

# Soil Nitrous Oxide Emissions from a Continuous Wheat Cropping System in Montana

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## Introduction

Nitrous oxide ( $N_2O$ ) is a trace gas in the atmosphere that has come under increasing scrutiny because it contributes to global warming and destruction of the stratospheric ozone layer. Human alterations of the global N cycle, including the use of N fertilizer, are known to promote the release of  $N_2O$  from soils into the atmosphere. Nitrous oxide production in soils occurs as a result of two microbial processes: 1) nitrification of ammonium and ammonium-producing fertilizers (e.g. urea) under aerobic conditions and 2) denitrification of nitrate under anaerobic conditions (Fig.1). During the past 250 years there has been a 17% increase in the atmospheric  $N_2O$  concentration to its present level of 316 parts per billion. Agriculture is reported to account for 65-70% of global anthropogenic emissions, and fertilizer N use (commercial and manure) is considered the primary contributor by the International Panel on Climate Change (IPCC, 2001). Regional and global estimates of  $N_2O$  production from agriculture have frequently been adopted using IPCC methodology. Beginning in 1997, IPCC methodology assumed as a default that 1.25% of all N inputs, including fertilizer N, are lost directly as  $N_2O$ . This default value was developed from databases currently available at the time, most of which came from regions that were considerably more humid than Montana. Given that soil  $N_2O$  emissions are known to be affected by differences in cropping systems and climate, there is uncertainty as to the accuracy of the 1.25% default value to Montana agriculture. This study was undertaken to describe seasonal patterns of  $N_2O$  release from a continuous wheat cropping system and provide an estimate of the effect of N fertilization on  $N_2O$  emission losses.

## Methods

Nitrous oxide gas samples were collected over two years (Apr. 14, 2004 to Apr. 15, 2006) at the Montana State University – Arthur Post Farm in Bozeman. The soil at the site is classified as an Amsterdam silt loam (fine-silty, mixed, superactive, frigid Typic Haplustolls) with pH 7.2, and organic matter content of 1.5% in the surface 8 in. The field study was part of a larger cropping system study, but only the results from the no-till

winter wheat – spring wheat rotation are presented here. The wheat-wheat system was divided into subplots representing three target levels of available N, including a low-unfertilized regime, a moderate available N regime (90 lb N/ac), and a high available N regime (180 lb N/ac). The treatments were replicated four times. Available N pool was estimated from the sum of soil  $NO_3-N$  (0–24 in.) plus fertilizer N applied, with the fertilizer N application rates in the moderate and high regimes calculated by the difference between soil  $NO_3-N$  tests and the target N level. Fertilizer N applications (as urea) were equivalent to 156 and 218 lb N/ac over two years for the moderate and high N regimes, respectively. Gas sampling was conducted using static chamber techniques. Gas samples were collected from the headspace during the early to mid-afternoon (1 – 3 p.m.). The concentration of  $N_2O$  in the container was determined using a gas chromatograph.

## Results

Nitrous oxide flux vs. time profiles (Fig. 2) from the continuous wheat rotation revealed that emissions were episodic and responsive to periods of high soil water-filled pore space and availability of N substrate (soil or fertilizer). Examination of the curves reveals that N fertilization was perhaps the single most important event that stimulated an increase in  $N_2O$  emissions. The elevation in emissions occurred within a week following fertilization, and peaked after approximately 2-4 weeks. The duration of elevated flux above background ( $>2.0 \mu g N_2O-N m^2/h$ ) for spring applications in 2004 and 2005 was approximately 10 weeks, but extended somewhat longer for the fall application in 2005 (Sept. 30). The majority of  $N_2O$  losses during the 10-wk period following fertilization were probably a result of nitrification, except in the fall 2005, when denitrification may have been important as soil water contents were high and frequently exceeded  $>70\%$  water-filled pore space. In addition to N fertilization, freeze-thaw cycles in the winter or early spring were also important in stimulating significant  $N_2O$  emission activity.

# Fertilizer

## F a c t s

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Feb 2007  
Number 44



