SOIL ACIDIFICATION CAUSES and CONCERNS

Montana Association of Conservation Districts Convention

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Questions for you

- How many think you have seen yield losses from acidic soils?
- How many of you have soil pH levels below 5.5?
- How many have observed decreasing pH levels on your soil test reports?
- Suggestions for determining how widespread this problem is?



- Show how number of acidic soils have been increasing in Montana
- Present natural and agronomic conditions that lead to acidic soils and some negative impacts
- Depict low-pH soil affected crops
- Present management options to slow acidification and increase productivity of acidic soil

Acidic soil samples (pH<6.5) are increasing in MT (though still rare)



Unpub data Agvise



Sampled by MSU, low pH and/or high Al

Reported by CCA or Ext Agent

Durum wheat - aluminum toxicity due to low pH



Low soil pH is generally only in upper 6 inches

Soil surface



Natural reasons for low soil pH

- Soils with low buffering capacity (low soil organic matter, coarse texture), granitic > calcareous
- Historical forest vegetation > historical grassland, which developed greater buffering capacity

Agronomic reasons for low soil pH

- Ammonium-based N fertilizer above plant needs due to Nitrification: ammonium or urea fertilizer + air + H₂O→ nitrate (NO₃⁻) + acid (H⁺)
- Leaching loss of nitrate prevents plants from buffering soil with basic anions (OH⁻ and HCO₃⁻)
- Crop residue removal removes Ca, Mg, K ('base' cations)
- No-till concentrates acidity where N fertilizer applied
- Legumes acidify their rooting zone through N-fixation.
 Perennial legumes (e.g., alfalfa) more so than annuals (e.g., pea).

Soil pH and nutrient availability

At low soil pH:

- Plants go hungry for some nutrients
- Nutrients can be lost to environment
- Al and Mn reach toxic levels



Low pH increases soil Al to toxic levels



Engel unpub. data, 2016, 5 farms near Highwood, MT

9 years of differing N rates on soil Mn levels in SK



Bouman et al., 1995, Scott, SK. Mn toxic level from Moroni et al., 1991

Acid soils have many additional negative impacts

- Herbicide persistence (Raeder et al., 2015)
- Damage to rhizobia (N-fixing by legumes)
- Increase in fungal diseases
- Toxic H⁺ levels (Kidd and Proctor, 2001)





What to look for. Unexplained poor health in low or mid-slope areas

Al toxicity

- stubby club roots (similar to nematode damage)
- small leaves, short thick internodes
- yellow along margin near tip on older leaves
- purple or brown lesions in chlorotic regions, indentations
- leaf withering and collapse in center



Image KS State Research and Extension



Field pea. A. Robson, https://agric.wa.gov.au/n/4487



Information and photo source: CIMMYT.org

What to look for. Unexplained poor health in low or mid-slope areas

Mn toxicity

- cupping and crinkling
- chlorotic and reddening spots to necrosis
- starting on older leaves at tips and margins





Information and photo source: CIMMYT.org

What to do if suspect low pH soil?

1. Soil test top 6"

- pH < 6 likely have spots with decreased yields (go to 2 below)
- 6 < pH < 7 don't assume you have no areas with low pH
- pH > 7, likely don't have problem (skip 2-4).
- 2. On top 3" where have low pH symptoms: Use pH color strips on a soil/water slurry or send sample to lab for pH (<5?) and KCl-extractable Al (> 5 ppm). Test 3-6" if might till.
- **3**. OR: Compare Al and Mn tissue analysis from 'good' and 'bad' crop areas.

Managing low pH

Remediate

- Use lime amendments (sugarbeet or Ag Lime)
- Consider
 occasional tillage
 (once in 5 years?)
- Plant acidtolerant perennial crops

Adapt

- Plant Al-tolerant crops or varieties (e.g. Judee, Alum. See CARC webpage for variety trial data)
- Seed deeper?
- Fertilize after vulnerable seedling stage

Prevent

- Optimize N use efficiency – no left-over N
- Consider different
 N sources
- Retain crop residue

Steps to minimize and prevent soil acidification due to fertilizer N

Increase efficiency of N use

Base nitrogen rate on spring soil test and realistic yield potential

Split nitrogen applications

Reduce nitrate loss

Use slow-release nitrogen sources

Use nitrogen sources with nitrification inhibitors

Recrop or plant deep rooted crops to 'catch' deep nitrate

Consider using nitrogen sources other than urea, UAN, anhydrous, ammonium phosphate, or ammonium sulfate

Legume rotations, manure

Calcium ammonium nitrate (27-0-0)



- Cropland soils are becoming more acidic, largely due to N fertilization
- This reduces yields for several reasons
- Sound N fertilizer, crop, and residue management can slow or prevent soil acidification
- Crop and variety selection can help adapt to acid soils
- Liming, perhaps tilling, or planting perennials can mitigate acidification

Questions?

Limed

Image from Oregon State University, Lane County, OR 1926.

For more information see the 2 *Soil Scoops* on soil acidification at MSU Soil Fertility website:

Not limed

http://landresources.montana.edu/soilfertility/soilscoop.html

Eastern Oregon Liming Guide OSU EM9060, Soil Acidity in Oregon: Understanding and Using Concepts for Crop Production OSU EM9061 <u>https://catalog.extension.oregonstate.edu/</u>