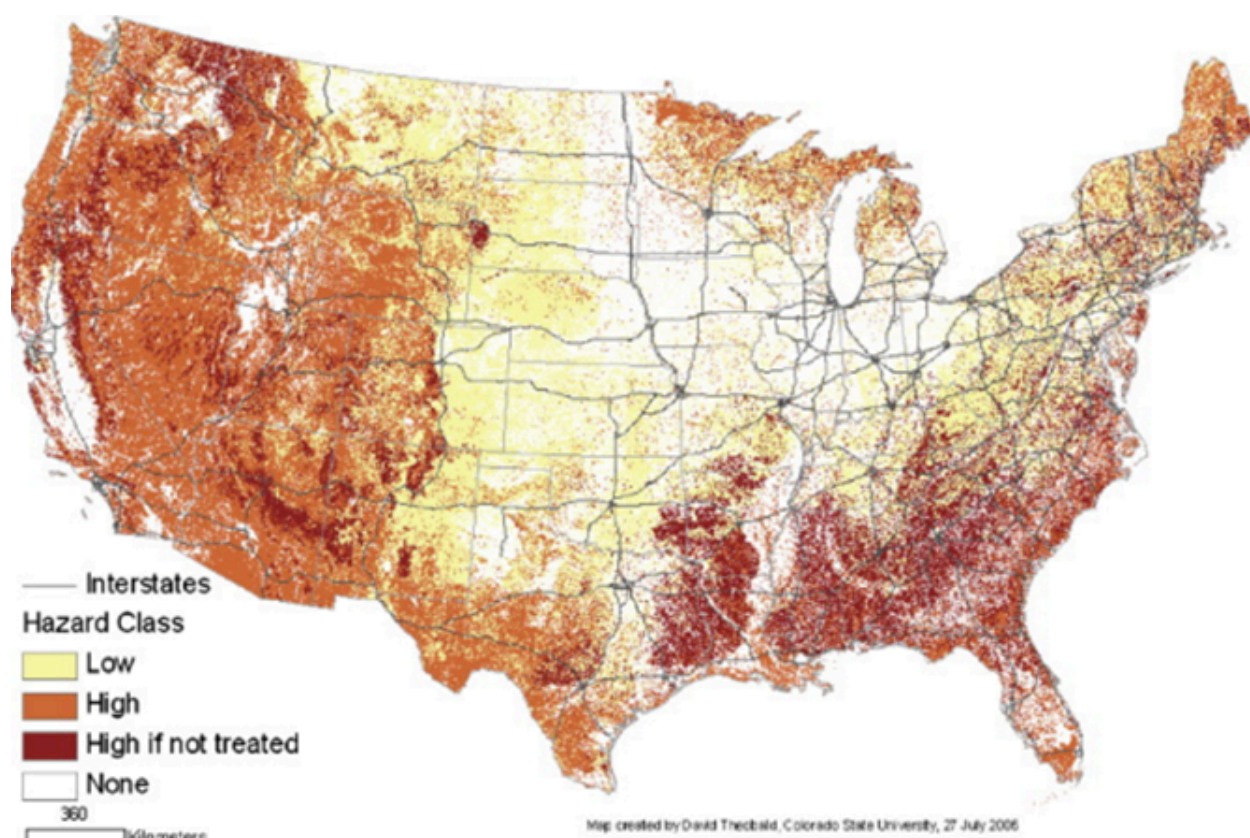


Figure 3: Adapted from Theobald et al. 2007. A map of the United States wildfire hazard classes. Note the relatively high distribution of “High” and “High if not treated” designations across the West.

Increases in Fire Occurrence and Severity

Altered fire regimes have driven not only a need for increased retardant use, but changes in fire behavior itself. With such a drastic change in natural fire regimes in combination with expansion into these wildland habitats, there has been an increased need for structural protection when fighting wildland fires. In the past four decades, areas burned by wildfires have nearly quadrupled. On average, 70 million hectares are burned each year. Additionally, houses in the wildland-urban interface have increased by 350,000 each year (Burke et al. 2021). Wildfire-caused structure loss within the American West saw an increase of 246% of loss per thousand hectares burned between 1999-2009 and 2010-2020. This increase in structure loss was not correlated to an increase in burned area alone, but from human-caused ignition, which accounted for 76% of the structure loss (Higuera et al., 2023). This stark increase in loss was strongly driven by large-scale events affecting communities that lie within the WUI. These increases in large-scale fire events in recent decades have necessitated the increased use of aerially applied ammonium polyphosphate based LTFRs to protect urban structures from wildfire. Between 1987 and 2017, severe wildfires (characterized as a fire that destroys >95% of trees in burn areas) increased by 800% (Parks & Abatzoglou, 2020). Correspondingly, from 2009 to 2021, the USFS and other government agencies dropped 440 million gallons of fire retardant on federal, state, and private land (Tabuchi, 2025; Fig 4). Current fire trends indicate that the West will continue to see greater fire occurrences and severity, prompting the question about how management strategies will follow suit. As long-term retardant use proliferates, it becomes essential to understand how they interact with soil systems, both the positive and negative effects, to effectively utilize this tool in a context of long-term ecological sustainability.

Application Rates

To thoroughly understand the ecological impacts of increased retardant use, it must be established how much Nitrogen (N) and Phosphorous (P) these applications are delivering to the soil surface. The formula of the most common LTFR used, Phos-Chek LC95A, is a proprietary blend and its exact composition is not publicly available. Application rates per m² can be roughly approximated using available compositions. The USFS typically uses a dilution ratio of 5.5:1 of Phos-Chek to water (U.S. Forest Service, 2023). Perimeter Solutions, the producer of Phos-Chek LC95A, state that undiluted Phos-Chek has a composition of “80-100%” ammonium polyphosphates, making the final diluted solution 10-15% ammonium polyphosphates (Perimeter Solutions, 2020). The U.S. Air Force is contracted by the USFS, in which C-130 aircraft may be used as wildfire attack resources. The C-130 (and similarly sized “large” tankers) can discharge up to 3,000 gallons of LTFR across an approximate area of 402.3 meters (1/4 mile) long, and 18.3 meters (60 feet) wide (U.S. Air Force, 2009). Using these numbers and assuming the retardant is discharged uniformly, it’s calculated that ammonium polyphosphates reach the ground at a rate of .18 kg m⁻¹. As Phos-Chek LC95A is a proprietary blend, it’s assumed that the ammonium polyphosphate fertilizers included is similar to typical agricultural blends, being comprised of 11% N and 37% P by weight. It’s calculated that N is applied at a rate of 176.65 lbs acre⁻¹, and P at a rate of 594.19 lbs acre⁻¹ (conversion to lbs acre⁻¹ follows standard agricultural reporting for ease of comparison to agricultural guidelines). In comparison to agricultural systems within Montana, the recommended P fertilizer (P₂O₅) addition to a field of winter wheat

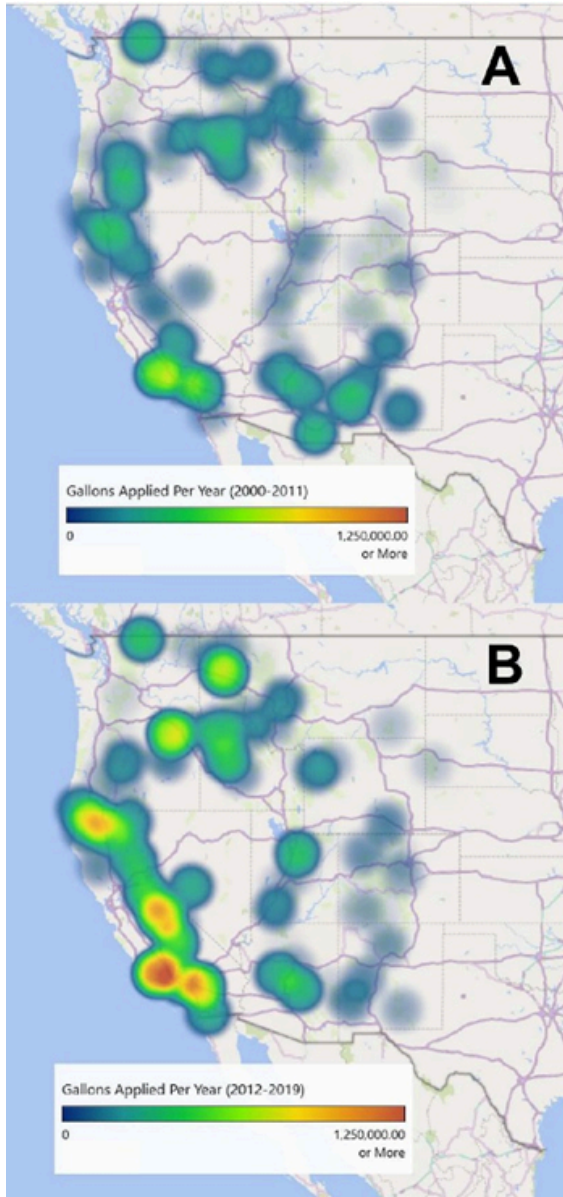


Figure 4: Adapted from Schammel et al.; Applications of long-term retardants across the U.S. West from 2000-2011 (A), and 2012-2019 (B).

with a baseline P concentration of 0 ppm is 55 lbs acre⁻¹ (Dinkins & Jones, 2019). Wildland systems are comprised primarily of grassland, shrub, and forested ecosystems that require roughly 1/2 to 1/3 less available P compared to agricultural crops (Plank & Kissel, n.d.). The P application rates from LTFRs per m² are 10x what would be used in systems that have no available P pool. This represents a nutrient pulse orders of magnitude beyond natural inputs, and even beyond intensive agricultural management. This creates conditions in which plant communities of the Intermountain West have no evolutionary context, creating a higher likelihood plants will experience P toxicity. LTFR dispersal's high P application rates, in

conjunction with its immobile nature, greatly impact microbial communities and the dynamics of these soil systems.

Soil Impacts

Fire retardants induce long-lasting effects on soil chemistry. Ammonium phosphate suppressants introduce large amounts of nitrogen and phosphorus to the soil. Per area, phosphorus inputs from fire retardants are 4 to 44 times greater than farming inputs. While nitrogen inputs are 0.3 to 16 times greater (Moorhead et al. 2025). Specifically, these inputs can cause three-fold increases in labile Nitrogen, 5-fold increases in labile Phosphorus, as well as threefold increases in labile sulphate (Hopmans & Bickford 2003).

These large-scale nutrient pulses rapidly influence the soil system, causing cascading biological and chemical changes. Due to LTFRs being mostly comprised of N and P fertilizer salts, they have the potential to alter soil pH, microbial communities, cation exchange capacity (CEC), immobilization of nutrients in microbial biomass, and shifting competition dynamics among plant species. Microbes are responsible for much of the biological dynamics that occur within soil. LTFR applications introduce large volumes N and P via these fertilizer salts into natural ecosystems. Nitrogen additions via fertilizer salts are linked to soil acidification.

When observing changes to soil pH, the effects of ammonium phosphate retardants vary. One study found an acidification of soil with pH decreasing from 6.5 to 6.0 (Gao & Deluca 2021). In soil with a pH of 5.5, average decreases of 0.3 and 0.2 units occurred. These pH changes were evident 12 months after retardant application. Additionally, ammonium phosphate application immediately increased salinization with a decrease to pre-treatment levels over 12 months (Hopmans & Bickford 2003).

Globally, research indicates that soil pH is responsive to N additions, with an average pH decrease of 0.49. Additionally, temperate forests displayed a greater correlation between pH response and N additions compared to tropical and boreal forests (Tian & Nu, 2015). These findings indicate that precipitation impacts the rate at which N deposition decreases soil pH. While these findings provide a global average, it can be interpreted that similar patterns would be seen within much of the Intermountain West, as the primary ecoregions contained are temperate forests and grassland/scrub landscapes. The Intermountain West in its lower and sub alpine elevations are characterized as “semi-arid,” with higher elevation regions seeing semi-arid characteristics, especially under the influence of climate change. These changes to soil chemistry cause significant alterations to plant and microbial communities.

A study within an Intermountain prairie system on Mt. Jumbo, approximately 1.5 miles NW of Missoula, MT, was conducted 9 years after LTFR application. The study site was determined visually through aerial imagery and field sampling, where pink coloration was present on vegetation and was compared to adjacent control areas. Neither the control nor the LTFR sites were burned, allowing for identification of LTFRs effects without fire. LTFR sites saw an average of 30.6 parts per million (ppm) of available P, compared to 13 ppm in control sites. There was no significant difference in available N between the 2 sites, indicating that it had dissipated via plant uptake, leaching, and microbial activity. This study also conducted the same analysis in sites in Bonner, MT, approximately 4.5 miles W of Missoula, MT. This analysis took place 1 year after LTFR application and saw an available P increase of 6.8x, from 34.8 ppm to

236 ppm. Available N saw similar trends, increasing 5.7x from 3.98 ppm to 22.7 ppm (Marshall, et al., 2016). LTFRs present the risk of heightened nutrient persistence on a decadal scale, having cascading effects. In water limited ecosystems, like the Intermountain West, this long-term persistence is intertwined with soil pH, microbial composition, and ecosystem recovery following a large-scale fire event.

Heavy Metal Deposition

Soil acidification and microbial disruptions caused by nutrient loading represents only a piece of the compounding effects of retardants. Added corrosion-inhibitors, in the form of heavy metals, introduce an additional layer of contamination that interacts with pH changes. Soil pH impacts the mobility of heavy metals in soil solution. Typically, soil acidification increases the mobility of heavy metals through dissolution of the adsorption of Fe, Mn, and Al hydrous oxides sorbed to soil particles. Protonation of soil surfaces leads to decreases in CEC and displacement of heavy metal cations (Cadmium [Cd], Zinc [Zn], and Lead [Pb]) from sorption to clay particles. Limited studies have shown Cd to become relatively mobile under acidic conditions, followed by Zn, and Pb in descending order of mobility. Up to 70% of Cadmium's total concentration can be extracted (meaning it is free in solution) at neutral pHs from 6.5-7.2 due to its tendency to be displaced from sorption by Ca^{2+} and Mg^{2+} (Kicinska et al., 2022). Due to long-term retardants being dispersed aurally, they contain heavy metals that act as anti-corrosion agents to protect the flight equipment that delivers it. One study, focused on Phos-Chek LC-95W, the primary retardant used by the USFS, contains Chromium at $72,700 \mu\text{g L}^{-1}$, Cadmium at $14,400 \mu\text{g L}^{-1}$, and Vanadium at $119,000 \mu\text{g L}^{-1}$ after dilution. These metals sit at 727, 2,880, and 2,380, times the EPA's limit for drinking water standards, respectively. They were compared against EPA standards as there is a high likelihood that they will return to waterways via leaching (Schammel et al., 2024).

Microbial Responses to Fire Retardants

Fire retardants have long lasting impacts on soil properties, leading to downstream effects on microbial communities. Changes in pH fundamentally restructure microbial community composition, and impact overall ecosystem function. Soil pH acts as a proxy for bacterial and eukaryote β -diversity (species richness) in terrestrial systems. Microbial communities tend to shift towards arbuscular mycorrhizal fungal dominance as pH decreases (Marshall et al., 2016). Decreased pH lowered the β -diversity for bacteria, while increasing the β -diversity for eukaryotes. A large mechanism of ecosystem stabilization that is impacted by decoupling of these bacteria-eukaryote dynamics, is that of disrupted nutrient cycling (Duan et al., 2025). This disruption becomes more impactful when considering the changes in competition between bacteria and fungi. The shift towards fungal dominance and change in decomposition rates influence an ecosystem's recovery when these effects are compounded with fire disturbance. The pH-driven effects on soil microbial systems interact with additional components of fire retardants, particularly heavy metals and corrosion inhibitors, which introduce further complexity to the dynamics in retardant-treated areas. Concerning positive feedback loops (a self-reinforcing loop) are created by increased mobility of heavy metals and the addition of metals contained within these retardants, which can be amplified by soil acidification.

Beyond responses to soil pH changes and heavy metal introductions, biomass, metabolic activity, and functional diversity reveal the general effect of fire suppressants on microbes. In the short term, one year after application, ammonium phosphate retardants decrease microbial biomass and respiration (Yu et al. 2021, Barreiro et al. 2010). Additionally, enzymatic activity varies with ammonium phosphate fire retardants. In ammonium phosphate treated fires, β -glucosidase and urease activity can be inhibited while stimulated in others (Barreiro et al. 2010). After ten years, reductions in microbial biomass, respiration, and enzyme activity still persist for the same ammonium phosphate treated soils. When compared with untreated burnt soil, these reductions are most prominent in the top two cm of soil (Barreiro et al. 2016). In contrast, others found an increase in functional diversity and metabolic activity with retardant treatment (Velasco et al. 2009). This study monitored changes monthly over a year with seasonal variations occurring due to soil conditions and moisture. Overall, the diversity and metabolic increases were likely stimulated by nitrogen and phosphorus inputs acting as a fertilizer to these soils. The impacts of ammonium phosphate on microbial biomass, functional diversity, and metabolic capabilities can lead to altered nutrient cycling within treated soils.

In addition to measuring changes in microbial biomass, functional diversity and metabolic activity, we can look at specific changes within the community. Five years after a prescribed fire, the abundances of fungi and gram-negative bacteria in burnt soils increased when compared to unburnt soils. Notably, the effect of ammonium phosphate on these soil communities was not significant as bacterial and fungal abundances were similar to untreated burnt soil (Barreiro et al. 2010). Ten years after a prescribed fire, bacteria activity increased while fungal biomass decreased in ammonium phosphate treated soils when compared to burnt soils (Barreiro et al. 2016). This trend contrasts the fungal dominance that occurs with decreases in pH, but can be explained by nutrient spikes that stimulate bacteria, leading to carbon starvation of the fungi. In the short term, fires can have a positive effect on fungal communities, but the addition of ammonium phosphate can exclude fungi in the long term. Going forward, researchers should consider specific changes within fungal communities. Additionally, researchers should consider the impact soil type has on fungal responses. There are many fungi that act as plant symbionts with 90% of terrestrial plants relying on mycorrhiza (Aerts 2003). Looking at specific changes to certain groups like arbuscular mycorrhiza, ectomycorrhiza, and ericoid mycorrhiza can give insights to plant responses to fire retardants.

Beyond community shifts, soil enzyme activity gives insights to microbial metabolic functioning. When looking at specific enzymes, ammonium phosphate treated soils see a reduction in β -glucosidase activity. This reduction is observed both one and ten years after a prescribed fire, indicating a long-term response. This enzyme is responsible for carbohydrate breakdown into sugars. A decrease in this enzyme activity indicates lower rates of overall respiration, demonstrating less metabolic activity. Additionally, ammonium phosphate treated soils see large spikes in urease enzyme activity one year after a fire. This spike in urease activity could have been caused by the fire-retardant inputs. These suppressants usually contain ammonium salts which are similar in structure to urea. After ten years, urease activity slightly decreased, indicating diminished nitrogen inputs in the long term (Barreiro et al. 2010, Barreiro et al. 2016). These changes to enzyme activities can indicate long term changes in soil nutrient cycling.

Plant Responses to Fire Retardants

The effects of fire suppressants on plant communities are better documented. On an unburned grassland in Montana, ammonium phosphate application resulted in exotic invasion, as nonnative plants take advantage of nutrient spikes more than native plants. Specifically, it resulted in cheatgrass (*Bromus tectorum*) and tumble mustard (*Sisymbrium altissimum*) invasion. Cheatgrass is a facultative mycorrhizal plant and tumble mustard is non-mycorrhizal. These plants pushed out obligate mycorrhizal species (Marshall et al. 2016). Linking to microbial responses, ammonium phosphate application results in increased bacterial activity with decreased fungal biomass in the long term (Barreiro et al 2016, Marshall et al. 2016). Though the invasion was likely caused by nutrient spikes, the impacts on fungal communities should be considered when observing plant responses to flame retardants.

Looking further into plant responses, the nutrient inputs from fire suppressants cause variable growth of different species. Ten years after a prescribed fire, forested plots treated with ammonium phosphate retardants resulted in larger pine heights and trunk diameters than untreated plots, but the root systems were smaller and more trunk deformities were found in treated plots. Additionally, phosphorus accumulations in ammonium phosphate treated trees were more than twice as high as other treatments. For shrubby vegetation, ammonium phosphate treatments favored resprouter species over obligate seeders. The disadvantage to seeder plants was due to the negative effect ammonium phosphate has on germination and seed viability (Fernandez et al. 2015). Moving to unburned wetland systems, retardant concentrations greater than 12% caused phosphorus values to rise above the maximum level instruments could detect. These nutrient inputs cause significant spikes in algal growth, providing evidence for a eutrophication effect from fire suppressants. Specifically, the algae blocked out sunlight and prevented seed germination of Cattail species, leading to decreased species richness (Rennert and Kneitel 2025). The impacts of these retardants on seed viability and plant community structure should raise concern with current application rates.

Glyphosate in Montana Farming Systems

By Anja Bower:

Introduction

Glyphosate is another anthropogenic contamination source that affects soil, water, and biodiversity of ecosystems. Glyphosate is a commonly used herbicide across the world and is especially widespread in Montana agricultural systems. This compound is applied to fields of crops to control weeds, reducing resource competition and preventing unwanted plants from being harvested. While farmers benefit from the removal of weeds, the environmental effects of herbicide application prove to be detrimental. The chemicals in glyphosate enter leaves and stems of plants and surrounding soil, impeding growth of crops that farmers are attempting to harvest. Chemicals also enter surrounding waterways, harming aquatic life and contaminating water. Pollinator exposure to herbicides often results in illness or death, and pollinators are essential to the growth of multiple plants and resulting crop yield. The adverse effects of herbicides threaten farms and can lead to reduced crop yields, harming farmers financially and consumers by reducing food availability (Aslam et al., 2023).

Background on Glyphosate

Glyphosate is the active ingredient in Roundup[®], a popular brand name herbicide registered for use in the United States in 1974 by the Environmental Protection Agency. Roundup[®] was produced under Monsanto and acquired by Bayer in 2018 (Henderson et al., 2010). Glyphosate is one of the most commonly used herbicides in agriculture across the world due to efficiency in removing weeds (Diagboya et al., 2024). Glyphosate is advertised as relatively harmless, as it biodegrades rapidly in the surrounding ecosystem. Despite this, chemicals are still released into the environment during the degradation process. A number of specific fates of this herbicide include absorption, precipitation, and hydrolysis, all of which carry toxins into natural systems. The main product released from glyphosate through degradation is aminomethylphosphonic acid (AMPA), which can inhibit plant growth through accumulation in the soil or application in high concentrations (Aslam et al., 2023). Glyphosate targets the shikimic acid pathway in plants, which is an essential pathway to plant survival. The enzyme 5-enolpyruvylshikimate-3-phosphate (EPSP) synthase is inhibited, resulting in EPSP deficiency. This reduces amino acid production, which plants depend on for protein synthesis and resulting growth. Along with suppressing growth, glyphosate exposure removes green color in plants, causes leaf deformities, and can kill tissue. Plants usually die completely between 4 and 20 days after tissue death. Mammals do not have the shikimic pathway, and glyphosate is therefore not as directly harmful to animals as it is to plants (Henderson et al., 2010).

Glyphosate Effects on Waterways

Glyphosate enters waterways mainly through soil leaching or runoff. Phosphorus is released into the environment when glyphosate degrades, resulting in eutrophication that is detrimental to waterways surrounding farmland. Eutrophication results in algal blooms, which deplete water of dissolved oxygen and ultimately kills fish and other aquatic life. This process alters ecosystem structures, with removal of aquatic organisms changing interactions between primary producer species and consumers (Lozana and Pizarro, 2024). Additionally, degradation of glyphosate in waterways releases chemicals, harming processes such as photosynthesis and respiration that aquatic plants rely on for growth. A literature analysis reviewing health effects of glyphosate on organisms in various study locations found that this compound has adverse effects on processes that microorganisms require for survival. Unicellular organisms such as *Euglenia gracilis*, a type of algae, experienced decreased chlorophyll, photosynthesis, and respiration when a 3×10^{-3} M concentration of glyphosate was applied. *Euglenia gracilis* are a primary producer essential to nutrient cycling and provide a food source for other aquatic organisms. When the population size of this algae declines, aquatic invertebrates that consume *E. gracilis* are faced with fewer food options and could experience malnutrition (Rivas-Garcia et al., 2022). Similar to the food chain disruptions resulting from eutrophication, deaths of primary producers such as *E. gracilis* alter trophic interactions which are key to ecosystem success.

Glyphosate Resistance in Plants

There are 48 grass and broadleaf weed species across the world that have developed glyphosate resistance, with 17 of these found in the United States (Baek et al., 2021). A study conducted in 2016 analyzed the resistance of Russian thistle to glyphosate on Montana farms. Russian thistle is an invasive plant species that competes with native plants for resources,

reducing biodiversity and altering natural ecosystems (Kumar et al., 2017). This study provides an example of the negative effects of glyphosate focused specifically on Montana, and the effect this has at a local scale. Russian thistle significantly reduces the growth of wheat, reducing yields up to 50 percent. Russian thistle is a drought-tolerant plant with early seed production, producing a lot of seeds that can easily and quickly take over an area. The study location was in Choteau County, MT, where farmers follow a wheat summer fallow system. Glyphosate is applied three to four times per year and is heavily relied on by farmers. Thistle takes up a lot of soil moisture, and thereby reduces the benefits of a summer fallow system, which aims to conserve water. Russian thistle was collected from a fallow wheat field, and this specific patch had survived multiple glyphosate applications. Seed dispersal resulting from wind is another method of rapid establishment of Russian thistle in farms, with tumbleweeds frequently forming and dispersing (Kumar et al., 2017). Increasing resistance to herbicides magnifies issues regarding invasive plants taking over dry areas where cereal crop production is abundant.

Effects of Herbicides on Soil Structure

Soil community structure is one of the environmental features most affected by herbicides. Due to soil housing a diverse microbiome with countless ecosystem services, it is also a feature that most greatly affects crop growth and yield. An analysis published in 2023 analyzed the effects of herbicides, fungicides, and insecticides on the abundance and diversity of soil fauna (Beaumelle et al., 2023). This study analyzed 54 different studies focused on the response of soil microorganisms to various groups of pesticides. Diversity was affected even when substances were applied at the rate recommended by manufacturers. The major findings of this analysis determined that herbicides decrease both abundance and diversity of soil microorganism communities. This raises concern at both a local and global scale, with reduction in soil biodiversity creating significant issues for ecosystem health. Microorganisms, which include bacteria, archaea, protozoa, and fungi, break down organic matter and cycle nutrients. The cycled nutrients, including nitrogen and phosphorus, become available to plants and are taken up through root systems (Ortiz and Sansinenea, 2022). Soil organisms compose approximately 25% of biodiversity at a global scale, indicating that their health is critical to the entire environment. (Beaumelle et al., 2023). The depletion of biodiversity results in the need to implement stricter regulations regarding herbicide use and raises questions about potential bans of compositions that are especially toxic.

Pollinator Exposure to Herbicides

Pollinator species experience direct and indirect effects of glyphosate use in crops, often resulting in illness or death of bees. A major indirect effect results from lack of plant diversity, reducing the amount of pollen and nectar available to bees. This also has repercussions for farmers, who rely on bees to pollinate some of their crops and to increase yield. Compounds in glyphosate directly impair the cognitive abilities of bees, disorienting them and making it difficult to find their way back to their hives (Battisti et al., 2021). While many other herbicides contain chemicals significantly more toxic than those found in glyphosate, glyphosate is considered moderately toxic to bees. Direct effects combined with indirect effects of glyphosate exposure are substantial enough to reduce population sizes of bees and create visible effects on crop yield.

Conclusion

Through the analysis of various studies researching the effects of glyphosate in agriculture, the importance of regulating herbicide application is made clear. Even though glyphosate is among the less harmful herbicides, the environmental impacts are great enough to alter ecosystem structure and function. When soil structure is disrupted by chemicals through the decline of microorganism diversity, plants will experience decreased growth from reduced nutrient cycling. The animals that rely on plants for food will need to seek alternative sources. Ecosystem function depends on healthy soil structure to support diverse plant communities and the animals that depend on vegetation for nutrition (Ortiz and Sansinenea, 2022). Other herbicides and pesticides as a whole pose even larger environmental risks, emphasizing the importance of bringing awareness to the issue and introducing alternative, more sustainable farming practices.

Dams and Contaminant Toxicity

By Taylor Hardegger:

Humans manipulate water systems for their advantages. Dams and reservoirs give us power, recreation and a constant supply of water. We often construct large structures to stop water flow and hold it for the many uses we see fit, but through anthropogenic contamination, invisible risks can hide below the surface. The prevalence of dams and reservoirs in Montana, and across the United States, reflects a balance between the desire to restore degraded ecosystems and the need to maintain sustainable sources of energy and water. While these structures play a crucial role in human activities, they also disrupt natural sediment, nutrient, and contaminant cycles, creating complex biogeochemical environments that can heighten the risk of toxic chemical exposure. Over recent decades, dam construction has steadily declined while removal projects have increased, proving a shift in water management strategies to promote ecosystem restoration (Maavara et al. 2020). In the continuation of this process, we must ask, what invisible harms could present themselves in detrimental ways with the removal of dams in Montana?

Contaminants of concern in Montana originate from both natural and anthropogenic sources. Potentially toxic elements (PTEs) in freshwater systems that are strongly influenced by biogeochemical alterations include arsenic and heavy metals. The chemical form, or speciation, of these elements determines their bioavailability by influencing their physical state and molecular structure. Additionally, dams physically retain these toxicants through sediment settling, leading to increased concentration over time. In doing so, dams not only trap contaminants but also transform them into more hazardous structures, elevating the exposure risk for aquatic ecosystems and the communities that depend on these waters (Maavara et al. 2020).

Dams create ecotoxicological risks by transforming upstream contaminants through altered chemical cycling. Water impoundments such as dams and reservoirs introduce unique biotic and abiotic conditions that change the physical state, and therefore the bioavailability, of PTEs. Elements that more readily enter biological systems, such as those capable of crossing cell membranes or the blood-brain barrier, pose greater health risks. This ability is strongly influenced by their chemical form and physical state. Microbial activity and altered redox

dynamics in reservoirs may similarly modify the speciation and bioavailability of PTEs, thereby influencing their toxicity and persistence within aquatic ecosystems. As dams continue to be used for energy production and removed to restore ecosystem connectivity, it is crucial to monitor water quality for the changes invisible to the human eye, an area where current practices and data remain limited.

When it comes to common PTEs, aqueous, clear, and odorless forms are often more toxic than their solid counterparts. Dissolved particles are more readily consumed and metabolized than the corresponding solid (National Research Council, 1977). Dammed waterways also accumulate toxic contaminants in the sediments of still or slow-moving water. While the total concentration of PTEs remains an important concern, this factor is more relevant to upstream sources of contamination as they are directly responsible for the total amount present. Water impoundment primarily influences the bioavailability and speciation of preexisting contaminants, thereby modifying their potential risk to ecosystems and human health.

Milltown Dam Case Study

Montana's long history of damming freshwater systems parallels the expansion of domestic, agricultural, and industrial land and water use. Industrial practices, such as mining, also use and release stream water, adding by-products in the process. Mining practices have contributed to the accumulation of toxic by-products, like arsenic and heavy metals, in downstream environments (Sigler & Bauder, n.d.). In 1981, residents of Milltown, Montana, located just east of Missoula along the Clark Fork River, began experiencing the effects of elevated levels of contamination in the Milltown Reservoir (Figure 5). Initially, the source of contaminants in the Milltown Reservoir was unknown. To investigate the origin of arsenic in the community's drinking water, researchers analyzed water samples for heavy metals alongside arsenic. The co-occurrence of arsenic and various metals supported the hypothesis that upstream mining and smelting activities in Butte and Anaconda were the primary sources of contamination (EPA 2003). Over the following century, roughly 6.6 million cubic yards of contaminated sediment accumulated in the reservoir, derived from upstream mining and smelting activities in Butte and Anaconda (EPA 2021). Subsequent hydrogeologic analyses identified dangerous concentrations of arsenic, lead, and zinc in both the reservoir, aquifer, and nearby groundwater wells (Moore and Woessner 2003).



Figure 5. Clark Fork River Superfund site complex. (EPA 2003a)

The area was designated as a part of the Milltown Reservoir Sediments/Clark Fork River Superfund site location under the Comprehensive Environmental Response, Compensation, and Liability Act (CERCLA) in 1983. This prompted investigations into the sources of contamination and the causes of variable arsenic concentrations among wells (EPA 2021). Researchers found that the potentially toxic elements introduced by upstream activities exhibited increased bioavailability depending on the presence or absence of other reactive elements. Moreover, concentrations of PTEs were significantly higher in the reservoir than in water discharged downstream through controlled flow paths (Moore and Woessner 2003).

Dams and Contaminants

Impounded water susceptible to toxic element formation and accumulation are ones located downstream of PTE producing sources. The construction of a dam in these areas has the potential to either assist in the removal of contaminated sediment from downstream water or increase bioavailability of contaminants through increasing biogeochemical cycling conditions. In mining areas, the use of sediment and tailing ponds are designed to collect and concentrate potentially toxic waste and sediment in a defined area, allowing water free of these compounds to flow down (Sankaran 2025). Hydroelectric dams and recreational and water supply reservoirs are not intended for that purpose. These waters, if unintentionally collecting and producing toxic elements of high bioavailability, present an unforeseen risk to people and other organisms dependent on the water.

Damming alters both the absolute and relative concentrations of elements by changing the residence time of water and sediment. Water impoundment significantly modifies nutrient dynamics within aquatic systems by extending how long materials remain in place. In this

context, *residence time* refers to the duration an element is retained in a body of water, serving as the key factor that balances the rate of transport against the rate of biogeochemical reaction (Maavara et al. 2020). Research shows that stagnant or slow-moving water limits the natural transport and dilution of nutrients while facilitating their transformation between soluble and insoluble forms. Compared to free-flowing streams, reservoirs exhibit much longer residence

Chemical cycling, speciation and toxicity

Arsenic cycling in freshwater environments is strongly influenced by redox conditions. The two primary oxidation states, arsenite [As(III)] and arsenate [As(V)], differ in mobility and toxicity. Under reducing conditions, arsenite predominates and is more toxic due to its higher solubility and greater ability to cross cell membranes. In contrast, arsenate compounds are generally less soluble and less bioavailable, as they primarily exist in a solid forms. Dissolved arsenite is colorless and odorless, with only a faint metallic taste, making it difficult to detect without chemical analysis (National Research Council, 1977).

The environmental mechanisms that drive changes in chemical speciation, and consequently in toxicity, are rooted in natural biogeochemical cycling. Among these, oxygen concentration plays a critical role in shaping redox reactions and microbial activity. Iron, though not itself highly toxic, serves as a key mediator of other elements' mobility and bioavailability. Under oxic conditions, iron exists mainly as Fe(III) oxides and hydroxides, which readily precipitate and act as sorption sites for metalloids such as arsenic. In contrast, under anoxic conditions, these iron compounds dissolve or are microbially reduced to Fe(II) forms, releasing previously bound arsenic into the water column (Borch et al. 2010). Furthermore, certain microbial communities can use arsenate as an electron acceptor under reducing conditions, converting it to the more mobile and toxic arsenite (Dowdle et al. 1996).

Heavy metals act similarly to arsenic in terms of mechanisms and speciation. Bioavailability of metals is affected by redox, microbial activity, pH and presence of other elements. In oxidized conditions, heavy metals exist as metal oxides, hydroxides and oxyhydroxides. In heavily reduced conditions and in the presence of sulfur, insoluble metal sulfides form, collecting in sediment. Slight reduction can result in the dissolution of insoluble metal compounds to their aqueous forms. Heavy metal speciation is also influenced by pH, where a lower pH results in reactions producing the dissolved form (de Paiva Magalhães et al. 2015).

times, allowing water and sediment to remain still and enabling more extensive chemical reactions and microbial metabolism. These transformations are primarily governed by redox conditions, with oxygen availability exerting a dominant influence on both biotic and abiotic processes (Wu et al. 2024). Flowing water continually replenishes oxygen, whereas stagnant, impounded water does not. As a result, oxygen in the water column is gradually consumed to depletion, creating anoxic conditions (Friedl and Wüest 2002).

Conclusion

Dammed streams and reservoirs accompanied by anthropogenic contaminants and microbial communities are breeding grounds for harmful elements at their most toxic forms. Retaining or removing these dams without investigating the biogeochemical cycling of elements in the water permits the potential release of toxic water and exposure to all downstream organisms and people. Potentially toxic element cycling in dammed water resulted in a significant Superfund site in Montana, and the consideration of this cycling led to a conclusively positive restoration. With continued generation of hydroelectricity, the risk of biogeochemical changes of

elements to a more toxic form and accumulation in sediments can become significant. Additionally, as efforts to reestablish freshwater systems proceed with the removal of dams, risk of potentially toxic exposures could increase. When considering the construction or retention of a dam, anthropogenic influences of the water upstream must be taken into account, and the invisible toxicity impacts that could come along with them.

Natural Contamination Sources: Geogenic Contaminants in Montana Groundwater

By Alexander Von Barleowen:

Natural Sources and Distribution: Groundwater Well Contamination

Groundwater is an important source of drinking water in Montana, as a large number of properties that have private wells. While many assume this water to be clean, it is possible for it to be contaminated with inorganic compounds and pathogenic bacteria. Inorganic elements and compounds such as arsenic, and uranium can be dangerous as they are odorless and colorless. A cumulative risk assessment was conducted using 84,000 water quality data points from 6,500 wells across the state of Montana and 19 inorganics were tested for. The inorganics that were tested for include arsenic, uranium, boron, fluorine, manganese, strontium, and zinc. 75% of the 51/81 Montana Watersheds that were tested had a cumulative risk greater than 1, which indicates potential concern. Arsenic and uranium were the two analytes that contributed the most to the cumulative risk assessment. (Eggers et al. 2025).

Well-water quality on reservations is also being looked at. On the Crow Reservation, a study was conducted looking for pathogenic and non-pathogenic bacteria in their well water. Mycobacterium species were detected in 35.1 % of the locations sampled, with 8 in the drinking water fraction. Of the 20 locations that tested positive for Mycobacterium, 8 were treated municipal systems and 12 were groundwater well systems. Legionella species were detected in 21% locations sampled with 5 of those in the biofilm fraction, 8 in the drinking water fraction and only one occurrence of Legionella in both the biofilm and drinking water. Of the 12 samples with Legionella, 8 were treated at municipal sites and 4 were in groundwater. Helicobacter species were detected in 7% of locations sampled, with 2 of those in the biofilm and 2 in the drinking water. There were no occurrences of Helicobacter in the drinking water and biofilm concurrently. All of the positive samples were identified by PCR (Richards et al. 2018).

Arsenic and Fluoride Contamination in River Systems

River systems in close proximity to geothermal areas are prone to water contamination with arsenic and fluoride due to the high temperatures releasing these compounds and making them more soluble (Stauffer et al. 1980). Yellowstone National Park is a prime example of an area where this happens. In geysers around the park, arsenic levels range from 900 to 3560 $\mu\text{g/L}$. The geothermal areas in the park feed arsenic and fluoride into the Firehole and Gibbons Rivers, which in turn flow into the Madison River (Thompson et al. 1979). Average total recoverable arsenic levels from the Madison River starting at West Yellowstone to its convergence at Three Forks ranged from 270 $\mu\text{g/L}$ to 69 $\mu\text{g/L}$. Average total recoverable arsenic levels from the Missouri River starting at Toston to Culbertson ranged from 30 $\mu\text{g/L}$ to 4 $\mu\text{g/L}$ (Nimick et al. 1998). These levels of arsenic are unsafe for human health as the EPA's standard for safe drinking water is 10 $\mu\text{g/L}$ of arsenic (EPA). These high levels of natural contaminants in the Madison and Missouri Rivers call into question the safety of the drinking water for the people living near these river systems.

Human Health Implications

Arsenic is a naturally occurring element that can have detrimental effects on human health. When unsafe levels of arsenic ($<10 \mu\text{g/L}$) are ingested in drinking water, characteristic skin manifestation, vascular disease, renal disease, neurological effects, cardiovascular disease, chronic lung disease, cerebrovascular disease, reproductive effects, and cancer of skin, lungs liver, kidneys, and bladder can all occur (Singh et al. 2007). In children arsenic in drinking water can lead to decreased visual perception, reduced intellectual function, and lower heights (Wasserman et. Al 2004 Siripitayakunkit et al. 2000).

Uranium is also a naturally occurring element and as a radioactive element, there are a variety of negative effects relating to human health. A major effect is kidney toxicity that leads to renal failure and death (Chandrajith et al. 2011). Oral administration of uranium (drinking it in water) can lead to nausea, vomiting, and diarrhea. Paralytic ileus also can develop which causes paralysis of the muscles in the small intestine (Domingo et al. 1987). In regions with high levels of Uranium in drinking water, a correlation between thyroid cancer was found (van Gerwen et al. 2020).

Effects of Pathogenic Bacteria on Humans

Bacteria in drinking water can also have detrimental effects on human health. Mycobacteria sp., Legionella sp., and Helicobacter sp. have all been found in well water around Montana and each have negative effects on humans. Mycobacteria in drinking water is believed to be connected to cervical lymphadenitis in children which causes the lymph nodes in the neck to become inflamed (Primm et al. 2004). Mycobacteria tuberculosis is a famous species of mycobacteria that affects the lungs and leads to death. Non-tuberculosis mycobacteria are increasingly being encountered and identified as human pathogens (van Ingen et al. 2009). Legionella in drinking water also leads to multiple types of disease in humans, Legionnaires disease and Pontiac fever. Legionnaires disease affects the lungs and eventually leads to pneumonia. Helicobacter in drinking water can also lead to disease in humans including esophageal cancer, functional dyspepsia, gastroesophageal reflux disease, asthma, and cardiovascular diseases (Mohebtash, Mahsa 2011).

For arsenic contamination in water, there are a variety of ways it could possibly be removed including ion-exchange, coagulation/flocculation, phytoremediation, oxidation, adsorption, bioremediation, and ultra-filtration. Ion-exchange works by replacing arsenic anions in a liquid with harmless anions on a solid in highly insoluble solutions. This technique works better when dealing with As(V) than with As(III). Coagulation/Flocculation works by coagulating, or sticking together the particles of arsenic in water, followed by filtering those particles out. This method is fairly simple and efficient and can be applied at large and small scales. Phytoremediation method is when plants uptake arsenic from the soil. This method could be used to remove arsenic concentrations from soil, which in turn could help remove groundwater concentrations. The oxidation method works by oxidizing As(III), which is very mobile, into As(V) a less mobile form of the element. Oxidizing agents such as ozone, chlorine, bleaching powder and hydrogen peroxide are used. The adsorption method works by using compounds such as ferric hydroxide and activated alumina to attach arsenic to their surface. Bioremediation works by using microorganisms to remove arsenic from environments.

Ultrafiltration works by using hydrogen peroxide to convert the mobile As(III) into As(V) and then using micellar-enhanced ultrafiltration to remove the arsenic (Dilpazeer et al. 2023).

Uranium has less variety for possible remediation strategies. The most widely used strategy is that of ion-exchange. At pH levels higher than 6 uranium exists in aqueous solutions and the theoretical removal capacity is higher than other common elements found in drinking water such as arsenic and selenium. This strategy has greater than 95% effectiveness at removing uranium. Other strategies include reverse osmosis, permeable reactive barriers using zero valent iron, and adsorption media each with greater than 90% effectiveness at removing uranium (Katsoyiannis et al. 2013).

In relation to pathogenic bacteria, filter materials coated with silver nanoparticles have been tested to determine their effectiveness at removing these organisms from groundwater. These filters were tested on the removal of *E.coli* and at high concentrations, 0.1 mM, there was removal effectiveness ranging from 21% to 100%. At low concentrations, 0.01 mM, there was removal effectiveness ranging from 7% to 50% (Mpenyana-Monyatsi et al. 2012). Filter materials coated with silver nanoparticles are a fairly cost-effective way to remove pathogenic bacteria from water. Another method that was more successful was using biologically active filters. *E.coli* was reduced by 99%, *E. faecalis* was reduced by 99% and *P. aeruginosa* was reduced by 92% (Steven et al. 2022).

Remediation Issues

The issues with remediating arsenic from groundwater include costs associated with certain methods and dealing with different forms of arsenic. Many of these remediation techniques are too expensive for an average household to conduct such as the coagulation method. However, methods such as bioremediation (although somewhat tricky) and ultrafiltration are fairly low in cost. The other problem is converting As(III) into As(V) as the prior is more mobile and harder to remove from the environment (Dilpazeer et al. 2023). This step seems to be necessary in every form of arsenic remediation.

The issue with removing uranium from water is disposing of the waste as there are very specific regulations that need to be taken into account. The type of waste, concentration of uranium, co-occurring contaminants, and state/local regulations need to be considered. To dispose of material with unsafe levels of uranium, sanitary sewer systems or solid drying beds must be used (Katsoyiannis et al. 2013).

The issue with removing pathogenic bacteria from groundwater is that there are not many effective techniques available. Using filter materials coated with silver nanoparticles is a cost-effective technique, however the success rates are highly variable and not reliable (Mpenyana-Monyatsi et al. 2012).

Protecting and Restoring Water Quality

Parasitoids as an Integrated Pest Management Strategy

By Isaac Olson:

In modern agriculture, pesticides are widely used to protect crops from pests, but this use also comes with significant environmental and health drawbacks. While pesticides do a great job of removing pests from a community, they have other implications on human health and the ecological systems that they are applied to. The effects of pesticide use are becoming more apparent all over the Northern Great Plains in soil and water health along with the biota that live within those ecosystems. With evidence linking insecticides to neurological diseases, continuing their widespread use poses unacceptable risks (EPA, 2025). Transitioning toward innovative and sustainable pest management practices offers a safer alternative and a path toward phasing out these hazardous chemicals. Although pesticides are known as one of the most effective ways to manage pests and maximize yield, they may not be the healthiest or safest in regards to our environment. Natural biological controls offer environmentally safer alternatives to harsh insecticides, and using a general or specialized parasitoid to reduce pest abundance may be a promising option. While not every pest has a parasitoid capable of achieving economically significant parasitism, research shows that parasitoids, when paired with other strategies, can support safe and sustainable pest management.

Across the Northern Great Plains, where pesticide effects are becoming more visible, sustainable pest management practices are beginning to take hold. Simple and effective strategies are being implemented in smaller farms across the Great Plains in regard to weed control, like the use of natural predators to reduce invasive species. These strategies have reduced the need for herbicides in smaller agricultural settings and are beginning to be implemented on larger scale farms all over the Northwest. These examples demonstrate that natural biological interactions—predators, pathogens, or parasitoids—can reduce dependence on chemical pesticides when incorporated into integrated pest management.

Unlike parasites that coexist with their hosts, parasitoids complete their development by feeding on and eventually killing the host, making them an effective natural pest control agent. Although parasitoids vary in which host life stage they target, most attack eggs or larvae when the pest is most vulnerable. Many parasitoids are small wasps or flies specialized for particular hosts or host stages. Because they directly kill their hosts and tend to be host-specific, parasitoids can naturally reduce pest populations while maintaining crop health without chemical inputs.

One major cereal crop suffering heavy yield loss from pests in the Northern Great Plains is wheat. Much of this loss comes from pests that feed on the developing seeds or the pith inside the stem. The wheat-stem sawfly is especially damaging and largely unaffected by conventional pesticides because it spends most of its lifecycle inside the wheat stem. This protection makes sawflies uniquely difficult to control and highlights the limits of chemical management in certain pest systems. New innovative strategies include using specialized parasitoids and pest-resistant crop varieties to decrease the sawfly population. Research has also explored biological alternatives such as specialized fungi and nematodes. Two parasitoid wasps—*Bracon cephi* and *Bracon lissogaster*—show parasitism rates high enough to potentially reduce sawfly populations (Portman et al., 2018). Although rearing these parasitoids is challenging because they require sawfly-infested stems for development, research continues to identify feasible release methods.

In practical agricultural settings, these wasps could be reared by identifying small patches of heavily infested wheat, enclosing them in fine mesh, and releasing adult parasitoids inside. This targeted approach offers a realistic pathway for local rearing and field establishment. Research in Montana suggests that sucrose supplementation drastically increases parasitoid

longevity and egg load. *Bracon cephi* lived an average of 10 days in the control and 30 days with sucrose, while *Bracon lissogaster* increased from 6 to 52 days (Reis et al., 2019). Sucrose also increased egg production; for example, *B. cephi* produced four mature eggs after six days compared to one in the control group. Although egg load was measured by dissection rather than lifetime parasitism, field-based tests could confirm how supplemental nutrition influences total reproductive output.

Successful biological control also relies on crop management decisions. The timing and height of harvest influence parasitoid survival because both *Bracon* species overwinter in the lower 10% of the stem (Beres et al., 2025). Cutting stems too low removes the overwintering habitat, reducing parasitoid populations the following year. Planting later-maturing wheat varieties can also increase parasitism because it shortens the time sawflies have to form their protective hibernacula, making them more vulnerable (Reis et al., 2019).

While these biological controls show promise for the wheat-stem sawfly, other pests like the orange wheat blossom midge are still largely managed with insecticides. Chlorpyrifos is commonly used, but its application can overlap with parasitoid emergence and unintentionally harm *Bracon* populations (Beres et al., 2025). Chlorpyrifos also poses major risks to human health by disrupting nerve function (Christensen et al., 2024). This creates a dangerous tradeoff: controlling one pest with chemicals may inadvertently increase another by killing its natural enemies.

Fortunately, the wheat blossom midge has its own natural parasitoid—*Macroglenes penetrans*—which can reduce midge populations by 40–80% without human intervention (Thompson & Reddy, 2016). This demonstrates that suitable biological alternatives already exist for major cereal pests and could significantly reduce reliance on harmful organophosphate insecticides.

Pesticides remain a widely used tool in agriculture, but growing evidence demonstrates that alternatives can maintain yields while protecting ecosystem and human health. Parasitoid-based management will not replace chemical control for all pests, but it represents a critical step toward reducing insecticide dependence and developing more resilient, ecologically grounded pest management systems. As agriculture adapts to new environmental and economic pressures, integrating parasitoids alongside cultural and genetic strategies may help shift production away from hazardous chemicals and toward long-term sustainable solutions.

This shift also intersects with broader environmental concerns, such as chemical contamination of soils and water, including PFAS-related issues. Developing safer biological pest management is one component of reducing chemical inputs across agroecosystems and aligning agriculture with more sustainable, low-toxicity practices.

Evaluating Biochar as a Remediation Strategy for PFAS

By Brenna Matthews-Jackson:

Per- and polyfluoroalkyl substances (PFAS) are quickly emerging as one of the most persistent and concerning environmental contaminants, resisting natural degradation and accumulating in water and soil across Montana. They are synthetic chemicals valued for their

water repellent and grease repellent properties in products such as firefighting foams, nonstick coatings, and packaging. Their chemical stability allows them to accumulate in soils and water across Montana and persist over long periods of time. This includes areas near airports and military sites (Ehsan et al., 2025). PFAS are expensive to remove once released into the environment and pose significant ecological and human health risks. As Montana searches for sustainable and low-cost solutions for water treatment, biochar is rising as a promising solution. Biochar is a carbon-rich product of pyrolysis and could be the key for PFAS remediation in Montana that is both climate-conscious and practical.

PFAS persist due to their strong carbon-fluorine bonds, resisting breakdown by heat, sunlight, and microbes. These compounds have been detected at elevated levels across U.S. freshwater systems as they bioaccumulate in aquatic organisms. PFAS exposure is linked to thyroid disorders, cancer, immune dysfunction, and reproductive problems, even at trace concentrations. There are health advisories set by the U.S. Environmental Protection Agency (EPA) of 4 parts per trillion (ppt) for both PFOA and PFOS in drinking water. In amphibians and fish, PFAS have the ability to impair growth and disrupt entire food webs. For Montana's rural communities that depend on local fisheries and shallow aquifers, PFAS persistence threatens both the safety of drinking water and ecosystem health.

Understanding the sources of PFAS is key for mitigation. In Montana, aqueous film-forming foam (AFFF) is the main contributor, which has historically been used at military bases and airports, such as the Malmstrom Air Force Base near Great Falls. Decades of training has led to PFAS leaching into nearby soils and groundwater, reaching concentrations exceeding hundreds of ppt. Landfills also release PFAS as consumer products like packaging or waterproof textiles degrade. Industrial processes can discharge PFAS-laden wastewater as well. Due to conventional treatment systems not effectively removing PFAS, contamination continues to spread through wastewater and surface water, specifically in regions that don't have centralized infrastructure.

While conventional PFAS treatments like high-temperature incineration, activated-carbon adsorption, ion-exchange resins, and membrane filtration are able to remove or destroy PFAS, they are costly and energy intensive. Incineration requires extreme levels of heat, and filters and resins require frequent replacement and safe disposal afterwards as they produce concentrated streams of waste. For Montana's small, decentralized systems, these technologies are impractical as the state often lacks the infrastructure and/or funding for such advanced treatment methods. Biochar, however, doesn't create mobile streams of waste, but rather immobilizes PFAS within its porous carbon structure, making it a lower-risk and vastly more feasible, cost-effective, option for Montana's rural water systems.

Biochar is produced through pyrolysis, by heating organic materials like crop residues or wood chips in limited oxygen environments. This forms a stable, porous carbon matrix. Its high surface area and reactive sites make for effective binding by PFAS through hydrophobic and electrostatic interactions. Biochar can also be tailored for specific contaminants by adjusting the feedstock and pyrolysis temperature. Studies have shown biochar pyrolysis at higher temperatures adsorbing long-chain PFAS more effectively, while biochar modified with iron or magnesium are able to capture both long- and short-chain forms. (Fabregat-Palau et al., 2022; Liang et al., 2024; Teng et al., 2024). This tunability is what makes biochar very adaptable to the chemical diversity of PFAS and the variable soil and water conditions seen across Montana.

Montana has various agricultural and forestry sectors that each produce an abundance of residues. These residues could serve as feedstock for biochar production which makes biochar serve the dual purpose of also supporting a circular-economy by turning waste into a resource. Decisions regarding whether to divert residues for biochar or to leave them in fields should remain site specific, however, as they hold many soil-health and water-retention benefits. After filtration, biochar that has been regenerated or safely disposed of could even be incorporated back into soils to enhance fertility and water storage. The contaminated material must be handled carefully to avoid any chemical re-release. Biochar's long-term stability also means that maintenance costs are significantly reduced compared to methods like activated carbon or synthetic resins (Behnami et al., 2024).

Despite its promising qualities, biochar is far from the perfect solution. Not only does the adsorption efficiency vary among different PFAS types, but spent biochar also has the potential to re-release contaminants if not properly regenerated or disposed of. Thermal regeneration may also weaken its structure or release adsorbed PFAS, and inconsistent feedstocks can produce even more variability in results. It is essential to do field-scale testing and use standardized guidelines to ensure reliable outcomes before large-scale application will be feasible.

PFAS contamination from firefighting foams, landfills, and industrial waste has left lasting "forever chemical" legacies in Montana's freshwater systems, and conventional treatments are costly for the state's rural infrastructure. Biochar offers a locally sourced, low-costs alternative that aligns with community-based stewardship and environmental goals. Its carbon-rich structure is able to adsorb pollutants while also contributing to carbon sequestration and the state's circular economy. Integrating biochar filtration into future greywater-reuse systems could even further extend Montana's depleting freshwater supply and reduce overall PFAS exposure. Biochar may be the solution and the bridge between pollution cleanup and sustainable water management, especially as water quality and reuse becomes increasingly connected.

Anthropogenic Changes & Demands Down Stream on Water Resources in Response to Climate Change: Graywater

By Benjamin Aupperle:

While biochar can reduce PFAS contamination on a broad scale, local communities such as Bozeman face immediate challenges from increasing water demand and household wastewater containing persistent chemicals. Graywater reuse, which captures water from showers, sinks, and laundry for non-drinking purposes, offers a practical solution that addresses both local water scarcity and ongoing chemical pollution at the household and community level. It reduces freshwater withdrawals, decreases the pollutant load entering treatment systems, and complements larger-scale remediation efforts. Implementing graywater reuse in Bozeman provides a sustainable way to manage water resources as the city continues to grow.

Graywater comes from household uses like showers, sinks, and laundry, other than toilets or highly contaminated sources. Since it has fewer pathogens and solids than blackwater from sewage or drains unsafe due to microbial contamination, it can be treated and reused for things that don't require drinking-quality water (Reclaim Water, 2025). Reusing graywater conserves

freshwater and decreases the volume of wastewater that treatment plants must process. Learning to treat and reuse graywater safely is critical for communities like Bozeman, which face growing pressure on limited water resources. Understanding the characteristics and potential of graywater highlights its relevance for communities like Bozeman, where a growing population and semi-arid conditions are placing increasing pressure on limited freshwater resources.

Bozeman's growing population and semi-arid climate have made water conservation a concern. Rapid population growth is steadily increasing water demand. If current trends continue, the city could eventually need more water than its legal water rights allow. Graywater reuse can reduce reliance on freshwater while providing water for essential non-potable uses such as irrigation, toilet flushing, and some industrial processes. The treatment process typically begins with filtration to remove solids, followed by biological or chemical methods to eliminate organic matter and trace contaminants. Treated graywater can then be safely used for non-potable purposes, helping advance toward a sustainable water cycle focused on capture, treatment, and reuse rather than single-use disposal.

Constructed wetlands provide a low-energy and effective method for treating graywater in Montana's cold, semi-arid climate. These systems mimic natural wetlands by using soil, plants, and microorganisms to remove suspended solids and nutrients. As graywater moves slowly through the wetland, sediments settle, and plants and microbes absorb and transform organic matter and nitrogen compounds (Kim et al., 2009). Even in colder months, microbial activity continues to reduce pollutants. Research indicates that constructed wetlands can remove over 90% of nitrogen and biochemical oxygen demand (Sanni et al., 2025). Small wetland systems could be implemented in community green spaces or residential areas, providing a natural complement to engineered treatment systems. While constructed wetlands offer a natural, low-cost solution, engineered systems provide greater control and consistency in water quality, particularly for urban applications.

Emerging engineered systems build on natural methods by enhancing contaminant removal. Activated carbon filters use highly porous carbon to capture PFAS, soaps, and trace pharmaceuticals (Ashghmoalla & Mehrvar, 2024). This approach differs from biochar, a natural, carbon-rich material derived from biomass that performs well in low-cost or distributed systems. Membrane bioreactors (MBRs) combine biological degradation with fine-pore filtration to efficiently remove suspended solids, pathogens, and nutrients. Electrocoagulation, another promising method, uses electrical currents to clump and separate contaminants without the need for added chemicals. Together, these systems create flexible options for graywater reuse depending on scale, cost, and contaminant type.

A practical example comes from the Shanghai Tower in China, where coagulation-flotation pre-treatment combined with an MBR system treated high-nitrogen graywater from residential and commercial use. The nitrogen levels were high mainly due to detergents and organic matter in the water. The system produced clear, low-ammonia water suitable for irrigation and toilet flushing (Liu et al., 2018). While biological activated carbon filters can handle large water volumes, they require frequent maintenance due to clogging. In Bozeman, decentralized MBR units could be applied on agricultural sites such as livestock ranches or heavily fertilized farms to capture and treat graywater on-site. This approach enables water reuse for irrigation or non-potable purposes, reduces nutrient runoff, and eases pressure on

municipal wastewater systems. This localized approach reduces strain on the municipal system and supports a circular urban water model.

By integrating natural and engineered graywater systems, Bozeman can demonstrate how local solutions address growing water demands and promote sustainability. Thoughtfully designed systems that incorporate wetlands and advanced filtration can reduce freshwater demand, limit wastewater volumes, and strengthen resilience to drought. When graywater is treated and reused rather than discarded after a single use, it creates a closed-loop system that conserves resources while supporting the health of the local watershed. With careful planning, supportive policies, and community education, Bozeman can serve as a model for sustainable, community-scale water reuse in Montana and other semi-arid regions, demonstrating how local solutions can address both environmental challenges and growing water demands.

Transitioning Away from Traditional Agricultural to Regenerative Practices

Regenerative Ag as a Solution

By Bode Kostick:

Since the beginning of the 20th century, industrialization has been the hallmark of modern agriculture. The transition towards industrialization has caused a rise in mechanization, use of genetically uniform monocultures, and consolidation towards fewer larger farms (Center for a Livable Future, 2025). These factors, and a host of others, are what has landed us in our modern industrialized agricultural system. It is how industrial agriculture (IA) farms have been able to produce common goods at lower costs without price premiums (Durham & Mizik, 2021). These practices, as mentioned, can acidify and erode soil and organic matter, leach inorganic fertilizer and pesticides into our waterways, pollute the air through mining of fertilizers and use of fossil fuels, and are detrimental to our soil, air, downstream waterways, and human health (Horrigan et al., 2002).

Organic Agriculture (OA) was one of the first transitions away from IA that was developed in the late 20th century. With the organic certification approved in 1972 (Leu, 2019), it has been around for long enough to display advantages over IA, the biggest advantage being that OA is less environmentally detrimental than IA. It has less severe impacts on water quality, is better for soil health, and consumes less energy (Delate et al., 2015). However, with OA you can still use naturally derived pesticides and “organic fertilizers.” (Durham & Mizik, 2021). Regenerative agriculture (RA) is an improved concept that builds off the foundations of OA. RA prohibits the use of outside fertilizers unless crop nutrient demand dictates it. In dryland areas during seasons with low rainfall, imported N and P are allowed. Only organically approved pesticides can be used and are applied at the lowest efficacious rate; all effort is taken to find alternative controls (Regenerative Organic Alliance, 2023). RA overall advocates for animal welfare, social fairness, and takes a more critical look at soil health (Rodale Institute, 2023). With the threat of climate change increasingly impacting our agricultural systems, it is imperative to transition away from IA. A transition to adopt RA practices into IA and OA is essential to sustain healthy ecosystems and help combat climate change.

Industrial Agriculture

Industrial agriculture relies heavily on external inputs like chemical fertilizers, pesticides, and fossil fuels. It prioritizes short-term yield over long-term ecological structure and function (Aguilar & Paulino, 2025). Industrial agriculture contributes to soil degradation, biodiversity loss in agricultural systems, and substantial greenhouse gas emissions (Aguilar & Paulino, 2025). Due to the output of greenhouse gas emissions into the atmosphere, IA is becoming a threat to the well-being of the globe. IA, while doubling global grain production in the past 60 years, increased the global warming potential (GWP) eightfold. The main contributors to GWP are tillage, synthetic fertilizers, and increased need for irrigation (Abdo et al., 2025). GWP, as defined by the EPA, is the measure of how much energy the emission of 1 ton of gas will absorb over a given period, relative to the emission of 1 ton of CO₂. The higher the GWP, the greater the warming effect of that gas on the Earth (EPA, 2016). Without any mitigation, the GWP of grain production is projected to increase threefold by 2100 (Abdo et al., 2025). This is linked with crop yield globally being projected to decrease in the future. With the most severe climate change scenario and without any mitigation and adaptation, simulated crop yields in the future are projected to decrease 7% to 23%. This is coupled with a projected increase in global total food demand of 30% to 62% by 2050 (Rezaei et al., 2023). The issue is that crop production cannot come from increased conversion of land to new farmland, and farmland cannot be converted to housing developments. The immense threat to biodiversity and the environment is not worth the cost. Increased production needs to come from higher crop yields (Rezaei et al., 2023). Even then, lands with potential for future agricultural expansion have lower productivity than compared with current lands (Abdo et al., 2025). While these factors are all detrimental, the food production from IA is unmatched. The necessary strategy is to increase yield by improving water conservation and soil health, while decreasing fertilizer and pesticide usage. The way to improve these factors is through transitioning to alternative agriculture methods like organic and regenerative.

Organic Agriculture

The term “organic farming” was popularized by J.I. Rodale in the 1940s with the express goal of improving soil health and building up humus through practices that recycle organic matter. The organic movement began in the late 1800s as a response to industrialized agriculture. It arose over the concern of loss of crop quality from diseases and pest attacks, with the introduction of chemical fertilizers (Leu, 2019). This pioneering work by J.I. Rodale eventually led to the formation of the International Federation of Organic Movements (IFOAM).

IFOAM is the international organic governing body formed on November 5, 1972. It is an umbrella movement that seeks to unite and lead the organic sector around the world. They set international standards for policies and definitions on OA (Leu, 2019). The term “Organic Agriculture” as defined by the IFOAM, is a production system that sustains the health of soils, ecosystems, and people without the use of inputs that could have adverse effects. It relies on ecological processes, biodiversity, and biogeochemical cycles that are adapted to local ecosystems. The four principles of OA are the principles of health; ecology; fairness; and care (IFOAM, 2008). All four principles are thought of as equals; one cannot exist without the other.

Critically, OA can either underperform or outperform IA. A comprehensive meta-analysis conducted on the yield potential of OA and IA farming systems shows a lower yield compared to

IA. OA globally has 5% to 34% lower yields than IA. With the best organic practices being used there is an average of 13% decrease in yields compared to IA. However, this depends heavily on each system and its unique characteristics. Under conditions of good management, particular crop types, and growing conditions, OA can almost match IA yields (Seufert et al., 2012). Yields, as well, are not the sole factor to consider in farming practices. They are important for sustainable food security, but OA takes into consideration the benefits towards the social, ecological, and economic systems. The advantage of OA in the face of climate change is worth considering as well. During periods of drought, IA's resiliency is greatly impacted. OA significantly outperforms IA during these periods and has consistently higher yields by 70-90%. This is attributed to the soil's higher water-holding capacity (Lotter et al., 2003) and water-infiltration rate (Delate et al., 2015). OA advocates for a more holistic, ecologically beneficial system of agriculture that can sustain crop production during prolonged drought due to climate change.

Regenerative Agriculture

As defined simply by Regeneration International, "the opposite of regenerative is degenerative" (Regeneration International, 2025). IA is degenerative while regenerative agriculture (RA) is a holistic systems approach that encourages continual innovation for environmental, social, economic, and spiritual well-being. It is characterized by improving the land that is used, rather than degrading it. The primary aim of RA is to increase soil organic matter (SOM) (Leu, 2019). SOM is the substrate and habitat of soil microorganisms and fauna (Schnitzer & Monreal, 2011). By improving the SOM content in the soil, it is regenerated and revitalized and in turn, so is the environment. RA treats soil as a non-renewable resource. Its loss and degradation is finite and is non-recoverable within a human lifespan (FAO, 2015). With improved soil health and greater focus on conservation, RA can make farms more resistant. It can increase resilience to extreme weather events, contribute to higher amounts of water retained in soil, increase the abundance of beneficial soil biota, and increase nutrient bioavailability (Leu, 2019). RA is an essential tool for combating the effects of climate change. Carbon emissions alone have had a fourfold increase in the past 60 years (Abdo et al., 2025). However, soils from RA can be considered healthier than IA, meaning there is more SOM, greater levels of microbial activity, and greater potential to sequester carbon in the soil. According to some sources, RA has the potential to mitigate climate change (Rodale Institute, 2014, pg.2). RA could be a step back in time to return to our natural agricultural roots. RA could be hugely beneficial to current agricultural operations with its potential to sequester carbon and improve SOM.

Farming Systems Trial

The Farming Systems Trial (FST) by the Rodale Institute (RI), started in 1981 and is the longest-running study in the U.S. comparing industrial, organic, and regenerative organic agriculture. The RI is an organization founded in 1947 by J.I. Rodale. It seeks to advance the regenerative organic agriculture movement through rigorous scientific experimentation, farmer training, and education (Rodale Institute, 2025). The FST is characterized by its longevity, simplicity, and display of OA and RA response to drought conditions. The FST is arguably one of the most effective studies that has analyzed these farming systems, and is located in Kutztown,

Pennsylvania. The soil type at the farm is a moderately well-drained Comly silt loam (Delate et al., 2015).

In the FST the IA plots are a conventional synthetic system representing a typical U.S. grain farm. OA plots are an organic legume cover crop system representing a traditional organic cash grain system. RA plots are an organic manure system that uses those same leguminous cover crops and periodic applications of composted manure from livestock (Rodale Institute, 2025). The crops chosen were based on typical crops grown in Pennsylvania. The conventional system grew corn and soybeans for 23 years, then wheat was added in 2003. Both organic systems grow corn, soybeans, wheat, red clover, alfalfa, and hay. In 2008, genetically modified crops and glyphosate no-till treatments were added to the conventional plots (Delate et al., 2015).

The FST analyses factors of soil health measured using the Cornell comprehensive assessment of soil health (CASH) (Rodale Institute, 2025). Unique CASH assessments are made to represent different regional areas and their different needs. Overall, it measures 30 physical indicators, and more than 10 biological, chemical, and crop observation-based indicators of soil health. The physical factors range from 'soil feel,' crusting, water infiltration, retention or drainage, and compaction. Soil biological properties encompass soil smell, color and molting, earthworm, or overall biological activity. Crop indicators include root proliferation and health, signs of compaction, legume nodulation, and signs of residue decomposition (Moebius-Clune et al., 2016).

The results over the course of the 44-year study show the promise of organic and organic regenerative agriculture. The FST has been able to analyze a period known as the "transition effect" and was one of the first long-term trials to report on this effect. The transition effect is the transition from conventional to organic agriculture, a period that lasts 36 months. After an initial yield decline during the transition years, organic grain yields eventually matched the conventional grain yields (Delate et al., 2015).

Since 2008 with the addition of genetically modified crops and glyphosate to the system, conventional yields have not improved over organic yields. Organic yields have also shown greater resiliency during drought periods. During a drought year in 2016, organic corn yielded 8,411 kg ha⁻¹ and conventional corn yielded 6,403 kg ha⁻¹ (Delate et al., 2015).

The FST was one of the first studies to monitor underground water quality using a lysimeter. It helps to assess the NO₃-N leachate out of the system. In conventional systems, the leachate exceeds the drinking water standard of 10 ppm, while the organic systems did not. In an analysis of the energy used in the manufacturing, transportation, and application of fertilizers identified that the organic systems used 45% less energy than the conventional system. The N fertilizer alone accounted for 41% of the total energy consumption. Similarly, GHG emissions associated with the conventional system were 40% greater per volume of production than the organic systems (Delate et al., 2015).

Organic systems can be economically competitive with conventional agriculture. Only a 10% premium is needed to achieve the same economic viability as the conventional system. When actual organic price premiums were compared, the organic systems achieved 2.9 to 3.8 higher returns than the conventional system.

The FST demonstrates that organic and regenerative organic can match conventional yields, resist drought conditions, leach less nitrate, produce less fertilizer related energy, and be economically viable.

Discussion

Without a doubt, IA food production is unmatched. IA has been able to double grain production in the past 60 years. Most farmers in the U.S. practice a form of this agriculture. It is usually consistent, predictable, and has worked for decades. However, climate change is challenging the viability of IA. No longer can the production be as consistent year after year. Once predictable patterns are becoming less certain, we will continue to see this unpredictability.

The most pertinent solution must be a transition away from these conventional practices. It is necessary to preserve our finite water and soil and protect our air. The way to do this is to incorporate organic and regenerative practices into our current industrial practices. We need to help farmers understand that our soil is not simply a growing medium, but a living breathing dynamic resource that must be conserved. Farms must be considered ecosystems that can nourish the soil, animals, and microbes, if you treat them right. The benefits of this resource cannot be mitigated by overuse of pesticides and fertilizers. In treating plants and animals more naturally, the benefits toward human life can be numerous.

There are two possible solutions to achieve this. One solution must involve a cooperative government that provides subsidies to RA farms that can have competitive pricing equivalent to OA, and IA. This can make it so people do not have to choose between their health and the health of the environment; they can support regenerative practices without weighing on their wallet. However, this relies on large-scale farms and requires small incremental change to transition all of IA agriculture. The other solution is small. Small farms that have no mechanization and yield maximization. They create a better connection between the consumer and the product. We must shrink our farms to small backyard plots with backyard chickens. Gardens would provide healthy produce, that could subsidize purchasing those at the store. The chickens could eat what you cannot. The cost could be governmentally subsidized, and extension specialists could provide the necessary education. This solution could take the strain off our global food supply chain and help mitigate emissions. It could create a greater awareness of our food system and could help us be more mindful of how our food ends up on our plate. Both solutions I believe are viable; it is up to us to choose which will benefit us and the planet the most.

Unpredictable Precipitation & Adoption of Sustainable Agriculture Practices in North Dakota

By Alyssa Harmel:

Challenges of Changes in Climate

Rises in global carbon dioxide levels are contributing to increases in temperature and precipitation variability in the Northern Great Plains (NGP) region. Precipitation variability includes unpredictability of both frequency and intensity of precipitation events. Locally, this will impact the agricultural integrity, or long-term sustainability of agriculture practices, of the NGP region. The NGP has seen an average temperature increase of 1.7 °F over the past few

decades, greater than that of any other region in the U.S., as well as an increase in severe heat events. In addition, precipitation patterns in the NGP have shifted in favor of increasing precipitation amounts in spring and fall and decreasing amounts in the winter months. Extreme precipitation event frequency has also increased, resulting in higher amounts of precipitation in shorter periods of time, which does not always favor an increase in effective precipitation (Cross et al., 2021). These irregularities in precipitation timing in the region have a large influence on agricultural productivity, making agriculture in the region environmentally and economically vulnerable.

For this review, I will consider precipitation variability in North Dakota (ND). ND receives greater precipitation amounts than its western neighbor, Montana, but still experiences moisture differences across the state. Precipitation in North Dakota is expected to continue to increase during colder spring and fall months, aligning with predictions for the NGP region. This has potential to increase soil moisture but would delay the crop planting season and potentially impact harvest timing. This predicted precipitation increase during colder months is also thought to result in greater drought intensities for the state, because rising temperature rates could influence an increase in evaporation of colder-season precipitation (Frankson et al., 2022). In response, North Dakota crop producers have begun to utilize mitigative approaches to maintain yield and minimize potential soil health impacts in a time of irregular precipitation and greater drought risk.

Based on these considerations, the question driving this review is: How do unpredictable precipitation patterns affect the adoption of sustainable agriculture practices in North Dakota? The goal of this review is to assess North Dakota producer insight of strategies for maximizing water usage in a time of unpredictable precipitation variability.

North Dakota Agriculture

To better understand production pressures that are associated with increasing precipitation irregularity patterns in North Dakota, I consulted Kipp Harmel, a crop producer from Rugby, ND. Harmel gave producer insight for the 2025 growing and harvest season in north-central North Dakota, as well as for observations of water use efficiency strategies applied in the area. His observations of the 2025 season and the strategies that have come along with the season's variability serve as recognition for each producer's crop rather than an average, as well as the hard work put into each yield.

When asked, "What did this past growing/harvest season look like in central ND in relation to weather impacts?" Harmel had observational insight that aligned with current climate predictions of precipitation variability for the state. As previously mentioned, precipitation drives not only yield but also crop production timing. Over this past crop production season, April-September, central North Dakota producers experienced a later planting season and harvest season because of irregular precipitation event timing, as well as immense weed pressure. Many producers were not able to plant wheat until late spring for the 2025 crop season due to high precipitation frequency during the normal planting timeframe. Once the crop emerged, later than usual, many producers integrated pesticide application. Subsequent to application, rain fell yet again, resulting in pesticide removal and weed pressure. Heat and dryness increased throughout the summer, resulting in reduced crop productivity from drought impact and even greater weed pressure. Many producers applied pesticides again for the sake of their yield around the "normal"

harvest window of August, but were then hit with additional, unexpected rainfall. This final rainfall not only washed off the applied pesticide a second time, but postponed the harvest season. At the time of the interview, September 28, 2025, many producers in central ND were still harvesting their wheat crop from the season (K. Harmel, personal communication). Crop production timing relies on precipitation patterns, and today's increasing precipitation variability is making production economically and environmentally vulnerable.

Challenges Related to Soil Health

Precipitation irregularity not only impacts crop production timing but also has substantial effects on soil health. In agriculture, minimizing soil disturbance and maximizing surface cover is the most effective way to maintain, or even enhance, the health of a landowner's soil (USDA, n.d.). By minimizing disturbance and maximizing coverage, soil-water infiltration capacity is strengthened, and soil health is maintained. This is integral to crop production as precipitation patterns become more irregular and unpredictable, because soil health can be the determining factor of drought resilience.

Accounting for crop residues in agriculture is one of the most proficient ways to minimize soil disturbance and maximize soil surface coverage. Crop residue, the remaining biomass of a crop after harvest that includes stubble and chaff, is a large factor in soil surface coverage when less disruptive agricultural practices are implemented. Not only are crop residues an input of organic carbon and nutrients into the soil, but the physical residue is essential to water retention (Ghimire et al., 2017). Stubble biomass decreases soil compaction, which increases surface infiltration ability and precipitation capture. Soil aggregation, or structure, is a product of crop root systems, which can also be crucial to the infiltration ability of a soil. Residues also help to protect the soil surface from high wind speeds by creating a small-scale windbreak (Nielsen et al., 2005). Crop residues are important for decreasing soil erosion from both wind and water, and maintaining the overall health of a producer's soil in times of unpredictable precipitation.

Making Strategies Worthwhile

No-till, stripper header utilization, and cover cropping are current agricultural strategies that support soil health, and in turn may help to mitigate precipitation variability. No-till, the absence of tillage before planting, is already commonly adopted by many producers for allowing residues to keep soil intact and increase soil water storage (Nielsen et al., 2005). Even though no-till is commonly adopted, some ND producers prefer to do a rotational tillage method. This includes tillage once every 3-4 years, or longer, to break up possible compaction that still occurs with crop residue in times of drought, which ND is prone to. No-till has been an easily adopted strategy because asking producers to abandon a practice, and in turn "lighten their load," has proven easier than implementing other mitigation strategies.

Stripper header utilization has also gained traction from producers as an efficient water-use strategy in the face of unpredictable precipitation. A stripper header, compared to a conventional header, leaves higher stubble that offers more soil stability and water infiltration capacity while maintaining crop residue biomass. Harmel approached stripper header technique advancements with optimism but relayed that crop rotation must be considered prior to utilization. In the early growing season, Harmel's neighbor's soybean crop that was planted in

high stubble left behind by a stripper header expressed growth deformities. The shade from high stubble at the time of emergence reduced growth, as high amounts of sunlight are essential to soybean growth. This resulted in a diminished yield for the producer at harvest time, which can make or break profitability (K. Harmel, personal communication). Even though tall residues may be effective for increasing water retention, maintained or increased yield productivity may not be directly linked to the technique without intentional crop rotation.

Cover crop implementation is not new in North Dakota agriculture, as existing federal and state soil health programs help to incentivize ND producers to plant cover crops with grant funds. Cover cropping is the practice of planting a crop to maintain soil surface coverage. In North Dakota, this often includes planting the cover crop in late fall after harvest of a producer's cash crop, or early spring before the planting of the cash crop. Common cover crops include rye, winter wheat, oats, clover, and radishes. Each crop is chosen for specific benefits, and sometimes for grazing, but all help to maximize soil surface cover and reduce erosion potential (USDA, n.d.). Many producers still feel deterred from utilizing cover crops because of the cost, time, and effort that goes into planting and harvesting another crop at a different time of the year. At times, the program money is not always incentive enough for producers to work overtime. It can even be "economically risky" in the short-term for producers to adopt the practices of both planting and harvesting, as the payoff is linked to future improvement in yield from soil health improvements of increased water retention, rather than to an immediate profit (Kelly et al., 2021). Increases in cover crop implementation in North Dakota will be more likely when implemented alongside grazing practices. This would allow producers to cut the cost of harvesting the crop but still earn a return in the form of soil aggregation and infiltration ability of their field. In turn, future crop production may be more climate resistant because of increased soil health. With grazing of cover crops, soil aggregation and water capture as a soil health parameter are maintained (Kelly et al., 2021). This suggests that the grazing practice itself does not pose a risk to the water retention of the soil when concurrent with cover cropping. Implementing grazing practices alongside cover cropping may not be easy for producers without livestock, but a producer could lease the cover crop for grazing rather than starting a livestock operation. This would allow them to not only profit in the short-term from the lease income, but also in the long-term in regard to the health of their soil. If a producer already has a livestock operation, they will be able to cut the cost of buying feed if utilizing grazing with cover cropping, while also reaping soil health and water retention benefits.

Conclusion

As precipitation becomes increasingly more irregular and unpredictable with rising carbon dioxide levels, soil health and crop production timing are being impacted. To minimize these potential pressures of precipitation variability on agriculture, North Dakota producers have begun to utilize and develop water efficient strategies for maintaining yield. Assessment of producer insight of no-till, stripper header utilization, and cover cropping as mitigation strategies are integral to address the impact that variable precipitation is having not only on production, but also on our producers. This returns to the question, "How do unpredictable precipitation patterns affect the adoption of sustainable agriculture practices in North Dakota?" Producer insight concluded that eliminating a practice, in this case tillage, proves easier than integrating practices as mitigation strategies because of the workload and costs associated. The integration of water use efficiency strategies is intensive, especially cover cropping, but the long-term benefits are

increased soil health and production resiliency leading to an overall greater agricultural integrity. Increases in precipitation variability are leading to agricultural vulnerability, and producers need support to make mitigation strategies environmentally and economically worthwhile. Reliable precipitation is an essential component of agriculture and helping landowners navigate the challenges of increasing precipitation irregularity should be our priority.

Cropping Systems and Cover Crops

By Cole Edwards:

Continuous Cropping

Adapting cropping systems to changing climate conditions is a way to improve efficiency and maintain long-term sustainability. Fallow has historically been used to conserve water in semi-arid and arid regions. However, nitrate leaching, decreased microbial biomass, and degraded soil physical properties are all issues stemming from fallow (Ruis et al., 2023). Continuous cropping with cash crops is one way to avoid some of the negative effects associated with fallow. Continuous cropping increases microbial biomass in the soil, compared to fallow (Drijber et al., 2000). However, microbial communities are less diverse in continuous monocropping systems compared to diverse cropping systems. Reduced microbial diversity can lead to increased susceptibility to crop diseases (Pervaiz et al., 2020). Continuous cropping does increase SOM, and reduce erosion, compared to fallow (Lenssen et al., 2007). Along with those benefits, major drawbacks do exist when planting continuous cash crops, especially continuous wheat. Continuous cropping systems often use more water than diversified cover cropping systems that have early-terminated cover crops. Farms in the northeast corner of Montana are typically more successful in continuous cropping systems, as average annual precipitation approaches 14 inches. Although the Lewistown area experiences relatively high precipitation, shallow soils, which are common in the area, have less water holding capacity (USDA NRCS, 2008; Figure 6).

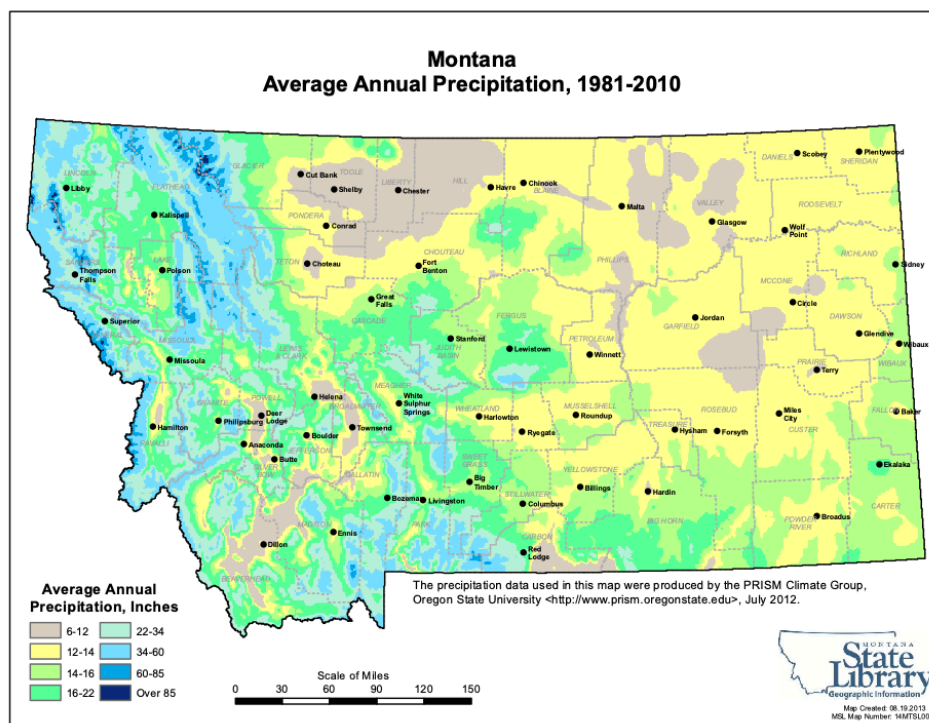


Figure 6: Average annual precipitation in Montana, 1981–2010. Data from the PRISM Climate Group, Oregon State University; map created by the Montana State Library (2013).

Cover Cropping Options

According to a 2015 survey conducted by Montana State, Montana producers who have grown cover crops in cereal rotation most commonly used pea, turnip, radish, and lentil (Jones et al., 2015). Leguminous cover crops, like peas and lentils, fix N with the help of rhizobia in root nodules. Because of the symbiotic N fixing relationship, legumes also tend to have higher N content in aboveground plant biomass compared to other plants such as wheat. Within N fixing plants, a majority of the N persists in the aboveground biomass (Liu et al., 2024). Leaving crop residues in the field can be helpful to both increase soil N and C, as well as retain soil moisture, especially during fallow periods or when living crops are in early stages of growth (Simon et al., 2022).

Field pea and lentils have relatively shallow rooting depths, possibly contributing to less water use. Wheat, sunflower, and safflower, on the other hand have extensive root systems which can access water across a wide range of soil depths (McVay, 2022). In addition to root depth, the timing of cover crop termination is also an important factor in water use and subsequent crop yield, which will be explored in a later section.

Other, less popular cover crops include canola, sunflower, safflower, camelina, sorghum, and millet. These crops, although not N fixers, provide other benefits to soil health such as increasing SOC, and assisting in water infiltration. In a study focusing on sorghum and camelina in rotation with wheat, the two cover crops increase soil aggregate stability as a result of

increased SOC, and improve soil microbial activity (Obeng et al., 2024). The deep root systems of safflower and sunflower can help to break up compacted soil, and improve deep infiltration of water after cover crop termination. Sunflower and safflower may also be effective at accessing nitrate that has leached deeper into the soil profile. It is important to note that because of their deep rooting nature, sunflower and safflower tend to use more water than some other cover cropping options (Miller & Holmes, 2012).

Unlike the cool season cover crops already mentioned, sorghum and millet are C4 plants, which makes them better adapted to warm growing conditions and tolerant to some drought conditions (Chaturvedi et al., 2023). These two cover crops not only provide benefits to soil health, but also serve as viable, late summer forage options for livestock, when many other forage options are dried out. Although sorghum and millet can grow in dry regions, they still use considerable amounts of water, which can affect subsequent cash crops. This tradeoff must be acknowledged by farmers to see if the benefits from growing a cover crop for forage can outweigh possible losses in wheat yield.

Timing and Management of Cover Crops

The timing of cover crop termination is an important variable to consider, especially in semi-arid environments like much of Montana. Later termination of cover crops can contribute more organic carbon into the soil, however, subsequent wheat yields may suffer as water availability is decreased. Early cover crop termination can allow for some carbon addition to the soil, while still maintaining adequate soil moisture (Miller et al., 2023). Farmers should consider current and predicted precipitation when determining when to terminate cover crops. In dry years, or in dry areas such as north central Montana, farmers will want to consider terminating cover crops early to conserve soil moisture for the following cash crop, especially in soils with lower water holding capacity. In contrast, farmers in regions with higher moisture may be able to terminate cover crops later, achieving increased benefits of organic carbon additions or nitrogen additions from legumes; however, N release from decomposing plants is also affected by termination timing. Late termination can add plant matter with high C:N ratios, causing plant decomposition to take longer (Pesini et al., 2023). The decision of when to terminate cover crops is highly site and season-specific and should consider soil conditions, weather forecasting, and cropping goals.

Innovating Agricultural Sustainability with Compost

By Eve Heeley-Ray:

Sustainable farming recognizes the need to maintain nutritious, bioactive, and carbon-rich soils that can produce crops without eroding away. Farmers are faced with changing up their practices, in some cases reverting to practices that have been around since the dawn of

agriculture. One such practice is called “composting”, and it has been a natural source of nutrients and soil health as far back as the Stone Age (Social Farms & Gardens, 2021).

“Compost” refers to a process of recycling organic waste into a material that can improve soil quality and provide nutrients for plants (U.S. Department of Agriculture, n.d.). Individual people and families may compost their yard-waste and food scraps to fertilize their gardens. Farms may use food scraps, unwanted crop parts, animal manure and other sources of organic material to create fertilizer for their crops. Keeping soil rich in organic matter keeps it healthy. The USDA Natural Resource Conservation Service (NRCS) defines soil health as “the continued capacity of soil to function as a vital living ecosystem that sustains plants, animals, and humans,” (NRCS, 2025). Compost is used as a soil amendment because it adds organic material back into the soil which improves its structure, loads it with plant nutrients, increases water-holding capacity, and nurtures micro- and macrofauna that live beneath the soil surface. Using compost to grow crops, with its high nutrient value and its water-holding capacity, can greatly decrease the amount of nutrient leaching caused by synthetic fertilizers. Making compost a component of farms big and small can aid in the cultivation of healthy soils by increasing carbon and water storage, cultivating a diverse and active macro and microbiome, and decreasing the need for mineral fertilizers which acidify the soil and pollute Earth’s waters.

Compost improves soil structure by increasing soil aggregation and biological activity, which both increase the pore space within the soil structure (U.S. Environmental Protection Agency, 2025). Pore space is important for soil because it creates a less dense soil, allowing for greater water infiltration, percolation, aeration, nutrient mobility, and flora and fauna growth (Nimmo, 2013). Increased organic matter also makes the soil more resistant to erosion from wind due to increased aggregability as well as water due to less vulnerability to runoff (Jarvis et al., 2024). On particularly sandy soils, adding organic matter can hold the sand particles together so they are able to hold onto water that would otherwise leach through (Old Farmer’s Almanac, n.d.). Organic matter added to hard clay soils attaches to the fine, densely packed clay particles and creates more space between them so they can hold and drain water that would otherwise not be able to penetrate. On perfectly loamy soils, the type well-suited for growing crops, keeping organic matter robust and alive with care, through things like compost additions, keeps the soil from losing water and nutrients and eroding away.

Compost improves soil chemistry and nutrient availability by stabilizing the pH (Zhao et al. 2022), improving cation exchange capacity (Medina et al., 2025), and adding nutrients back into the soil (Kwiatkowska-Malina & Maciejewska, 2021). These nutrients are also slowly released, which is preferable to synthetic fertilizers which give plants a boost of nutrients which quickly dwindle (Ellis, 2020). The essential macro-, secondary, and micronutrients plants take from the soil are returned to the soil when you compost them. Essential macronutrients plants need to grow are carbon, hydrogen, oxygen, nitrogen, phosphorous, and potassium. Secondary nutrients which plants need in smaller amounts are calcium, magnesium, and sulfur. Micronutrients (or “trace nutrients”) are only required in tiny amounts and include boron, chlorine, copper, iron, manganese, molybdenum, and zinc (Provin & McFarland, n.d.). Although needed in different quantities, all nutrients listed are vital to plant growth. Aside from carbon which is taken up by plants through the air during photosynthesis, most other nutrients are taken from the soil by roots (Provin & McFarland, n.d.). Oxygen and hydrogen are consumed by plants both from the air through stomata and by roots in the soil. Compost, because it contains organic

matter which could only grow because it took up all essential nutrients needed for plant growth, contains the nutrients essential for new plants to grow. However, in the raw form nutrients are in when one first throws them into the compost pile, the nutrients are not yet available to plants. The organic matter must be broken down enough so that the nutrients exist in their elemental or ionic forms (Provin & McFarland, n.d.). Macro-organisms (like mites, slugs, or worms), microorganisms, and fungi do this for plants. Organic matter is consumed by smaller and increasingly smaller organisms until bacteria and fungi release it as elements and ions for plants to get their turn (Paul, 2015). Enzymes secreted into the soil by microorganisms are also important for breaking down organic matter (Daunoras, 2024). Some fungi, called mycorrhizae, have symbiotic relationships with plants where they create inter-connected networks with plant roots which allow them wider access to nutrients (Figueiredo et al., 2021). This entire community of organisms is essential to making nutrients available from organic matter: macro-organisms, fungi, and microorganisms like bacteria. Adding organic matter to the soil in the form of compost has a positive feedback loop effect where the biota of the soil is fed and can thrive, increasing plants nourishment, and increasing the organic matter produced by the thriving soil. A robust microbiome in a soil yields healthy soils and plants which are stronger and more impervious to pests and pathogens (De Corato, 2020).

Compost piles can be used to generate soil-amendment ready material on both the small scale and the large scale. One could create a compost pile with household food scraps in the backyard to feed their home veggie or herb garden. On a farm, a farmer could create as many compost piles as possible with the amount of organic waste they generate. They can also add farm-animal manure and animal parts in some cases, which are rich in nutrients. Side note: it is important to be careful with animals and their products (including dairy) when composting because animal material is particularly vulnerable to dangerous bacteria and pathogens (Michigan State University Extension, 2015). Any amount of compost can enrich the soil health and add plant nutrients. However, the application of a farm's compost onto their crops doesn't just magically provide crops with the perfect amount of nutrients, amend the soil to perfection, and solve all the problems agriculture poses to the planet's health. If only it were that simple. Though any amount of compost can be beneficial to the soil, creating and maintaining a compost can be energy intensive. Farmers already have a lot on their plates without adding another step to the process. Once you start a compost pile, you must maintain it to make sure the microbes have the right ratio of materials, enough oxygen, and enough water. It also takes about a year for the material to be ready to be put into the soil, so there is a buffer period before when you can start using your own compost. Furthermore, one cannot rely solely on compost to fertilize their crops. Stoichiometrically speaking, the organic material being recycled by the farm is not the same as the mass of organic matter being grown in the farm's soil. The bulk quantities of nutrient-rich fruits, vegetables, nuts, and grains produced by farms are not replaced when only the scraps are composted. Also, a single family living on and running a farm most likely wouldn't be producing enough food scraps to provide enough compost for their crops. So, if a farm tried to return nutrients back to the soil with their compost alone, their soil would get increasingly more depleted of nutrients.

Are we to throw away the entire idea of composting because a single farm can't produce enough organic matter to maintain a sufficient amount of compost, or could we somehow utilize the vast quantities of food scraps produced by every farm's nearby community which are thrown away just to break down and fertilize landfills? Food is the largest factor of waste produced in

the US, making up 22% of municipal solid waste (RTS, 2025). The US throws away more food than any other country in the world, around 60 million tons every year, which is almost 40% of our entire food supply. Per person, that is 325 pounds of waste every year. The sources of this food waste range from food spoilage- either real or because of expiration dates (which a lot of the time are way more conservative than they need to be or aren't even actual expiration dates at all and are actually sell-by dates), take-out culture, overproduction, restaurant waste, and more (ReFED, n.d.). If every community collected their food waste and composted it, there could be a dependable source of compost for local farms!

Here at Montana State University in Bozeman, Montana, a team in the Office of Sustainability is executing a student-led initiative to compost food-waste generated by the school. Food waste from the two dining halls and the cafes on campus as well as the organic matter waste generated by landscaping such as grass cuttings and leaves are saved and picked up by a local composting company called Happy Trashcan. Happy Trashcan gets paid to collect compost from individual homes, restaurants, establishments, and now the university in Bozeman. At their operation, they process the compost so that it is soil-ready and then sell it back to those with gardens, farms— whoever has a use for it and can pay for the service. A system like this where a community's organic waste is recycled and put back to nurture the community is called 'community composting'. Community composting reduces greenhouse gasses from waste-transport and reduces the methane emissions organic material can release when dumped in a landfill (U.S. Environmental Protection Agency, 2025). Another effect of engaging the community to support soil health and sustainable nutrient cycling is that it exposes individuals to the systems that nurture them. It connects people to the production of their food, grown in the scraps of their previous meal. Supporting your community by sourcing locally and giving back with reciprocity with things like your compost strengthens your community, addresses local food insecurity, creates local jobs, and increases individual environmental awareness. By stewarding systems like community composting, we can provide farms access to enough compost to nourish their soils and maximize the benefits of compost applications.

Two main caveats of composting, especially at a large scale, are that it 1) produces greenhouse gases and 2) poses a threat of contamination. In aerobic decomposition of organic matter, carbon dioxide is produced as a byproduct. In anaerobic decomposition of organic matter, methane is produced as a byproduct (Live to Plant, 2025). Carbon dioxide and methane are major greenhouse gases which both contribute greatly to global warming. Some argue that composting more intensely on farms will increase greenhouse gas emissions, effectively cancelling out the positive effects composting may have on the planet. However, I posit that organic matter will be discarded and decompose whether we use it as an application on soil or not. When we just throw it away and it ends up in a landfill. The main form of decomposition there is anaerobic, so more organic matter may produce more methane when thrown away than when it is used in agriculture. Methane is a much more potent greenhouse gas than carbon dioxide is (International Energy Agency, 2021), and it would be advantageous to limit production in landfills. Furthermore, some argue that using compost is dangerous because it could be a source of toxic contamination— something we definitely do not want around our food. Compost is a conglomeration of discarded organic matter. Sources like lawns, farms, restaurants, and stores all contribute chemicals that end up in or on the organic matter. Pesticides and herbicides from food production can persist in compost, and some can outlive the decomposition process for years (US Composting Council, n.d.). Trash, PFAS (Per- and Polyfluoroalkyl substances),

and microplastics can make their way through the sorting process, regardless of a composter's best efforts to create clean compost (US Composting Council, n.d.). This issue stems from the place we are in as a society where we are learning that many of the chemicals we use in our products are persistent and toxic to both us and the environment. I would argue that contamination is an issue we should be tackling parallel to the issues of soil health and erosion. We should not cease to use compost as a soil amendment because it may contain contaminants. We should figure out how to eliminate contaminants from organic matter so that we are not poisoning ourselves and our planet in the first place.

How much can we really reduce the need for synthetic fertilizers with compost? For a backyard garden, one may be able to grow their produce using nothing but compost. However, growing produce in a large agricultural operation requires that the grower pays close attention to the ratios of nutrients being supplied by their compost. One may need to test the soil to ensure adequate quantities and ratios and adjust the ingredients of their organic matter accordingly (GardenerBible, 2025). It is also helpful in this setting to grow a diverse array of produce to ensure that one crop is not competing for specific nutrients provided by a standard compost. On a farm, this is harder to do as a producer usually aims to produce a large quantity of one type of crop. In a study conducted on an 'Anna' apple orchard in Giza, Egypt, researchers determined that apples grown in 75% NPK fertilizer with compost and microbial inoculant increased shoot length, shoot diameter, leaf area, as well as leaf-specific weight of the apple trees significantly compared to that of 100% NPK fertilizer alone (Okba et al., 2025). A study conducted at the Experimental Garden of the Faculty of Agriculture at the University of Muhammadiyah Jakarta found that soybeans grown in 50% inorganic fertilizers and household-waste compost were not affected compared to soybeans grown in 100% inorganic fertilizers (Elfarisna et al., 2023). In another study conducted near Cairo, Egypt, researchers found that using 50% inorganic fertilizers with compost instead of 100% inorganic fertilizers had superior production in "Superior Seedless" grapes as well as the lowest contamination levels of nitrates and nitrites in the grapes (Abdel-Mohsen et al., 2024). Nitrates and nitrites can be produced in excess by inorganic fertilizers and are toxic at high enough concentrations (10 ppm for nitrates and 1 ppm for nitrites) (Maine Center for Disease Control and Prevention, n.d.). However, another study found that cowpeas grown in compost produced fewer pods per plant, seeds per pod, and seeds per plant, than when plants were grown in inorganic fertilizers. However, the plants with the greatest number of seeds per plant were those grown in compost combined with inorganic fertilizers (Diatta et al., 2024). The consensus of these and similar trials conducted across crop types is that to produce the same, or better, yield than inorganic fertilizers, compost must be applied in combination with fertilizers.

Even though it doesn't seem possible to put an end to fertilizers entirely, compost can decrease the amount being applied to soils and leaching into water by at least 50% according to these studies. That's huge! That's 50% less fertilizer acidifying soil and leaching into water systems and creating eutrophication. Do we have to use inorganic and mined fertilizers for the other 50%? No, there are alternatives such as organic fertilizers!

What's the difference between organic and inorganic fertilizers? Aside from raw, fresh compost, another way to deliver nutrients to plants is through dry, concentrated, granular organic matter that comes from the same sources as straight up compost (Fertilizer Production Line, 2025). Organic fertilizer is commonly derived from animal manure, compost, bone and blood

meal, emulsified fish, seaweed, and green manure crops which are high in nutrients (Fertilizer Production Line, 2025). One increasingly used source of these fertilizers is treated sewage sludge due to the large availability of organic material in this resource (U.S. Environmental Protection Agency, n.d.). However, some cases of land-application of these fertilizers have led to contamination of PFAS (Via & Singh, 2024). Organic fertilizers are like compost in that they add organic matter back to the soil and therefore nourish the soil and microbial health, though in smaller, more concentrated quantities than actual compost. Inorganic fertilizers, on the contrary, are produced artificially or mined from the Earth and then refined to be highly concentrated (Cherlinka, 2023; Spooner Agricultural Research Station, 2014). Common inorganic fertilizers are ammonium nitrate, urea, triple superphosphate, potassium chloride, and ammonium sulfate (Fertilizer Production Line, 2025). Inorganic fertilizers are synthetic, mineral based nutrients that are concentrated and water-soluble, making them highly effective at promoting rapid plant growth and high yields (University of Minnesota Extension, n.d.). However, regular application of inorganic fertilizers can acidify and erode soil, erode away organic matter, destroy water retention capacity of the soil, and greatly harm the microbial communities of soils (Xing et al., 2025). Inorganic fertilizers also leach through the soil into water resources, causing eutrophication and nitrate toxicity in their path (Campos & Pereira, 2021). The consequences of using inorganic fertilizers are dire, and it is important for the health of humans and the environment that we find solutions. Even merely minimizing the use of inorganic fertilizers with the help of compost and organic fertilizers could make a huge impact.

Despite their advantages, inorganic and mined fertilizers are much more affordable than organic fertilizers on a per-nutrient basis (Nebraska Extension – Lancaster County, 2024). That means that farmers would have to pay more for the same quantity of organic fertilizers as they buy in inorganic and/or mined fertilizers as well as purchase more to make up for the difference in nutrient concentrations. It's not surprising that most farmers would probably go with the more efficient and cheaper option. This is where I propose a paradigm shift in order for us as a species to figure out how to produce food without compromising our environment. If there was a greater demand for organic fertilizers over inorganic and mined fertilizers, the system would evolve to make organic fertilizers easier to produce, cheaper, and more available. Basic supply and demand dynamics here. Something that would greatly enhance a societal shift like this would be government incentives. If policies recognized the benefits of using compost and organic fertilizers instead of inorganic and mined fertilizers, both for our soils and our produce, the government could offer farmers financial incentives to adopt these practices. If more and more farmers adopted these practices, the resources would become more efficiently produced, cheaper, and more accessible. Future studies that evaluate the yield differences between crops grown with 50% compost and 50% organic fertilizers in comparison to crops grown with 50% compost and 50% inorganic fertilizers as well as contextual trade-offs would be helpful in illuminating this idea.

Normalizing compost use on farms big and small through community composting, financial incentives, and paradigm shifts can steer modern agriculture in a more sustainable, soil-conscious direction. Reducing applications of inorganic fertilizers with compost and organic fertilizers can increase carbon storage, increase water retention, nurture essential soil microbiomes, provide slow-release nutrients for crops, and decrease methane production from landfills.

Final Conclusion

Human beings, in our rush to industrialize, modernize, and expand, have impacted the world and its many other species in innumerable ways. We, as environmental science students, were tasked with identifying and explaining examples of our impacts. Rapid expansion and urbanization are leading to constant habitat loss and fragmentation. Surface waters are declining, impacting the West Slope Cutthroat trout populations. Grasslands are becoming more developed, devastating the native grassland bird species. Forest fires are more frequent and the chemicals we use to fight them have many negative impacts, such as on soil microbial communities. Dams create environments that convert contaminants into their most toxic forms, posing threats to potential dam removal projects. Modern, industrial agriculture is in dire need of innovation to prevent further soil erosion, pollution, and other issues. We as a species need to find practical solutions to these and many other problems that will allow for us to live symbiotically with the planet and its many other species. In our careers as environmental scientists, whatever forms that may take, we want to be a part of these solutions. We can do this by striving for and supporting environmentally-minded change, uplifting our communities to do the same, and supporting policies and policy-makers that take responsibility for our impact on this planet and want to do better because of it.

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