


Soil Acidification

MSU IPM Workshop
Bozeman, June 19, 2018

pH 5.1

pH 3.8

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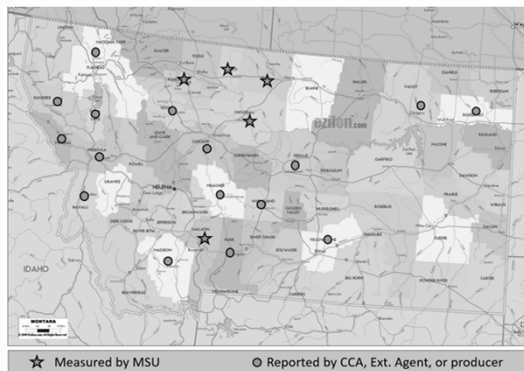
 MONTANA STATE UNIVERSITY MSU Soil Fertility Extension

Objectives

Specifically, I will:

1. Show prevalence of acidification in Montana
2. Review acidification's cause and contributing factors
3. Depict low-pH soil affected crops
4. Discuss steps to prevent or minimize acidification

Prevalence: MT counties with at least one field with pH < 5.5



40% of 20 random locations in Chouteau County have pH < 5.5.

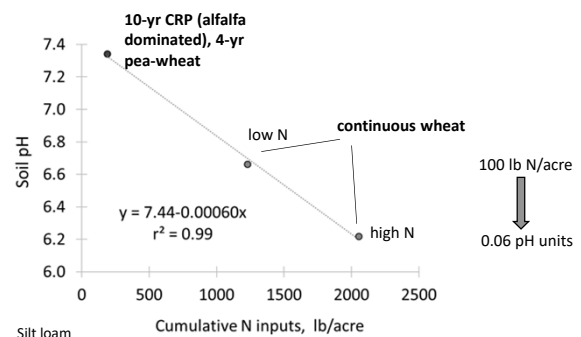
Natural reasons for low soil pH

- Soils with low buffering capacity (low soil organic matter, coarse texture, granitic rather than calcareous)
- Historical forest vegetation soils have lower pH than historical grassland
- Regions with high precipitation, leading to leaching of nitrate

Agronomic reasons for low soil pH

- Ammonium-based N fertilizer above plant needs due to Nitrification: ammonium or urea fertilizer + air + H₂O → nitrate (NO₃⁻) + acid (H⁺)
- Leaching loss of nitrate – less nitrate uptake and less root release of basic anions (OH⁻ and HCO₃⁻)
- Crop residue removal – removes Ca, Mg, K ('base' cations)
- No-till concentrates acidity where N fertilizer applied
- Legumes acidify their rooting zone through N-fixation. Perennial legumes (e.g., alfalfa) more so than annuals (e.g., pea). Yet apparently much less than fertilization of wheat.

Effect of 14-yr of N fertilization on dryland cropping west of Bozeman, November 2016



Prevention: Scout and soil test

1. Scout your fields to look for 'unexplained' poor plant health
2. Look at pH on prior soil tests from composited samples
 - pH < 6 likely have spots with pH near or below 5 (go to step 3)
 - 6 < pH < 7.5 don't assume you have no areas with low pH
 - pH > 7.5, likely don't have problem (yet).
3. Field test with pH color strips kits or pH 'stick' on a soil/water slurry of top 3". Test both healthy and unhealthy areas for comparison.
4. Send 0-3" depth sample to lab for pH (<5?) and KCl-extractable Al (> 5 ppm?). Test 3-6" if might till.
5. pH varies seasonally and annually, test from same area and time of year to compare for a trend

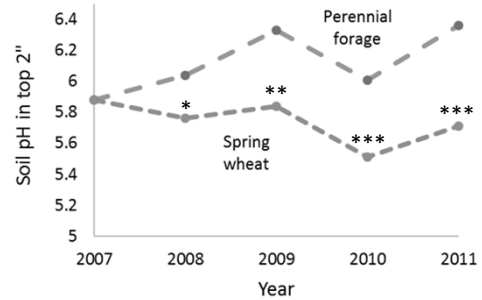
**Management to prevent acidification:
Increase nitrogen fertilizer use efficiency**

- Soil test close to application time. Make sure enough PKS
- Use conservative pre-plant rate, top-dress if adequate moisture
- Apply N close to peak crop uptake
- Plant lower N-needing crops, including pulses
- Plant perennial forages (list of acid-tolerant varieties on our soil acidification website)
- Reduce N rates especially when protein discounts low
- Use variable, site specific rates: Less N in low production areas limited by other than N (e.g., low pH, shallow soils)

**Management to prevent acidification:
Change N source?**

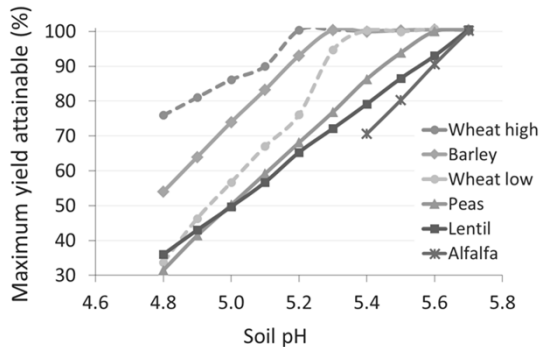
- Minimize use of ammonium fertilizers (11-52-0, 21-0-0-24)
- Use calcium ammonium nitrate (\$\$) instead of urea or UAN (CAN shouldn't volatilize so can likely also lower rate)
- Include legume rotations, manure if available

Perennial forage can maintain or increase soil pH



crops differ with * > 90%, ** > 95%, *** > 99% confidence
Mandan, ND Liebig et al., 2018

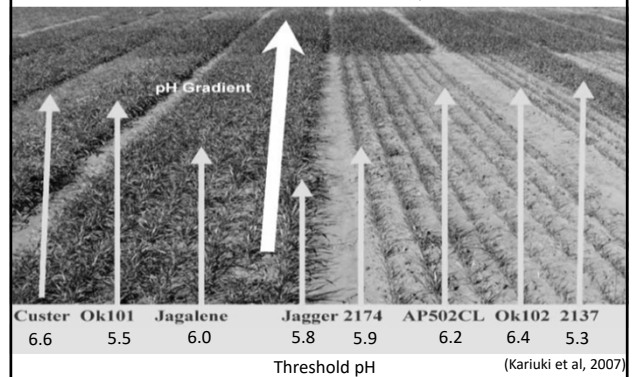
**Adaptation: Crop species vary in tolerance to low soil pH
legumes are least tolerant**



McFarland et al., 2015

"Wheat high" are Al and acid tolerant varieties

Wheat varieties have different tolerance to pH and Al



MT variety trial results are available at

<http://landresources.montana.edu/soilfertility/acidif/index.html>

Liming rate

- Know:
 - Calcium carbonate equivalent (CCE; how source compares to pure CaCO_3)
 - Lime score (LS; adds factors for moisture and fineness to CCE)
 - Current soil pH and desired pH
 - > 5 to reduce Al toxicity
 - > 5.5 to have some buffer
 - > 6 to be good for 10+ years
 - Desired crop
- Rate: from soil test lab or calculate (WSU equation)
 $\text{Lime rate (ton/acre)} = 1.86 * (\text{final desired pH} - 4.6)$
 Note: R. Engel is conducting research to develop MT specific recommendations



Sugarbeet lime

- good – it doesn't cost anything
- bad – shipping costs (up to \$35/ton); challenging material to work with (moisture and clumping), need a wet lime spreader, contains chunks and some trash, and incorporation w/ tillage needed for best results
- rates of 3-6 tons/acre may be necessary to bring pH to acceptable level (pH > 6)



Stoltzfus lime spreader, Stoltz Mfg.

Summary

- Cropland soils are becoming more acidic, largely due to N fertilization
- This reduces yields for several reasons
- Sound nutrient, crop, and residue management can slow or prevent soil acidification
- Crop and variety selection can help adapt to acid soils
- Liming, perhaps tilling, or planting perennials can mitigate acidification

Questions?



Image from Oregon State University, Lane County, OR 1926.

For more information see the 2 *Soil Scoops* on soil acidification and more presentations at MSU Soil Fertility website:
<http://landresources.montana.edu/soilfertility/soilscoop.html>

MSU's cropland soil acidification website
<http://landresources.montana.edu/soilfertility/acidif/index.html>